FOREWORD INTRODUCTION

[HYDROXYPROPYL ACRYLATE](#page-1-0)

CAS N°: 25584-83-2

SIDS Initial Assessment Report

For

SIAM 20

Paris, France, 19-22 April 2005

- **1. Chemical Name:** Hydroxypropyl acrylate
- **2. CAS Number:** 25584-83-2

3. Sponsor Country: United States

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which were then reviewed by U.S. EPA.

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companies of the consortium.

703-669-5688

of the consortium.

See 5 above

- **4. Shared Partnership with:** HEA/HPA Consortium
- **5. Roles/Responsibilities of the Partners:**
- Name of industry sponsor /consortium
- Process used Data searches included published scientific literature, databases

6. Sponsorship History

- How was the chemical or category brought into the OECD HPV Chemicals Programme?
- **7. Review Process Prior to the SIAM:**
-

8. Quality check process: U.S. EPA reviewed the information in the industry sponsor's submission.

The HEA/HPA Consortium prepared the initial documents,

and handbooks as well as the internal files of the member

The IUCLID Data Set has been revised and the SIAR prepared by a consortium of chemical industry producers in 2004. Data searches included published scientific literature, databases and handbooks as well as the internal files of the member companies

- **9. Date of Submission:**
- **10. Date of last Update:** December 2005
- **11. Comments:**

SIDS INITIAL ASSESSMENT PROFILE

SUMMARY CONCLUSIONS OF THE SIAR

Human Health

The kinetics of hydroxypropyl acrylate have not been studied. However, results from oral, inhalation, and dermal single dose studies with the closely related, hydroxyethyl acrylate (CAS No. 818-61-1), indicated rapid metabolism via hydrolysis of the ester functionality, similar to many other acrylic acid esters. For hydroxyethyl acrylate, rapid metabolism to CO₂ and urinary metabolites was not route-dependent. The half-lives of elimination of radioactivity were approximately 14 hours for urine and 17 hours for CO2. The half-life of elimination of radioactivity from plasma was approximately 26 hours. Based on the similarity of the results for hydroxyethyl acrylate with other acrylic acid esters, similar kinetics of hydroxypropyl acrylate is anticipated.

Acute toxicity studies with experimental animals indicate oral and dermal LD_{50} values of 820 mg/kg bw (rat) and 306 mg/kg bw (rabbit), respectively. In an acute inhalation study, exposure to a saturated vapor for eight hours did not result in deaths. In standard primary irritation studies, hydroxypropyl acrylate is a severe skin irritant. Upon eye contact, hydroxypropyl acrylate may cause severe irritation with corneal injury which may result in permanent impairment of vision or blindness. Skin sensitization studies in animals and humans indicate that hydroxypropyl acrylate is likely to be a sensitizer and will cross-react with other acrylates in some exposed individuals.

Repeated exposure to vapors of hydroxypropyl acrylate (6 hr/day, 5 days/week for 21 or 20 exposures to rats and mice and rabbits and dogs, respectively) results in severe irritation of the upper respiratory tract, resulting in death due to respiratory failure at higher concentrations and concentration-related local irritation at sublethal exposures. The LOAEC for subchronic exposure, based on irritation, was 5 ppm (27 mg/m^3) for hydroxypropyl acrylate. Except for irritant effects, no systemic toxicity was observed.

Hydroxypropyl acrylate was not mutagenic to *Salmonella typhimurium* (bacterial reverse mutation assay) *in vitro* with or without metabolic activation but was positive with metabolic activation when tested with two *E. coli* strains. In mammalian cells *in vitro*, hydroxypropyl acrylate was negative in a gene mutation assay but had clastogenic activity in cytogenetic and chromosomal aberration assays. In these mammalian cell assays, positive results occurred only at concentrations that resulted in significant cell death. Thus, the positive results are considered equivocal. Hydroxypropyl acrylate was not mutagenic in an *in vivo* mouse micronucleus study. Overall, hydroxypropyl acrylate is considered not to have mutagenic potential *in vivo* based on available data.

Hydroxypropyl acrylate had no effect on male reproductive organs in four species following repeated exposures via inhalation for 30 or 31 days (20 or 21 exposures). The principal treatment-related effects observed following 18 months exposure of laboratory rats to 5 ppm of the closely related, hydroxyethyl acrylate, were related to irritation of the respiratory tract, without significant evidence of systemic toxicity. Histopathological examination of the reproductive organs revealed effects in the arteries of the testes (fibrinoid degeneration in the

vascular channels — an age-related lesion) and uterus (inflammation) which were interpreted as neither treatment-related nor adverse to reproduction.

In a well-conducted inhalation study exposing pregnant rats to hydroxypropyl acrylate from gestation day 6 to 20, maternal body weights were reduced at the two highest exposure concentrations (5 and 10 ppm; 27 and 53 mg/m³), but no embryo-fetal or developmental toxicity was observed. Overall, based on the available studies, hydroxypropyl acrylate does not show evidence for developmental toxicity.

Environment

The melting point is -23.4°C and the boiling point is 205.7°C. The vapor pressure is 0.038 hPa at 25°C. The measured log Kow is 0.35. The water solubility of hydroxypropyl acrylate is estimated to be 307 g/L (25°C) and specific gravity is 1.049 g/cm³ at 25° C.

Hydroxypropyl acrylate is photodegraded by reaction with hydroxyl radicals in the atmosphere with a half-life of 7.4 hours (calculated). The hydrolysis rate of hydroxypropyl acrylate is pH dependent with hydrolysis halflives of > 490 days and >230 days at pH 3 and pH 7, respectively. The hydrolysis half-life at pH 11 is 0.056 days.

Distribution modeling using Mackay Level I indicates hydroxypropyl acrylate released into the environment partitions almost completely (99.8%) to the water phase. Fugacity model Level III with 100% of the hydroxypropyl acrylate release to air distribution is: $\langle 1\%$ (air), 27% (water), 73% (soil) and $\langle 0.1\%$ (sediment). Fugacity model Level III distribution with 100% of the hydroxypropyl acrylate release to water is: <0.1% (air), 100% (water), <0.1% (soil) and <1% (sediment). Fugacity model Level III distribution with 100% of the hydroxypropyl acrylate release to soil is: <0.1% (air), 21% (water), 79% (soil) and <0.1% (sediment). Fugacity model Level III distribution with equal distribution of hydroxypropyl acrylate release to air, water and soil is: $\leq 0.1\%$ (air), 45% (water), 55% (soil) and $\leq 0.1\%$ (sediment)

A low bioaccumulation potential is expected based on the partition coefficient of Log Kow of 0.35. Based on an OECD Guideline 301C study, hydroxypropyl acrylate is readily biodegradable (83% degraded over 28 days).

The 96-hour LC₅₀ for fathead minnow was 3.1 mg/L (measured), the 48-hour EC₅₀ for *Daphnia magna* was 24 mg/L (nominal) and the 96-hour EC₅₀ values for biomass and growth rate of algae (*Selenastrum capricornutum*) were 3.53 and 6.67 mg/L (nominal), respectively.

Exposure

The worldwide production volume of hydroxypropyl acrylate is estimated to be 6000 to 7000 tonnes per year. The US production is estimated to be 5000 tonnes per year. Hydroxypropyl acrylate is produced and primarily used in closed systems. Its principle use is either as a co-monomer in the manufacture of polymers or as a chemical reactant in the manufacture of chemical intermediates. The polymers and chemical intermediates made with hydroxypropyl acrylate find applications in automotive top coatings, architectural coatings, photocure resins, and adhesives.

Based on the use pattern of the substance, significant environmental releases are unlikely. Impact on the environment is expected to be low due to photolysis, biodegradation and lack of bioaccumulation. Results from workplace measurements at a US production site indicated that hydroxypropyl acrylate did not exceed the TLV of 0.5 ppm in 87 samples collected over 20 years. Worker exposure is limited by the use of enclosed processing systems, industrial hygiene controls and personal protective measures. Hydroxypropyl acrylate has a characteristic acrylic odor, which can provide a measure of warning of the presence of vapors. End-use consumer products contain only trace levels of acrylic acid and esters (as a result of polymerization). Therefore, consumer exposure to acrylate monomers is not anticipated.

RECOMMENDATION AND RATIONALE FOR THE RECOMMENDATION AND NATURE OF FURTHER WORK RECOMMENDED

Human Health: The chemical is currently of low priority for further work. The chemical has properties indicating a hazard for human health (severe eye irritation with corneal injury, which may result in permanent impairment of vision, even blindness, skin and upper respiratory tract irritation, potential skin sensitization, and acute toxicity from inhalation exposure). Based on exposure data presented by the Sponsor Country (relating to

production in one country which accounts for approximately 80% of global production and relating to the use pattern in several countries), this chemical is currently of low priority for further work. Countries may wish to investigate any exposure scenarios that were not presented by the Sponsor country.

Environment: This chemical has properties indicating a hazard for the environment (fish, invertebrate, algae). However, the chemical is of low priority for further work for the environment because of its ready biodegradability and the limited potential for bioaccumulation.

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1 IDENTITY

1.1 Identification of the Substance

1.2 Purity/Impurities/Additives

A typical commercial sample of hydroxypropyl acrylate contains approximately 75-80% 2 hydroxypropyl acrylate and 20-25% 1-methyl-2-hydroxyethyl acrylate. The purity of the salable product is at least 97% combined isomers. A typical commercial sample of hydroxypropyl acrylate may contain acrylic acid ($\leq 0.98\%$ w/w), other esters ($\leq 1.8\%$ w/w), propylene glycol diacrylate (≤0.2% w/w) and propylene oxide (ca. 0.001% w/w). Methyl ethyl hydroquinone may be added at 250 to 650 ppm as an inhibitor of spontaneous polymerization.

