FOREWORD

INTRODUCTION

DIPROPYLENE GLYCOL (MIXED ISOMERS AND DOMINANT ISOMER) CAS N°: 25265-71-8 & 110-98-5)
SIDS Initial Assessment Report
for
11th SIAM
(USA, January 23-26, 2001)

Chemical Name: Dipropylene glycol, mixed isomers and dominant isomer

CAS No: 25265-71-8 and 110-98-5

Sponsor Country: USA

National SIDS Contact Point in Sponsor Country: Oscar Hernandez
US EPA
401 M St., SW
Washington, DC 20460

HISTORY: At SIAM 11 the recommendation of Low Priority for Further Work was agreed. Revisions to the SIAR and preparation of Robust Summaries were requested and agreed.

COMMENTS:

Deadline for circulation

Date of Circulation:
SIDDS INITIAL ASSESSMENT PROFILE

<table>
<thead>
<tr>
<th>CAS No.</th>
<th>25265-71-8 and 110-98-5</th>
</tr>
</thead>
<tbody>
<tr>
<td>Chemical Name</td>
<td>Dipropylene glycol, mixed isomers and dominant isomer</td>
</tr>
<tr>
<td>Structural Formula</td>
<td>CH₃-CHOH-CH₂O-CH₂-CHOH-CH₃</td>
</tr>
</tbody>
</table>

RECOMMENDATIONS

The chemical is currently of low priority for further work.

SUMMARY CONCLUSIONS OF THE SIAR

Human Health

Dipropylene glycol (DPG) is not acutely toxic by oral (LD₅₀ > 13 g/kg bw/day from 7 rat studies and 17.6 g/kg bw/day from a guinea pig study), dermal (LD₅₀ > 5 g/kg bw/day in 2 rabbit studies) or inhalation (no deaths observed in rats and guinea pigs at 6 to 8 g/m³) routes of exposure. DPG is slightly irritating to the skin and eyes of rabbits. Based on human data, DPG is not a skin sensitizer. Repeated exposures of rats to DPG did not result in adverse effects at levels up to 5% (estimated NOAEL is about 6.2 g/kg bw/day) in drinking water. At about 12.5 g/kg bw/day (10%), kidney lesions appeared in about 30% of the rats. Results from an OECD 422 combined repeat dose/reproductive/developmental toxicity test on the structural analogue, tripropylene glycol (TPG), demonstrated a NOAEL of 200 mg/kg bw and a LOAEL of 1000 mg/kg bw for repeated dose toxicity, with increased relative weight for liver and kidney. Metabolic fate data on TPG demonstrates that TPG is readily converted to DPG, PG, and CO₂ in rats. Thus, data from TPG are relevant to DPG. DPG did not cause fetal toxicity or teratogenicity in rats (NOAEL = 5 g/kg bw/day) or rabbits (NOAEL = 1.2 g/kg bw/day). No reproductive studies have been conducted on DPG. However, the structural analogues, propylene glycol and TPG, have been tested for reproductive effects and shown to have NOAELs of 10.1 g/kg bw in mice and 1 g/kg bw in rats, respectively. Thus, the lack of