1.3 Physico-Chemical properties

Property	Value	Reference
Physical state	Liquid	
Melting point	$-23.4 °C$	Rowley et al. in DIPPR, 2004
Boiling point	205.7 °C at 1013 hPa	Rowley et al. in DIPPR, 2004
Density	1.049 g/cm ³ at 25° C	DOW, 2002
Vapour pressure	0.038 hPa at 25° C	Rowley et al. in DIPPR, 2004
Water solubility	307 g/L (a) 25°C (Considered miscible)	U.S. EPA, 2000a
Partition coefficient n- octanol/water (log value)	0.35 (a) ca. 25 °C	Tanii and Hashimoto, 1982
Henry's law constant	0.00161 Pa x m ³ / mol	HENRYWIN v 3.10

Table 1 Summary of physico-chemical properties

1.4 Analogue Justification

Justification for Use of Limited Hydroxyethyl Acrylate Data to Support Hydroxypropyl Acrylate for Mammalian Toxicity: Hydroxyethyl acrylate (CAS RN 818-61-1) is a member of the acrylate ester family with similar structure, physical/chemical properties and fate and effects profile as hydroxypropyl acrylate. The molecules are very similar structurally with the addition of a single carbon on the ester chain of hydroxyethyl acrylate. The structures are shown below:

As noted in the mammalian toxicity study reviews below, the LOAEC values from inhalation studies for the two chemicals are similar (5 ppm for hydroxypropyl acrylate and hydroxyethyl acrylate) and are based on local irritation. As with other acrylates, at sublethal levels, the major effects of both chemicals are related to irritation at the site of contact (stomach from gavage dosing, nasal and respiratory irritation from inhalation exposure, skin irritation from cutaneous exposure). Therefore, where appropriate for mammalian toxicity endpoints, hydroxyethyl acrylate data are included herein to support the toxicological profile of hydroxypropyl acrylate.

2 GENERAL INFORMATION ON EXPOSURE

The worldwide production volume of hydroxypropyl acrylate is estimated to be 6000 to 7000 tonnes per year. The US production is estimated to be 5000 tonnes per year.

Release of hydroxypropyl acrylate into the environment is unlikely based on the use pattern (closed system) and virtually complete polymerization in the final products. Small accidental releases are considered to be the only potential source of hydroxypropyl acrylate environmental contamination.

Hydroxypropyl acrylate is a chemical monomer, manufactured and processed within closed systems. The primary routes of potential exposure to hydroxypropyl acrylate are skin contact and vapor inhalation. The lower vapor pressure (0.038 hPa at 25ºC) helps reduce potential vapor inhalation exposure. In an industrial setting, ingestion is not an anticipated route of exposure. The primary use of hydroxypropyl acrylate in the production of polymeric coatings (used predominantly in the automotive industry) results in virtually no unreacted monomer in the finished coatings. The potential exposure to aerosols of hydroxypropyl acrylate is, therefore, highly unlikely.

Consumer exposure to hydroxypropyl acrylate is expected to be negligible because hydroxypropyl acrylate is used as a co-monomer in polymers (virtually complete polymerization occurs in the final products) or a reactant to make chemical intermediates and consumer products do not contain unreacted hydroxypropyl acrylate.

2.1 Production Volumes and Use Pattern

The worldwide annual production volume of hydroxypropyl acrylate for the SIDS-sponsoring companies (Cognis Performance Chemicals UK Limited, Degussa/Röhm GMBH & Co. KG, The Dow Chemical Company, Nippon Shokubai Co., Ltd., and Rohm and Haas Company) is in the range between 6000 and 7000 tonnes (2001 data).

Hydroxypropyl acrylate is mainly used either as a co-monomer in the manufacture of polymers or as a chemical reactant in the manufacture of chemical intermediates. In the manufacture of polymers, hydroxypropyl acrylate can be co-polymerized with acrylic acid, acrylates, methacrylates, vinyl acetate, vinyl chloride, vinylidene chloride, styrene, butadiene, and the like. Co-reactants with hydroxypropyl acrylate include aromatic and aliphatic isocyanates, anhydrides, and epoxides. The polymers and chemical intermediates made with hydroxypropyl acrylate find applications in automotive top coatings, architectural coatings, photocure resins, and adhesives. Examination of the Substances in Preparations in Nordic Countries database indicated no additional uses.

The Western Europe consumption of commodity acrylate esters by end use in 1984 was as follows (ECETOC, 1994).

End Use	Percent	
Surface Coatings	35-40%	
Adhesives / Sealants	15%	
Textiles	$10-15%$	
Paper Coating	15-20%	
Fibers and Plastics Comonomers	10%	
I thers	10%	

Table 2 Western Europe consumption of commodity acrylate esters

Table 3 United States consumption of commodity acrylate esters

2.2 Environmental Exposure and Fate

2.2.1 Sources of Environmental Exposure

Releases of hydroxypropyl acrylate into the environment: Release of hydroxypropyl acrylate into the environment is unlikely based on the use pattern (closed system) and virtually complete polymerization in the final products. Small accidental releases are considered to be the only potential source of hydroxypropyl acrylate environmental contamination. For example,. no releases existed to require a Toxics Release Inventory (TRI) submission in the United States in the past decade.. The TRI is a publicly available EPA database that contains information on toxic chemical releases and other waste management activities reported annually. This inventory was established under the Emergency Planning and Community Right-to-Know Act of 1986 (EPCRA) and expanded by the Pollution Prevention Act of 1990.

2.2.2 Photodegradation

The indirect photodegradation of hydroxypropyl acrylate by reaction with hydroxyl radicals in the atmosphere is estimated to occur with a half-life of approximately 7.4 hours (U.S. EPA, 2000b).

2.2.3 Stability in Water

The hydrolysis rate of hydroxypropyl acrylate, as with other acrylates, is pH dependent with little hydrolysis at acid and neutral pH and rapid hydrolysis in alkaline conditions. The hydrolysis halflives determined at pH 3, 7, and 11 were > 490 days, >230 days, and 0.056 days, respectively (Gonsior *et al.*, 1997).

2.2.4 Transport between Environmental Compartments

The theoretical distribution of hydroxypropyl acrylate has been estimated using the fugacity model of Mackay, Level I (Canadian Environmental Modeling Centre, 1999). According to this model, hydroxypropyl acrylate, when released into the environment, partitions almost completely (99.8%) to the water phase.

Table 4 Estimated Distribution Between Environmental Compartments (Level I)

The environmental distribution of hydroxypropyl acrylate has also been examined using the Level III fugacity model (U.S. EPA, 2000c). Results are shown in the table below.

	Percentage distributed to				
Emission Scenario	Air	Water	Soil	Sediment	
$1,000$ kg/hr to Air	1.38 %	26.6%	71.9%	4.5×10^{-2} %	
$1,000 \text{ kg/hr}$ to Water	7.4×10^{-5} %	99.8%	3.9×10^{-3} %	0.17%	
$1,000$ kg/hr to Soil	0.013%	21.0%	78.9%	3.6 x 10^{-2} %	
$1,000 \text{ kg/hr}$ simultaneously to Air, Water, and Soil	0.4%	46.3%	53.2 %	7.6×10^{-2} %	

Table 5 Estimated Distribution Among Air, Water, Soil, and Sediments Under Various Emission Scenarios (Level III)

Conclusion

Hydroxypropyl acrylate has very high water solubility, very low vapor pressure, and very low log Kow. These properties dictate that the material has low potential to volatilize from water to air or adsorb to soil and sediments. When released to water (the most likely emission scenario), the material will remain dissolved in water and will be removed through biodegradation and hydrolysis. Since hydroxypropyl acrylate is susceptible to destructive reactions such as indirect photolysis and biodegradation.

2.2.5 Biodegradation

In a biodegradation assay according to OECD Guideline 301 C, hydroxypropyl acrylate was 83% biodegraded after 28 days (MITI, 2001). In a biodegradation assay according to OECD Guideline 301 D, hydroxypropyl acrylate showed 34.9% biodegradation after 28 days expressed as ThOD (Wu, 1996). A third study reported no degradation after 5 days but toxicity of hydroxypropyl acrylate to the organisms, Polyseed, could not be ruled out (Schaefer, 1995). A fourth study, employing methods similar to current OECD Guideline 302 testing, that used industrial effluent (i.e. acclimated sludge) showed HPA was approximately 73% biodegraded after 20 days (Dow Chemical Company, 1975). Based on the OECD Guideline 301C study, hydroxypropyl acrylate is readily biodegradable.

The difference in results for the OECD 301C and 301D studies is considered to be affected by two factors: First, the inoculum concentration used in the 301D study is lower than for the 301C study. A number of studies have investigated the effect of inoculum concentration on results of the ready biodegradation tests. Three basic observations have been made: first, precision among repeated tests or different labs is improved with increasing inoculum concentration; second, the lag phase, or

time for onset of biodegradation, is typically shortened with increased inoculum concentration (one study suggests that one order of magnitude difference in competent cell concentration can result in about one week difference in the lag phase); and third, the composition/diversity of wastewater microbial populations will differ with time and location, and is mostly determined by organic composition of the wastewater. These observations are linked to the number of "competent" organisms (i.e. capable of degrading the substance) initially present in the test mixtures, and whether these competent organisms will survive and grow under the conditions of the test. A higher total inoculum concentration is expected to result in higher number of competent organisms initially present, and thus, a shortened lag phase during the test. In the case of 301C vs. 301D for HPA, a longer lag phase is expected in the 301D test. However, degradation equal to that in the 301C test would eventually be attained with extended incubation of the 301D test beyond the standard 28 days.

In addition to the above, there is inherent variability in biodegradation assays performed with these types of chemicals dependant on laboratory variability, solubility of the test substance and a variety of other variables. Overall, the high degradation in the 301C study clearly indicates that HPA is readily biodegradable under the conditions of the assay which is consistent with its classification in Japan and elsewhere.

2.2.6 Bioaccumulation

No experimental data on bioaccumulation is available. With a measured log K_{ow} of 0.35, a low bioaccumulation potential is expected.

2.3 Human Exposure

2.3.1 Occupational Exposure

Hydroxypropyl acrylate is a chemical monomer, manufactured and processed within closed systems. The primary routes of exposure to hydroxypropyl acrylate are skin contact and inhalation. The lower vapor pressure (0.038 hPa at 25° C) helps reduce potential inhalation exposure. In an industrial setting, ingestion is not an anticipated route of exposure.

Exposure may occur during manufacture, transportation and industrial use. Results from workplace measurements at a US production site indicated that hydroxypropyl acrylate did not exceed the ACGIH TLV of 0.5 ppm in 87 samples collected over 20 years. Worker exposure is limited by the use of enclosed processing systems, industrial hygiene controls and personal protective measures. Hydroxypropyl acrylate has a characteristic acrylic odor, which can provide a measure of warning of the presence of hydroxypropyl acrylate vapors.

Loading, unloading, and transportation of tank trucks, railroad tankers, and drums are potential activities that could lead to dermal and inhalation exposure. However, dedicated systems designed to handle hydroxypropyl acrylate are typically used for loading and unloading purposes and procedures are typically in place to prevent spills or leaks during transportation. In addition, the use of personal protective equipment during the loading and unloading operation along with good personal hygiene practices reduces the risk of a potential exposure.

Current occupational exposure limit values for hydroxypropyl acrylate:

2.3.2 Consumer Exposure

It is highly unlikely that any consumer product contains unreacted hydroxypropyl acrylate because hydroxypropyl acrylate is used by chemical manufacturers as a co-monomer in polymers (virtually complete polymerization occurs in the final products) or a reactant to make chemical intermediates. Consumer exposure to hydroxypropyl acrylate is expected to be negligible.

3 HUMAN HEALTH HAZARDS

3.1 Effects on Human Health

3.1.1 Toxicokinetics, Metabolism and Distribution

Studies in Animals

In vivo Studies

No reliable studies were identified for hydroxypropyl acrylate. However, 2-hydroxyethyl acrylate (HEA) has been evaluated in a study that examined metabolism and excretion in male Fischer 344 rat using oral, intraperitoneal, dermal and inhalation routes of exposure (Dow Chemical Company, 1992a). For the oral and intraperitoneal routes of exposure the rats (four animals/dose level/route of exposure) received a single dose of ¹⁴C-HEA at 2.5 or 50 mg/kg body weight. For the inhalation exposure, six rats were exposed to a target concentration of 8 ppm 14 C-HEA for six hours in a head only inhalation chamber. For the dermal exposure, four rats were treated with ¹⁴C-HEA at a dose of 50 mg/kg. No qualitative differences in urinary metabolites between routes were observed, indicating no marked route-dependent differences in the metabolic fate of HEA. The results of the study indicate that once the chemical becomes systemically available it is rapidly metabolized and eliminated from the body as either $CO₂$ in the expired air or urinary metabolites with approximately equal percentage of administered/exposed dose eliminated by each route. The half-lives of elimination of radioactivity were approximately 14 hours for urine and 17 hours for $CO₂$. The halflife of elimination of radioactivity from plasma was approximately 26 hours although the label was not associated with parent HEA. The available metabolic data on HEA is consistent with information on studies with other acrylates where hydrolysis of the ester functionality is the primary metabolic pathway. By analogy with ethyl acrylate and acrylic acid, it is expected that a minor metabolic pathway for HEA will be via conjugation with glutathione with the resulting mercapturic acid derivatives being excreted in the urine. Based on the similarity of the results for HEA with other acrylic acid esters, similar kinetics of hydroxypropyl acrylate is anticipated.

Conclusion

Animal studies with hydroxyethyl acrylate indicated rapid metabolism *via* hydrolysis of the ester functionality with the subsequent rapid metabolism of the hydrolysis products to produce exhaled $CO₂$ or urinary metabolites (mercapturic acid derivatives). There were no marked route-dependent differences in the metabolic fate of hydroxyethyl acrylate when administered by the oral, intraperitoneal, dermal or inhalation routes of exposure.

3.1.2 Acute Toxicity

Studies in Animals

Inhalation

No reliable studies determining the LC_{50} for hydroxypropyl acrylate were identified. When six rats were exposed to a substantially saturated vapor of hydroxypropyl acrylate for eight hours, no deaths occurred (Union Carbide, 1971). When four rats were exposed to saturated vapor of hydroxypropyl acrylate at room temperature for seven hours or for one hour when the test substance was placed in a water bath at 100ºC, no deaths occurred. However, after 4.25 hours of exposure to the material warmed to 100ºC, all five exposed animals died (Dow Chemical Company, 1964).

Dermal

The acute dermal LD_{50} in rabbits was 306 mg/kg body weight (Rohm and Haas, 1983). In this study, groups of six male rabbits received a single occluded application of undiluted hydroxypropyl acrylate for 24 hours at doses of 0, 115, 132, 152, 174, 180, 192, 206, 220, 250, 304, 370, and 450 mg/kg. The skin of half the rabbits in each group was abraded. Mortality was $0/6$, $0/6$, $0/6$, $0/6$, $0/6$, $0/6$, $1/6$, $1/6$, $0/6$, $0/6$, $1/6$, $3/6$, $4/6$ and $6/6$, respectively. The signs of toxicity included passiveness, hypothermia, ataxia, iritis, skin irritation (erythema and edema) and eschar. At necropsy, black foci on stomach mucosa; pitted, mottled kidneys; and tan and congested liver were noted.

Other acute LD_{50} values ranging from approximately 117 to 214 mg/kg were reported as shown in the following table:

Oral

For the key study evaluating the acute oral toxicity of hydroxypropyl acrylate in rats, the LD_{50} was 820 mg/kg body weight (Rohm and Haas, 1983). In this study, groups of 10 male rats were administered hydroxypropyl acrylate at 0 (distilled water, control), 840, 1000, 1410, and 2000 mg/kg and a group of 20 male rats were dosed at 710 mg/kg and the animals were observed for 14 days. The mortality was 0/10, 4/20, 5/10, 9/10, 10/10 and 10/10 for the 0, 710, 840, 1000, 1410 and 2000 mg/kg groups, respectively. The clinical signs included ataxia, salivation, and passiveness. Animals that died during the study showed stained muzzle; lungs, stomach and intestines red and fluid filled; and enlarged stomach.

Other acute LD_{50} values ranging from approximately 252 and 1300 mg/kg were reported in various sources as shown in the following table:

Conclusion

Hydroxypropyl acrylate has a moderate acute toxicity by oral and dermal routes of exposure. It does not pose an acute vapor inhalation hazard at ambient temperatures due to its low vapor pressure.

3.1.3 Irritation

Skin Irritation

Studies in Animals

Hydroxypropyl acrylate is highly irritating to skin but is not a corrosive. The available studies are summarized in the following table:

DOT = Department of Transportation

Conclusion

Hydroxypropyl acrylate is severely irritating to the skin.

Eye Irritation

Studies in Animals

Hydroxypropyl acrylate applied in the eyes of rabbits was moderately to severely irritating. The available studies are summarized in the following table:

Conclusion

Upon eye contact, hydroxypropyl acrylate may cause severe irritation with corneal injury which may result in permanent impairment of vision, even blindness.

3.1.4 Sensitisation

Studies in Animals

Skin

Hydroxypropyl acrylate has been tested in a number of studies using experimental animals. These studies are summarized in the following table:

HPA = Hydroxypropyl acrylate; HEA = Hydroxyethyl acrylate;

HEMA = Hydroxyethyl methacrylate; HPMA = Hydroxypropyl methacrylate

Conclusion

Hydroxypropyl acrylate may cause skin sensitization in experimental animals.

Studies in Humans

Skin

In human patch tests, three out of 24 patients showing irritation to initial exposure to acrylates including hydroxypropyl acrylate, showed sensitization reactions to hydroxypropyl acrylate exposure (Kanerva *et al.*, 1988).

A number of case reports (see IUCLID Dossier) are presented for exposure to hydroxypropyl acrylate. These reports suggest that hydroxypropyl acrylate is likely to cause sensitization and cross-react with other acrylates in some exposed individuals. The human exposure to hydroxypropyl acrylate, however, is extremely limited since the only potential for repeated exposure is in workplace environments that have stringent protective procedures to avoid contact with the highly irritating chemical.

Conclusion

Hydroxypropyl acrylate may cause skin sensitization in humans. Cross-reactivity with and to other acrylate exposure is possible.

3.1.5 Repeated Dose Toxicity

Studies in Animals

Several studies, summarized below, were conducted with male animals of several species via inhalation exposure. In addition, a 28-day repeated-dose toxicity study and a 18-month chronic toxicity study on the closely related, hydroxyethyl acrylate (CAS RN 818-61-1) are used to address this endpoint.

Inhalation

Studies with Hydroxypropyl Acrylate

An inhalation study with rats, mice, rabbits and dogs was conducted with hydroxypropyl acrylate (Quast and Rampy 1983). Three groups of animals per species, consisting of 10 male Sprague-Dawley rats, 20 male Swiss-Webster mice, 4 male New Zealand white rabbits, or 2 male beagle dogs, were used in the study. The animals were exposed (whole body) to hydroxypropyl acrylate vapors at 0 (unexposed control), 5 ppm (27 mg/m^3) or 10 ppm (53 mg/m^3) . Exposures were 6 hours/day, 5 days/week for a total of 21 exposures for the rats and mice and 20 exposures for rabbits and dogs. Food and water were available at all times except during exposure. Food and water were removed from the control animals for 6 hours on the first day of exposure only. The tissues examined for each species included adrenal glands, parathyroid glands, thyroid glands, aorta, sternum, bone marrow, kidneys, urinary bladder, prostate, accessory sex glands, epididymides, testes, urethra, brain, pituitary gland, spinal cord, peripheral nerve, large intestine, small intestine, esophagus, salivary gland, stomach, pancreas, liver, lung, skin, larynx, trachea, nasal turbinates, heart, lymph nodes, spleen, thymus, skeletal muscle, mammary tissue, gall bladder (except rats) and eyes.

In the rat study, no mortality was observed. Five of 10 rats exposed to 10 ppm developed slight cloudiness of the cornea. No treatment-related decreases in mean body weights were found, and the absolute and relative organ weights were not affected. No treatment-related effects on the hematological measurements, red cell counts, white cell counts, differential white cell counts,

hemoglobin concentration, and packed cell volume, the clinical chemistry measurements, BUN, glucose, SGPT, and alkaline phosphatase or the urinalysis parameters, specific gravity, pH, glucose, protein, and the presence or absence of ketones, bilirubin and blood were identified. At necropsy, focal corneal cloudiness was noted in 8 of 10 rats from the high exposure group suggesting an irritant effect. Histologically, a small increase in the number of animals with changes in the lungs at 10 ppm (subacute pneumonitis) and the nasal mucosa (both concentrations) was observed. No other treatment-related effects in the examined tissues were observed and no systemic toxicity was identified. The LOAEC was 5 ppm (27 mg/m^3) based on the effects in the nasal mucosa.

In the mouse study, no mortality was observed. Three of 20 mice exposed to 10 ppm, but none exposed to 5 ppm showed signs of eye irritation. No compound-related, statistically significant effects on mean body weights were found. However, the high dose mice showed a slightly lower body weights during weeks 1, 3 and 4 of exposure (up to 4.5% lower in week 1) which recovered during the weekend without treatment. No gross or histopathological changes related to exposure were observed. The NOAEC was 5 ppm (27 mg/m^3) .

In the rabbit study, no mortality was observed. All rabbits exposed to 10 ppm and 2 of 4 animals exposed to 5 ppm developed slight rhinitis and eye irritation (moderate conjunctivitis). No treatment-related effects on mean body weight were observed and all groups gained a similar amount of weight during the study. The absolute and relative organ weights were not affected. No treatment-related effects on the hematological measurements, red cell counts, white cell counts, differential white cell counts, hemoglobin concentration, and packed cell volume or the clinical chemistry measurements, BUN, glucose, SGPT, SGOT and alkaline phosphatase were identified. Gross and histopathological changes related to exposure were found in the upper respiratory system (rhinitis, squamous metaplasia, and ulcerations), trachea and lungs (bronchitis, focal pneumonitis) of both hydroxypropyl acrylate exposure groups. The tracheal changes varied from no effect in some low dose animals to focal ulceration, squamous metaplasia and tracheitis in the high dose group. The most marked effects were seen in the upper respiratory system of all rabbits, especially those in the 10 ppm group that were characterized by mucopurulent rhinitis with squamous metaplasia and ulceration of the nasal turbinate mucosa. The LOAEC was 5 ppm (27 mg/m^3) based on effects on the upper respiratory system.

In the dog study, no mortality was observed. During the exposures, both dogs exposed to 10 ppm lost body weight (from 2 to 10%) and exhibited nasal irritation (exudative rhinitis) and eye irritation (bilateral corneal cloudiness, slight corneal edema, and bilateral suppurative conjunctivitis). The body weights showed some recovery during the weekends when no exposures occurred. One dog exposed to 5 ppm exhibited exudative rhinitis approximately half way through the study. At necropsy, exudative rhinitis, tracheitis, and suppurative bronchopneumonia were observed in all hydroxypropyl acrylate-exposed dogs. Microscopic lesions included suppurative rhinitis, squamous metaplasia and hyperplasia of the lining epithelium and focal area of ulceration in the mucosa of the nasal turbinates. These lesions extended to the trachea and lungs of the high exposure group resulting in bronchopneumonia. The LOAEC was 5 ppm (27 mg/m^3) based on effects on the upper respiratory system.

Studies with Hydroxyethyl Acrylate

A 28-day inhalation study with hydroxyethyl acrylate in male rats at concentrations of 0, 5, 10 and 25 ppm $(0, 24, 48$ and $120 \text{ mg/m}^3)$ was reported (Leong and Trice, 1970). Groups of 15-20 animals (5 –10 animals sacrificed at 10 days) were exposed 7 hours/day for five days/week. The 25 ppm group was exposed for only 10 days due to excessive toxicity. Eight animals in the 25 ppm group died in the first 10 days and 9 additional died after cessation of dosing. One animal in the 10 ppm group died. At 5 ppm, only irritation of the cornea was observed. Ulcerative corneal changes, nasal irritation and decreased body weight were found in the 10 ppm survivors. However, the animals

were able to recover weight during the weekend when no exposures occurred. In the high exposure group, nasal irritation and sever respiratory distress occurred within two days. Thereafter, the animals lost weight and died of respiratory failure. It was concluded that the respiratory system and the eyes are the only affected systems following vapor exposure. The LOAEC based on minimal irritation was 5 ppm (24 mg/m^3) . This value compares well with the minimal irritation observed at 10 ppm (54 mg/m^3) in rats exposed to hydroxypropyl acrylate under a similar regimen.

In a chronic inhalation study, male and female Sprague-Dawley rats (99 or 100 animals per dose group) were exposed to hydroxyethyl acrylate 6 hours per day, 5 days/week for 18 months at doses of 0.5 ppm (2.4 mg/m^3) or 5 ppm (24 mg/m^3) (Kociba *et al.*, 1979). The control group, consisting of 100 animals of both sexes, was exposed to air. After termination of treatment, the male and female animals were left for a recovery period of 5 and 6 months respectively before being killed for examination *post mortem*. The study included a 12-month interim kill for clinical and histopathological evaluation. Body weights, terminal organ weights and cumulative mortality, urinalysis, clinical chemistries and hematology did not appear to be altered by chronic hydroxyethyl acrylate exposure. Treatment was not associated with adverse effects except that the rats in the 5 ppm treatment group developed yellow staining of the fur and a marginal increase in *Mycoplasma*induced pneumonia which was interpreted as being treatment related. No treatment-related effects were seen in the 0.5 ppm group. Overall, chronic inhalation exposure to hydroxyethyl acrylate at a dose of 5 ppm caused only a minimal toxicological effect while no toxicity was seen at 0.5 ppm. Gross and histopathological examination of tissues showed no indication of chronic toxicity or a carcinogenic effect in either the 5 or 0.5 ppm treatment groups, however nasal tissues were not examined. In addition, animals were treated for 18 months and, although maintained untreated for a further 6 months (approximately), the study does not meet current guideline requirements of a 24 month exposure period. Despite these and other limitations (e.g. bacterial infection), the data provide some evidence that hydroxyethyl acrylate was not carcinogenic *via* inhalation, which is potentially a significant route of occupational exposure.

Conclusion

Hydroxypropyl acrylate and the closely related, hydroxyethyl acrylate, are highly irritant at vapor concentrations greater than approximately 10 ppm (approximately 50 mg/m³) resulting in severe localized effects and, in some cases, death due to respiratory distress. Exposure below lethal concentrations results in concentration-related irritant effects of the nasal passages and respiratory system. The systemic toxicity for either chemical was minimal with all of the effects observed in various studies related to the irritant properties.

3.1.6 Mutagenicity

Studies in Animals

In vitro Studies

Hydroxypropyl acrylate was tested in the bacterial reverse mutation assay test (OECD Guideline 471) with *Salmonella typhimurium* TA98, TA100, TA1535, and TA1537 at concentrations up to 5000 µg/plate with and without metabolic activation. Hydroxypropyl acrylate was considered not mutagenic under the conditions of the assay (Roehm, 1995a). Hydroxypropyl acrylate was positive in two *E. coli* strains (WP2UVrA/PKM101 and WP2/PKM101) and negative in two *Salmonella typhimurium* strains (TA 102 and TA 2638) when tested only with metabolic activation (Watanabe *et al.*, 1996).

A mammalian cell (CHO V79) mutation assay (OECD Guideline 476) was negative with and without metabolic activation at concentrations up to 150 µg/ml. Cytotoxicity was observed in this assay, however, at concentrations as low as 30 µg/ml (Roehm, 1995c).

Hydroxypropyl acrylate was tested in a cytogenetic assay (CHO V79 cells) according to OECD Guideline 473 at concentrations up to 100 µg/ml and was positive with and without metabolic activation in this assay (Roehm, 1995b).

In a chromosomal aberration assay (CHO cells), hydroxypropyl acrylate up to 5000 μ g/ml was positive for chromosomal damage. Cytotoxicity was observed at all concentrations in the assay with viable cells ranging from 3 to 75% of control in a concentration-related manner (Roehm, 2000).

Other (meth)acrylates, including acrylic acid, methyl (meth)acrylate, ethyl (meth)acrylate, 2 ethylhexyl acrylate, 2-hydroxyethyl acrylate, and several multifunctional (meth)acrylates (Moore and Doerr, 1990), have been evaluated in *in vitro* mutagenicity assays with mammalian cells. In these studies, positive results were reported at concentrations that led to a clearly reduced cell survival rate. Studies have indicated that there is an association between chromosomal aberrations and cytotoxicity at exposure concentrations which reduce cell growth to less than 50% of the control value (Galloway, 2000 and references cited therein). These data suggest that the increase in mutagenicity reported in some of the chromosomal aberration assays with hydroxypropyl acrylate may be an artifact of the experimental method.

In vivo Studies

A mouse micronucleus study with hydroxypropyl acrylate conducted in compliance with OECD Guideline 474 (Hamann, 2000) indicated that, as with other acrylates (e.g. methyl acrylate and ethyl acrylate; see respective SIAR documents) , hydroxypropyl acrylate is not mutagenic in this assay and further support the conclusion above that the positive *in vitro* findings noted above in in vitro studies are possibly an artifact of the assay design and cytotoxicity of the test chemical.

Conclusion

Based on the results of the reliable studies, hydroxypropyl acrylate shows no potential to induce gene mutations, at least in mammalian cells, and no potential to induce chromosomal aberrations *in vivo*.

3.1.7 Carcinogenicity

In vivo Studies in Animals

No carcinogenicity studies were identified for hydroxypropyl acrylate. As reviewed for Repeated Dose toxicity, no evidence of carcinogenicity was observed for the closely related, hydroxyethyl acrylate, following 18 months of inhalation exposure (Kociba *et al.*, 1979; Rampy *et al.*, 1978).

3.1.8 Toxicity for Reproduction

Studies in Animals

Effects on Fertility

No reproductive toxicity studies are available. In the inhalation studies described under Repeated Dose Toxicity (Quast and Rampy, 1983), the prostate, accessory sex glands, epididymides, and testes, were examined for males of four species (rats, mice, rabbits, and dogs) exposed to hydroxypropyl acrylate vapor at 5 or 10 ppm $(27 \text{ or } 53 \text{ mg/m}^3)$. No effects on any reproductive organs were identified following 20 or 21 exposures to concentrations that resulted in significant irritation of the nasal passages and respiratory system. Similarly no effects on reproductive organs were seen with the closely related hydroxyethyl acrylate following a 28-day exposure study in males at 5, 10, or 25 ppm $(24, 48, \text{ or } 120 \text{ mg/m}^3)$, respectively) (Leong and Trice, 1970).

As part of the chronic inhalation study described above (Kociba *et al*., 1979) a detailed pathological examination of the male and female reproductive organs from the animals on study was conducted. The analysis indicated that the female rats in the 5 ppm group showed an increased incidence of uterine inflammation as compared to the negative control animals. The incidence of uterine inflammation was 2/21, 1/3 and 11/27 for controls, 0.5 and 5 ppm groups, respectively. No other statistically significant differences for histopathologic observations of the female reproductive organs were found, including the ovaries.

An evaluation of the histopathological data from the male animals exposed to 5 ppm indicated a statistically significant increase from the controls in the incidence of fibrinoid degeneration in the vascular channels of the testes which was a local vascular manifestation of mesenteric periarteritis syndrome observed as age-related lesion in this rat strain (8/14 in controls *vs.* 17/19 for the 5 ppm group). The authors of the study indicate that the fibrinoid degeneration in the testes was not a substance-specific toxic effect as the laboratory conducting this study commonly observed this lesion in aging rats of this strain at similar incidence as was observed in this study (the historical control incidence of this lesion from seven chronic toxicity/oncogenicity studies ranged from 37 to 85%). Polyarteritis (polyarteritis or periarteritis nodosa) is the most conspicuous inflammatory lesion of the blood vessels of rats. The etiology is unknown and the incidence varies among strains and colonies (Mitsumori, K. (1990) Chapter 29 in Pathology of the Fischer Rat. Eds: Boorman et al., Academic Press, Inc. p 477). Common sites in male rats are the arteries of the testicle and to a lesser extent the arteries of the spermatic cord (Burek, J.D. (1978) Pathology of the Aging Rat, CRC Press p. 87). Carlton and Engelhardt (Polyarteritis, In: Cardiovascular and Musculoskeletal Systems Eds: Jones, T.C., Mohr, U. and Hunt, R.D., Springer-Verlag, 1991, p 71) also indicate that this lesion can be present in spermatic arteries.

Developmental Toxicity

The developmental toxicity of hydroxypropyl acrylate was evaluated in Sprague-Dawley rats after inhalation exposure for 6 hours/day, on gestation days 6 to 20 (Saillenfait *et al.*, 1999). The exposure concentrations were $0, 1, 5,$ or 10 ppm $(0, 5.3, 27,$ or 53 mg/m³). Dose groups consisted of 20 to 22 pregnant rats. Maternal body weights were lower than control for the 10 ppm group and weight gain was reduced for the 5 and 10 ppm groups. There was no significant difference in the numbers of implantation sites and live fetuses, in the incidence of non-live implants and resorptions, or in the fetal sex ratio, or fetal body weight between control and treated animals. No treatmentrelated increase in embryo/fetal lethality or fetal malformations was observed at any dose level. The incidence of external, visceral, and skeletal variations was similar to controls. The NOAECs were 1 ppm (5.3 mg/m³) for maternal toxicity and >10 ppm (53 mg/m³) for embryo-fetal toxicity and teratogenicity.

The closely related hydroxyethyl acrylate was similarly tested in the study design and at the exposure concentrations described above for hydroxypropyl acrylate (Saillenfait *et al.*, 1999). Exposure to 10 ppm hydroxyethyl acrylate caused maternal toxicity evidenced by a transient decrease in body weight change, a decrease in absolute weight gain and a continuous reduction of food consumption during exposure. There were no effects in dams exposed to 5 ppm hydroxyethyl acrylate. The NOAECs were 5 ppm (24 mg/m³) for maternal toxicity and > 10 ppm (48 mg/m³) for embryo-fetal toxicity and teratogenicity.

Conclusion

Based on results of animal studies evaluating exposure to hydroxypropyl acrylate and hydroxyethyl acrylate vapors, no reproductive toxicity is anticipated for hydroxypropyl acrylate. Hydroxypropyl acrylate is not selectively toxic to the embryo or fetus and is not teratogenic via inhalation exposure. This conclusion is supported by a similar study with the analog chemical, hydroxyethyl acrylate.

3.2 Initial Assessment for Human Health

Studies with experimental animals indicate that the acute oral and dermal toxicity of hydroxypropyl acrylate is moderate. The acute oral and dermal LD_{50} values for hydroxypropyl acrylate are 820 mg/kg (male rats) and 306 mg/kg (rabbits), respectively. In an acute inhalation study, exposure to a saturated vapor for eight hours did not result in deaths. In standard primary irritation studies, hydroxypropyl acrylate is a severe skin and eye irritant. It may cause corneal injury which may result in permanent vision impairment or blindness. Skin sensitization studies in animals and humans indicate that hydroxypropyl acrylate is likely to be a sensitizer and will cross-react with other acrylates in some exposed individuals. Repeated dose studies in rats via inhalation with hydroxypropyl acrylate and the closely related, hydroxyethyl acrylate, indicate that systemic toxicity is minimal. Exposure to vapors of these two monomers results in severe irritation of the upper respiratory tract, resulting in death due to respiratory failure at higher concentrations and concentration-related local irritation at sublethal exposures. The LOAEC for subchronic exposure, based on irritation, was 10 ppm (53 mg/m³) for hydroxypropyl acrylate and 5 ppm (24 mg/m³) for hydroxyethyl acrylate. Only minor irritant effects, no systemic toxicity, including effects on reproductive organs, and no carcinogenicity was observed following 18 months exposure to 5 ppm of hydroxyethyl acrylate.

Hydroxypropyl acrylate was not mutagenic to *Salmonella typhimurium* (Salmonella reverse mutation assay) *in vitro* with or without metabolic activation but was found to be positive with metabolic activation when tested with two *E. coli* strains. In mammalian cells *in vitro*, hydroxypropyl acrylate was found to be negative in a gene mutation assay but had clastogenic activity in cytogenetic and chromosomal aberration assays. In these mammalian cell assays, positive results occurred only at concentrations that resulted in significant cell death. Thus, the positive results are considered equivocal. Hydroxypropyl acrylate was not mutagenic in an *in vivo* mouse micronucleus study. Overall, hydroxypropyl acrylate is considered not to have mutagenic potential *in vivo* based on available data.

Hydroxypropyl acrylate had no effect on male reproductive organs in four species following repeated exposures via inhalation for 30 or 31 days (20 or 21 exposures). The principal treatmentrelated effects observed following 18 months exposure of laboratory rats to 5 ppm of the closely related, hydroxyethyl acrylate, were related to irritation of the respiratory tract, without significant evidence of systemic toxicity. Histopathological examination of the reproductive organs revealed effects in the arteries of the testes (fibrinoid degeneration in the vascular channels – an age-related lesion) and uterus (inflammation) which were interpreted as not treatment-related nor adverse to reproduction. No evidence of a carcinogenic effect was observed. In a well-conducted inhalation study exposing pregnant rats to hydroxypropyl acrylate from gestation day 6 to 20, maternal body weights were reduced at the two highest exposure concentrations (5 and 10 ppm; 27 and 53 mg/m³), but no embryo-fetal or developmental toxicity or teratogenicity was observed. Overall, based on the available animal studies, hydroxypropyl acrylate is not toxic to reproduction or development and is not teratogenic.

Knowledge of the effects of hydroxypropyl acrylate exposure in humans is minimal and based primarily on isolated incidence reports. These reports along with animal data indicate the primary hazard from exposure to hydroxypropyl acrylate in humans is irritation. It is anticipated that there

is potential for sensitization from hydroxypropyl acrylate exposure and cross reactivity with other acrylates in some exposed individuals. However, use of personal protective equipment during production prevents exposure and there is virtually no exposure to residual hydroxypropyl acrylate following polymerization.

4 HAZARDS TO THE ENVIRONMENT

4.1 Aquatic Effects

Acute Toxicity Test Results

Acute toxicity to fish (Geiger *et al.*, 1986; Russom *et al.*, 1988) yielded a 96-hour LC₅₀ for fathead minnows of 3.1 mg/L (NOEC = 1.8 mg/L). This study followed ASTM (1980) guidelines using a flow-through design. Nominal exposure concentrations ranged from 1.8 to 9.8 mg/L and analyses at 96 hours ranged from 1.2 to 10.6 mg/L. Mortality occurred in the three highest concentrations $(3.99, 6.54 \text{ and } 10.6 \text{ mg/L}, \text{measured})$. The LC₅₀ was determined based on analytical values.

An acute toxicity study with *Daphnia magna* (Dow Chemical Company, 1992b) indicated the 48 hour EC₅₀ was 24 mg/L (NOEC = 10 mg/L). This study followed OECD Guideline 202 and Directive 84/449/EEC, C.2 using a static design. Nominal exposure concentrations ranged from 1.0 to 100 mg/L; no analytical confirmation was performed. Immobilization occurred in the four highest concentrations (18, 32, 56 and 100 mg/L, nominal).

A toxicity study with *Pseudokirchneriella subcapitata* (formerly known as *Selenastrum capricornutum*) (Scheerbaum, 2004) indicated the 96-hour EC₅₀ values for biomass and growth rate were 3.53 mg/L and 6.67 mg/L, respectively and the 96-hour NOECs were 0.625 mg/L and 5 mg/L, respectively. This study followed OECD Guideline 201 using a static design. Nominal exposure concentrations ranged from 0.625 to 10 mg/L. Growth inhibition of algae was observed at concentrations of 2.5 mg/L and greater. The EC_{50} values were determined based on nominal values.

Toxicity to Microorganisms

Toxicity to Microorganisms: Seven studies (see Robust Summaries) examining the bacterial inhibition of hydroxypropyl acrylate are reported in different species. The MIC values range from 20 to 10,000 mg/L. Based on the broad range of results, no conclusion as to the potential toxicity of hydroxypropyl acrylate to specific microorganisms can be made.

4.2 Terrestrial Effects

No data available

4.3 Other Environmental Effects

No data available

4.4 Initial Assessment for the Environment

Although hydroxypropyl acrylate is produced in high volumes, releases to the environment are small. Hydroxypropyl acrylate has a low vapor pressure and would not be expected to extensively volatilize in the environment. Based on the Level III fugacity modeling, the majority of the releases are expected in water and soil with little in the air or sediment compartments. Modeling data indicate that hydroxypropyl acrylate is degraded rapidly in air with a half-life of 7.4 hours. Based on the OECD Guideline 301C study, hydroxypropyl acrylate is readily biodegradable (83% degraded over 28 days).

Hydroxypropyl acrylate is toxic to aquatic organisms. The LC_{50} value for fish in standard laboratory tests is approximately 3.1 mg/L and the EC_{50} value for daphnia is 24 mg/L and for biomass and growth rate of algae are 3.53 and 6.67 mg/L, respectively. Due to the low log K_{ow} of 0.35, the bioaccumulation potential is very limited.

5 RECOMMENDATIONS

Human Health

The chemical is currently of low priority for further work. The chemical has properties indicating a hazard for human health (severe eye irritation with corneal injury, which may result in permanent impairment of vision, even blindness, skin and upper respiratory tract irritation, potential skin sensitization, and acute toxicity from inhalation exposure). Based on exposure data presented by the Sponsor Country (relating to production in one country which accounts for approximately 80% of global production and relating to the use pattern in several countries), this chemical is currently of a low priority for further work. Countries may wish to investigate any exposure scenarios that were not presented by the Sponsor country.

Environment

This chemical has properties, indicating a hazard for the environment (fish, invertebrate, algae). However, the chemical is of low priority for further work for the environment because of its ready biodegradability and the limited potential for bioaccumulation.

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S I D S

Dossier

1.0.1 APPLICANT AND COMPANY INFORMATION

27.12.2004

1.0.2 LOCATION OF PRODUCTION SITE, IMPORTER OR FORMULATOR

1.0.3 IDENTITY OF RECIPIENTS

1.0.4 DETAILS ON CATEGORY/TEMPLATE

1.1.0 SUBSTANCE IDENTIFICATION

1.1.1 GENERAL SUBSTANCE INFORMATION

02.01.2004

1.1.2 SPECTRA

1.2 SYNONYMS AND TRADENAMES

1,2-Propanediol, monoacrylate

02.01.2004

1,2-Propanediol-1-acrylate

02.01.2004

2-Propenoic acid,2-hydroxypropyl ester

02.01.2004

Acrylic acid, 2-hydroxypropyl ester

02.01.2004

Acrylic acid, monoester with propylene glycol

02.01.2004

beta-Hydroxypropyl acrylate

02.01.2004

Hydroxypropyl acrylate

02.01.2004

Propylene glycol acrylate

02.01.2004

Propylene glycol monoacrylate

02.01.2004

1.3 IMPURITIES

02.01.2004

1.4 ADDITIVES

27.12.2004

1.5 TOTAL QUANTITY

1.6.1 LABELLING

1.6.2 CLASSIFICATION

1.6.3 PACKAGING

1.7 USE PATTERN

Remark : Hydroxypropyl acrylate is mainly used either as a co-monomer in the manufacture of polymers or as a chemical reactant in the manufacture of chemical intermediates. In the manufacture of polymers, hydroxypropyl acrylate can be co-polymerized with acrylic acid, acrylates, methacrylates,

- **1.9.2 COMPONENTS**
- **1.10 SOURCE OF EXPOSURE**
- **1.11 ADDITIONAL REMARKS**
- **1.12 LAST LITERATURE SEARCH**
- **1.13 REVIEWS**

2.1 MELTING POINT

2.2 BOILING POINT

2.3 DENSITY

2.3.1 GRANULOMETRY

2.4 VAPOUR PRESSURE

2.6.2 SURFACE TENSION

- **2.7 FLASH POINT**
- **2.8 AUTO FLAMMABILITY**
- **2.9 FLAMMABILITY**
- **2.10 EXPLOSIVE PROPERTIES**
- **2.11 OXIDIZING PROPERTIES**
- **2.12 DISSOCIATION CONSTANT**
- **2.13 VISCOSITY**
- **2.14 ADDITIONAL REMARKS**
3. ENVIRONMENTAL FATE AND PATHWAYS ID: 25584-83-2

3.1.1 PHOTODEGRADATION

3.1.2 STABILITY IN WATER

formic acid were added prior to analysis to adjust the dample to the pH range of 5 to 6 to minimize further hydrolysis. The following sampling schedule is described in the TSCA guidelines:

Procedure 1- If 60-70% conversion occurs within 28 days, then a minimum of six measurements will be made at regular intervals between 20 and 70% hydrolysis.

Procedure 2- If the reaction is too slow to conveniently follow the hydrolysis to a high conversion in 28 days, but is still rapid enough to attain at least 20% conversion, then the test solution should be analyzed at 15- 20 time points at regular intervals after 10% conversion is attained.

Procedure 3- If less than 20% conversion occurs after 28 days, then the concentration of test chemical after 28 days will be determined, and a halflife of >x days reported.

For each hydrolysis experiment, the natural logarithm of the test substance concentration was plotted as a function of time. At a constant pH, a straight line was obtained, indicating pseudo-first order kinetics. The slope of the linear regression line was equal to -Kh, where Kh was the pseudo-first order rate constant. Using the relationship T1/2=ln 2/Kh, the half-life of the hydrolysis reaction was determined. The following relationship holds for hydrolysis reactions in buffered systems:

Kh=Ka[H+]+Kb[OH-]+Kn where Ka, Kb, and Kn are the second-order rate constants for acid and base catalyzed, and neutral water hydrolysis reactions, respectively, and Kh is the measured pseudo-first order rate constant. At a given pH, the equation contains three unknowns, Ka, Kb, and Kn; therefore, three equations are required to determine the three unknown values. This was accomplished by measuring the hydrolysis rates at pH 3, 7 and 11.

(a) pseudo-first-order rate constant determined at indicated pH

10.90 12.27 0.056 0.9983

3.2.1 MONITORING DATA

3.2.2 FIELD STUDIES

3.3.1 TRANSPORT BETWEEN ENVIRONMENTAL COMPARTMENTS

3.3.2 DISTRIBUTION

3. ENVIRONMENTAL FATE AND PATHWAYS ID: 25584-83-2 DATE: 23.08.2005 Soil (Level I) = 0.2% **Reliability** : (2) valid with restrictions **Flag : Critical study for SIDS endpoint** 23.08.2005 (6) **Media** : other: air (emission rate to air of 1000 kg/hr) **Method : Calculation according Mackay, Level III Year :** 2003 **Remark :** The EPIWIN model (v3.11) was run using the following physical chemical properties: Vapor Pressure (mm Hg) : 0.028 Log Kow (octanol-water): 0.35 Boiling Point (deg C) : 205.70 Melting Point $(\text{deg } C)$: -23.35 **Result** : Concentration (%): Air - 1.4 Water - 27 Soil - 72 Sediment - < 0.1 Level III Fugacity Model (Full-Output): ======================================= Chem Name : 2-Propenoic acid, monoester with 1,2-propanediol Molecular Wt: 130.14 Henry's LC : 1.58e-008 atm-m3/mole (Henrywin program) Vapor Press : 0.028 mm Hg (user-entered) Log Kow : 0.35 (user-entered) Soil Koc : 0.918 (calc by model) Mass Amount Half-Life Emissions (percent) (hr) (kg/hr) Air 1.38 13.5 1000 Water 26.6 360 0 Soil 71.9 360 0 Sediment 0.0452 1.44e+003 0 Fugacity Reaction Advection Reaction Advection
(atm) (kg/hr) (kg/hr) (percent) (percent) (atm) (kg/hr) (kg/hr) (percent) (percent) Air 8.62e-012 235 45.9 23.5 4.59 5.37e-014 170 88.4 17 Soil 5e-012 460 0 46 0
Sediment 4.46e-014 0.0723 0.003 0.00723 0.0 Sediment 4.46e-014 0.0723 0.003 0.00723 0.0003 Persistence Time: 332 hr Reaction Time: 384 hr Advection Time: 2.47e+003 hr Percent Reacted: 86.6 Percent Advected: 13.4 Half-Lives (hr), (based upon Biowin (Ultimate) and Aopwin): Air: 13.52 Water: 360 Soil: 360 Sediment: 1440 Biowin estimate: 3.212 (weeks) Advection Times (hr):

OECD SIDS HYDROXYPROPYL ACRYLATE

Air: 100
Water: 1000 Water:

OECD SIDS HYDROXYPROPYL ACRYLATE

OECD SIDS HYDROXYPROPYL ACRYLATE

DATE: 23.08.2005

3.4 MODE OF DEGRADATION IN ACTUAL USE

3.5 BIODEGRADATION

OECD SIDS HYDROXYPROPYL ACRYLATE 3. ENVIRONMENTAL FATE AND PATHWAYS ID: 25584-83-2

OECD SIDS HYDROXYPROPYL ACRYLATE 3. ENVIRONMENTAL FATE AND PATHWAYS ID: 25584-83-2

- **3.6 BOD5, COD OR BOD5/COD RATIO**
- **3.7 BIOACCUMULATION**
- **3.8 ADDITIONAL REMARKS**

4.1 ACUTE/PROLONGED TOXICITY TO FISH

equilibrium prior to death.

musculature, and showed signs of edema, were deformed, and lost

Cumulative Mortality (total number of animals in each group = 20):

Number of animals with effects (total number of animals in each group = 20):

These results are from a second test, the first test resulted in an LC50

value of 3.61 mg/L.
Species: P. pro Test condition : Species: P. promelas, Age: 28-34 days, Weight: 0.092 +/- 0.030 g,
Length: 19.4 +/- 2.41 mm, $19.4 + (-2.41$ mm, Loading: 0.92 g/l Test medium: filtered Lake Superior water, Water quality parameters as measured during the test: Temperature = 24.7 degrees C; Dissolved oxygen = 6.7 mg/l; pH = 7.87; Total hardness = 45.9 mg/l CaCO3; and Total alkalinity = 45.0 mg/L CaCO3. Test substance : 2-hydroxypropyl acrylate, Scient. Polymer Prod., purity >97% **Reliability** : (1) valid without restriction Equivalent to guideline **Flag** : Critical study for SIDS endpoint 04.04.2003 (23) (45)

4.2 ACUTE TOXICITY TO AQUATIC INVERTEBRATES

4.3 TOXICITY TO AQUATIC PLANTS E.G. ALGAE

OECD SIDS HYDROXYPRO 4. ECOTOXICITY

4.4 TOXICITY TO MICROORGANISMS E.G. BACTERIA

4.9 ADDITIONAL REMARKS

5.0 TOXICOKINETICS, METABOLISM AND DISTRIBUTION

5.1.1 ACUTE ORAL TOXICITY

5.1.2 ACUTE INHALATION TOXICITY

5.1.3 ACUTE DERMAL TOXICITY

Data were audited by QAU but no reference to GLP standard.

5.2.2 EYE IRRITATION

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5.4 REPEATED DOSE TOXICITY

Statistical analyses of hematological, biochemical and body weight data were carried out using an analysis of variance and Dunnett's test (p<0.05).

OECD SIDS HYDROXYPROPYL ACRYLATE 5. TOXICITY ID: 25584-83-2

Route of admin.

ID: 25584-83-2

over the weekends. No effects on absolute and relative body weights at 5

290

a- exposure terminated because of high mortality

-Pathology findings: HEA produced ulcerative keratitis and chronic-active tracheitis at 5, 10 and 5 ppm. Focal ulcerative rhinitis and chronic-active laryngitis resulted from HEA exposure at 10 and 25 ppm. Lesions at the 25 ppm level were more severe than those at 10 and 5 ppm. Bronchopneumonia and severe upper respiratory lesions were responsible for the spontaneous deaths at the 25 ppm exposure. The 14-day recovery period did not significantly reduce the number of lesions observed, except for the absence of ulcerative rhinitis.

-Gross Lesions:

a- One rat died after the 14 exposure due to diffuse pneumonia. b- One rat was euthanized after the 15th exposure due to otitis media.

-Microscopic Lesions:

a- One rat died after the 14 exposure due to necrotizing bronchopneumonia.

b- One rat was euthanized after the 15th exposure, the lesions observed were purulent otitis media, pneumonia and tracheitis.

Test condition

ul S9 mix or S9 mix substitution buffer and 100 ul bacterial suspension were mixed in a test tube and incubated at 37 degrees C for 60 minutes. After pre-incubation, 2.0 ml overlay agar was added to each tube. The mixture was poured on minimal agar plates. After solidification, the plates were incubated upside down for at least 48 hours at 37 degrees C in the dark. Colonies were counted using an automatic counter.

A test article was considered positive if either a dose-related and

Experiment I: without S9 mix: 1, 3, 10*, 13, 17 and 20 ug/ml with S9 mix: 3, 10*, 30, 60*, 100 and 150 ug/ml

Confirmatory, Experiment II: without S9 mix: 1, 10*, 20, 30, 50, 70** and 100** ug/ml with S9 mix: 3, 30*, 60, 100 and 150 ug/ml respectively.

* = not evaluated, culture not continued ** = not able to evaluate, toxic effects.

Solvent control (distilled water), untreated control and positive control groups were also included. The positive control substance utilized for tests without metabolic activation was ethylmethanesulfonate. 7,12 dimethylbenz(a)anthracene was the positive control substance utilized for tests with metabolic activation.

The S9 liver microsomal fraction was obtained at the testing facility from the livers of 8-12 week old male Wistar rats which received a single i.p. injection of 500 mg/kg b.w. Aroclor 1254 in olive oil five days previously. The protein concentration in the S9 preparation was 33.2 mg/ml. Before the experiment, an appropriate quantity of S9 supernatant was thawed and mixed with S9 co-factor solution.

Three-day old exponentially growing stock cultures were used. Approximately 500 cells for toxicity or 1.5 E06 cells for mutation rate were seeded. After 24 hours (Day 1), the medium was replaced with serum-free medium containing the test article with or without S9 mix. After 4 hours, the medium was replaced with untreated medium. The flasks were subcultured and incubated at 37°C (4.5% carbon dioxide) for 16 (toxicity) or 18 (mutation rate) days. At termination, the cells were fixed and stained with 10% methylene blue in 0.01% KOH. The stained colonies with more than 50 cells were counted.

A test article was considered positive if it induced either a concentrationrelated increase of the mutant frequency or a reproducible and positive response for one of the test points. A test article producing neither a concentration-related increase of the mutant frequency nor a reproducible positive response at any of the test points was considered non-mutagenic in this system.

A significant response is described as follows:

A test article was considered mutagenic if it induced a reproducible mutation frequency that is at least three times higher than the spontaneous mutation frequency in the experiment at one or more of the concentrations. The test article is classified as mutagenic if there is a reproducible concentration-related increase of the mutation frequency regardless of whether the 3-fold increase was observed. However, if there is a low spontaneous mutation rate in the normal range for the negative control, this is taken into account in determining whether the test substance is mutagenic in this system.

Result : In the pre-test toxicity experiment, reduced cell viability was observed at concentrations greater than 10 ug/ml without metabolic activation and greater than 100 ug/ml with metabolic activation. In the first experiment, the number of mutant colonies was similar across the solvent control (20.8 mutants/1E06 cells) and treated (with or without metabolic activation) groups. In the second experiment the values of the controls were exceeded at some concentrations both with and without activation. This "effect," however, was due to a low number of spontaneous mutant colonies in the solvent controls (4.3 mutants/1E06 cells) and did not indicate a mutagenic effect of the test article. The absolute number of mutant colonies (range in both experiments of 3.1 to 24.9 mutants/1E06

erythrocytes per animal. To describe a cytotoxic effect, the ratio of polychromatic to normochromatic erythrocytes was assessed by the examination of at least 1000 erythrocytes. Evaluation of Results: Cells were evaluated for large (aneugenic effects) and small (clastogenic effects) micronuclei. The test substance was classified as mutagenic if it induced either a statistically significant (Mann-

Whitney test), dose-related increase in the number of micronucleated

5.7 CARCINOGENICITY

Control and 5 ppm, at interim and terminal sacrifice: brain, heart, liver kidneys, testes, lungs, thoracic and/or mesenteric lymph nodes, salivary glands, pancreas, adrenals, spleen, thymus, aorta, skeletal muscle, small intestine, large intestine, thyroid gland, trachea, spinal cord, peripheral nerve, pituitary gland, epididymides, urinary bladder, accessory sex glands, adipose tissue, ovaries, uterus, nasal turbinates, and any gross lesion suggestive of a pathologic process or with tumor formation.

At 0.5 ppm terminal sacrifice lungs, livers, kidneys, lymph nodes tracheas and grossly visible lesions from all surviving animals; at interim sacrifice grossly visible lesions or tissues where lesions seen at 5 ppm.

Rats dying or culled during the course of the study, complete necropsy and microscopic exam as described above (except when autolysis precluded evaluation) and the presence and absence of neoplasms recorded.

STATISTICAL METHODS:

Hematology, clinical chemistries, body weights, absolute and relative organ weights were analyzed using analysis of variance and Dunnett's Test. Cumulative mortality data were analyzed using Fisher's Exact Probability Test. In both cases, p values of less than 0.05 considered statistically significant.

Gross and microscopic pathology data were analyzed using Fisher's Exact Probability Test (p<0.05) as follows: Gross necropsy: the total collated data from each of the high and low exposure groups were compared with the data of the control group. Each sex was compared separately. Microscopic observations: the incidence of lesions in tissue for each sex from highest exposure group (5 ppm) was compared with the data from controls. At the terminal sacrifice, data from the lower exposure group (0.5 ppm) were analyzed statistically when the number of tissues examined was similar to the controls. The incidence rate for each type of neoplasm was compared separately for each sex between the high exposure group and controls. For the lower exposure (0.5 ppm), statistical evaluation was conducted for neoplasms in those organs upon which microscopic exam was conducted to the degree comparable to the controls and highest exposure group (liver, kidney, lung and lymph nodes and subcutaneous masses/nodules).

To examine the possibility that neoplasms appeared earlier in treated vs. control rats the following parameters were compared for 6 month time periods using Fisher's Exact Probability Test and the Mantel-Haenzel Test with p<0.05:

1) Total number of rats bearing tumors, 2) Number of rats with benign tumors, 3) Number of rats with malignant neoplasms, and 4) Number of rats bearing subcutaneous masses/nodules.

Remark : The statistically significant increase in the incidence of fibrinoid degeneration of the vascular channels in the testes of male rats exposed to 5 ppm was known to be a local vascular manifestation of mesenteric periarteritis syndrome observed as age-related lesion in this rat strain (Sprague-Dawley, Spartan substrain). The laboratory conducting this study commonly observed this lesion in control aging rats of this strain at similar incidence as was observed in this study during it's period of use in the mid to late '70s (the incidence in historical controls in this period ranged from approximately 37 to 85% for seven chronic toxicity/oncogenecity studies).

> Polyarteritis (polyarteritis or periarteritis nodosa) is the most conspicuous inflammatory lesion of the blood vessels of rats. The etiology is unknown and the indidence varies among strains and colonies (Mitsumori, K. (1990) Chapter 29 in Pathology of the Fischer Rat. Eds: Boorman et al., Academic Press, Inc. p 477). A common site in male rats are the arteries of the testicle and to a lesser extent the arteries of the spermatic cord (Burek, J.D. (1978) Pathology of the Aging Rat, CRC Press p. 87). Carlton and Engelhardt (Polyarteritis, In: Cardiovascular and Musculoskeletal Systems Eds: Jones, T.C., Mohr, U. and Hunt, R.D., Springer-Verlag, 1991, p 71) also indicate that this lesion can be present in spermatic arteries.

Mean organ weight data for interim and terminal sacrifices (Tables 3 and 4,

respectively).

GROSS PATHOLOGY: A statistically significant number of both male and female rats exposed to 5 ppm HEA had a distinctive grossly visible yellow staining of the haircoat that persisted into the post-exposure portion of the study. The yellow staining was judged to be a result of the contact of the HEA vapor with the haircoat and was not observed in rats exposed to 0.5 ppm HEA. Chronic murine pneumonia caused by Mycoplasma sp. was observed in all groups as evidenced by pulmonary consolidation and mucopurulent inflammation along the tracheobronchial system. This sometimes included abscess formation, pleuritis, pericarditis, rhinitis and/or tracheitis. An increase in the incidence of numerous gross or microscopically visible lesions occurring as part of or secondary to the chronic murine pneumonia was observed in both male and female rats exposed to 5 ppm HEA.

An increase was observed in the incidence of female rats having a total of 3 grossly-visible subcutaneous masses in the groups exposed to 5 or 0.5 ppm HEA. However, this was not the case with female rats of either exposure group that had 1,2,4, or 5 subcutaneous masses.

HISTOPATHOLOGY: Statistical differences between control and HEA exposed rats in the respiratory tract lesions related to chronic murine pneumonia were observed. Specifically, at 5 ppm, an increase in the incidence and severity of the lesions associated with chronic murine pneumonia was observed.

Lymphoreticular System: Statistical increases in the incidence of edema, inflammation and reactive lymphoid hyperplasia of the thoracic lymph nodes in females at 5 and 0.5 ppm, secondary to chronic murine pneumonia were observed; an increased incidence of edema in mesenteric lymph nodes was also present in females at 0.5 ppm.

Liver: At the terminal sacrifice a statistically significant increase as compared to controls was observed in the focal areas of swollen hepatocytes and focal aggregates of mononuclear cells in males exposed to 5 ppm. A statistically significant increase as compared to controls in the incidence of focal bile duct proliferation in female rats exposed to 5 ppm HEA was also observed.

Female reproductive organs: At the terminal sacrifice only, a statistically significant increase in the incidence of inflammation of the uterus of female rats exposed to 5 ppm was observed (Table 5). No other statistically significant differences for histopathologic observations of the female reproductive organs were found. Specifically, there were no histopathological effects in the ovaries of 2-HEA exposed rats that were considered treatment-related. In addition, at the interim sacrifice, no treatment-related histopathological effects were noted in female reproductive organs of five animals evaluated immediately after 12 months of 2-HEA exposure at 5 ppm.

Male reproductive organs: At the terminal sacrifice only, there was a statistically significant increase in the incidence of fibrinoid degeneration of the vascular channels (local vascular manifestation of mesenteric periarteritis syndrome observed as age-related lesion in this rat strain) in the testes of male rats exposed to 5 ppm (8/14 or 57% in controls; 17/19 or 89% in the 5 ppm group) (Table 6A). The laboratory conducting this study commonly observed this lesion in control aging rats of this strain at similar incidence as was observed in this study during its use in the mid to late '70s (historical control values from seven chronic toxicity/oncogenicity studies ranged from 37 to 85%). Periarteiris of the mesenteric blood vessels was also common in the control and DEA exposed rats (Table 6B).

Table 1

CUMULATIVE PERCENT MORTALITY FOR MALE RATS EXPOSED TO VAPORS OF 2-HYDROXYETHYL ACRYLATE 5 DAYS/WEEK FOR 18 MONTHS FOLLOWED BY A 5 MONTH OBSERVATION PERIOD

 a_{Excludes} those rats used in interim kill (5/sex/dose), and used for cytogenetic examination (4/sex/dose).

statistically different from control data when analyzed
using Fisher's Exact Probability test, p(0.05.

Table 2

CUMULATIVE PERCENT MORTALITY FOR FEMALE RATS EXPOSED TO VAPORS OF 2-HYDROXYETHYL ACRYLATE 5 DAYS/WEEK FOR 18 MONTHS FOLLOWED BY A $6\,$ MONTH OBSERVATION PERIOD

 a_{Excludes} those rats used in interim kill (5/sex/dose), and used for cytogenetic examination (4/sex/dose).

*Statistically different from control data when analyzed
using Fisher's Exact Probability test, p(0.05.

Table 3. BODY WEIGHTS, ORGAN WEIGHTS, AND ORGAN/BODY WEIGHT RATIOS FOR MALE AND FEMALE RATS EXPOSED TO VAPORS OF 2-HYDROXYETHYL ACRYLATE 5 DAYS/WEEK FOR 12 MONTHS Organ Weights (g and g/100 g Body Weight) **Exposure** Sex **Body** level Weight Liver **Testes Brain** Heart **Kidneys** PPM $\frac{2}{100}$ $g/100$ $g/100g$ $g/100g$ $g/100g$ g g g g g g $\frac{g}{0.32}$ g 2.42 15.14 $\overline{0}$ M Mean 621 1.97 1.64 0.26 3.95 0.63 3.87 0.62 37 0.03 0.02 0.07 0.01 2.75 0.32 0.62 0.07 0.76 0.12 $+S.D.$ 565^8 5.0 M 1.93 0.34 1.59 0.28 14.98 2.64 3.77 0.66 4.08 0.72 Mean 0.08 0.01 0.12 0.02 0.79 0.18 0.02 3.17 0.46 0.12 \pm S.D. 24 1.98 0.36 1.55 0.28 12.92 2.35 3.30 4.38 0.80° 0.5 Mean 0.60 M $549²$ 0.06 0.02 0.24 0.03 0.03 0.17 1.32 0.13 0.15 0.06 \pm S.D. 40 1.85 0.59 0.33 7.36 2.32 2.16 $\mathbf{0}$ \overline{F} Mean 317 1.05 0.68 0.05 0.04 0.02 0.02 0.41 \pm S.D. 16 0.18 0.21 0.08 5.0 342 1.80 0.53 0.33 7.87 2.30 2.31 0.68 F Mean 1.13 0.63 26 0.07 0.04 0.07 0.02 0.12 0.27 0.04 \pm S.D. 0.5 $\mathbf F$ Mean 347 1.87 0.54 1.11 0.32 $\frac{9.18}{2.05}$ 2.63 2.17 0.63 0.04 0.02 0.03 0.46 $0.14\,$ \pm S.D. 15 0.13 0.02

^a Statistically significant difference from control mean by analysis of variance and Dunnett's test $p < 0.05$.

Table 4. BODY WEIGHTS, ORGAN WEIGHTS, AND ORGAN/BODY WEIGHTS OF MALE AND FEMALE RATS EXPOSED TO VAPORS OF 2-HYDROXYETHYL ACRYLATE 5 DAYS/WEEK FOR 18 MONTHS FOLLOWED BY A 5 MONTH (males) or 6 MONTH (females) OBSERVATION PERIOD

⁸Statistically significant difference from control mean by analysis of variance and Dunnett's test, p<0.05

+ { $[mean + T value = 2.07]$ $[mean - T value = 1.92]$ }

DATE: 23.08.2005

Table 5

MICROSCOPIC OBSERVATIONS ON FEMALE RATS EXPOSED TO VAPORS OF 2-HYDROXYETHYL ACRYLATE (Terminal Kill After Month 24)

Data listed as number of observations/total number of tissues, organs, or masses examined microscopically/ number of animals examined grossly.

a
Microscopic examination of all major organs limited to control and top dose group. Liver, kidney, lungs,
nasal turbinates, thoracic lymph nodes, and all lesions grossly suggestive of tumor formation were examined
from th

^bStatistically different from control by the Fisher Exact Probability Test, p<0.05.

Table 6A MICROSCOPIC OBSERVATIONS ON MALE RATS EXPOSED TO VAPORS OF 2-HYDROXYETHYL ACRYLATE (Terminal Kill During Month 24)

Data listed as number of observations/total number of tissues, organs, or masses examined microscopically/
number of animals examined grossly.

a
Microscopic examination of all major organs limited to control and top dose group. Liver, kidney, lungs,
nasal turbinates, thoracic lymph nodes, and all lesions grossly suggestive of tumor formation were examined from the low dose group.

 $^{\text{b}}$ Statistically different from control by the Fisher Exact Probability test, p<0.05.

DATE: 23.08.2005

Table 6B

MICROSCOPIC OBSERVATIONS ON MALE RATS EXPOSED TO VAPORS OF 2-HYDROXYETHYL ACRYLATE (Terminal Kill During Month 24)

Data listed as number of observations/total number of tissues, organs, or masses examined microscopically/ number of animals examined grossly.

a
Microscopic examination of all major organs limited to control and top dose group. Liver, kidney, lungs,
nasal turbinates, thoracic lymph nodes, and all lesions grossly suggestive of tumor formation were examined from the low dose group.

Table 7

MICROSCOPIC ORSERVATIONS ON FEMALE RATS EXPOSED TO VAPORS OF 2-HYDROXYETHYL ACRYLATE

(Terminal Kill After Month 24)

Data listed as number of observations/total number of tissues, organs, or masses examined microscopically/ ber of animals examined grossly.

"Microscopic examination of all major organs limited to control and top dose group. Liver, kidney, lungs, nasal turbinates, thoracic lymph nodes, and all lesions grossly suggestive of tumor formation were examined
from the

bStatistically different from control by the Fisher Exact Probability Test, p(0.05.

Conclusion : The results of this study indicate that chronic inhalation of 2-HEA by rats produced only a minimal degree of toxicity at 5 ppm (haircoat staining and increased incidence and severity of chronic murine pneumonia). Female rats in the 5 ppm group at the terminal sacrifice showed an increased incidence of uterine inflammation as compared to the control animals. However, no other statistically significant differences for histopathological observations of the female reproductive organs were found, including the ovaries. An evaluation of the histopathological data from the male animals exposed to 5 ppm indicated an increased incidence of fibrinoid degeneration in the vascular channels of the testes which was a local

5.8.1 TOXICITY TO FERTILITY

5.8.2 DEVELOPMENTAL TOXICITY/TERATOGENICITY

OECD SIDS	HYDROXYPROPYL ACRYLATE	
5. TOXICITY	ID: 25584-83-2 DATE: 23.08.2005	
NOAEL Fetotoxicity Result Method Year GLP Test substance	$> 10 - ppm$ not selectively toxic to the embryo or fetus other: consistent with OPPTS 870.3700 with minor exceptions 1999 no data other TS	
Remark	The developmental toxicity of seven acrylates were studied in Sprague- Dawley rats after inhalation exposure for 6 h/day, during 6 to 20 days of gestation. The exposure concentrations were 0, 1, 5 and 10 for	
Result	hydroxypropyl acrylate. Dose groups consisted of 20-22 pregnant females. MATERNAL TOXIC EFFECTS BY DOSE LEVEL: -Mortality and day of death: none -Body weight/body weight gain: Maternal weight gain was significantly less than controls during the first half of the exposure at 10 ppm and absolute weight gain was reduced at 5 and 10 ppm.	
	$Exp.$ # of BW BW gain (g) on GD Absolute Conc. Dams GD 6 $6 - 13$ 13-21 6-21 wt gain (g) 21 $27 + 13$ 0 266±16 28±6 99±19 126±22 1 20 272 ± 15 26±6 99±19 125±20 27 ± 9 271±13 25±7 22 $118 + 33$ 5 $93{\pm}30$ $19±11*$ 10 21 272 ± 15 22±8* $92 + 21$ 114±26 16±12**	
	*,** Significant differences from control (0 ppm) value, p<0.05, and p<0.01, respectively.	
	-Food consumption: Food consumption was slightly reduced during treatment at 10 ppm.	
	# of Food consumption (g/dam/day) on GD Exp. $0-6$ Conc. Dams $6 - 13$ 13-21 $6 - 21$ 21 ± 2 22 ± 2 26 ± 2 24 ± 2 21 0 21 ± 2 26 ± 2 24 ± 2 1 20 23 ± 2 5 22 22 ± 1 21 ± 2 25 ± 3 23 ± 2 21 10 22 ± 2 $20 \pm 2^*$ $24 \pm 2^*$ 22 $\pm 2^*$	
	* Significant differences from control (0 ppm) value, p<0.05, and p<0.01, respectively.	
	-Implantations and resorptions: There were no significant differences in the numbers of implantation sites and live fetuses, in the incidence of non-live implants and resorptions, or in the fetal sex ratio.	
	Litters with implants # of $%$ of $%$ of non-live implant resorp.	
	Exp. $#$ of # of sites/ implants/ sites/ Conc. females litter litters litter@ litter 6.10 ± 8.37 24 21 14.48±3.06 6.10 ± 8.37 0 23 1 20 14.40±2.23 9.76 ± 15.58 9.76 ± 15.58 24 22 14.27±4.09 5 10.83±21.79 10.83±21.79 23 21 10 14.52±3.74 6.70 ± 5.90 6.70 ± 5.90	
	@ Resorptions plus dead fetuses.	
	FETAL DATA: -Fetal body weights: There were no effects on fetal body weights at any exposure level.	
	Litters with Live Fetuses # of live	

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-Fetal malformations: Single occurrences of visceral malformations were seen in the control and in the 5 ppm-exposure groups. There were no external variations observed in any group. The incidences of visceral and skeletal variations were scattered, with no indication of adverse effects in any of the exposed groups when compared to controls.

5. TOXICITY

OECD SIDS HYDROXYPROPYL ACRYLATE 5. TOXICITY ID: 25584-83-2 ID: 25584-83-2 DATE: 23.08.2005

5.8.3 TOXICITY TO REPRODUCTION, OTHER STUDIES

5.10 EXPOSURE EXPERIENCE

radioactivity were monitored over the exposure period.

Rampy L.W. et al. (1978) Toxicol. Appl. Pharmacol., 45: 310.

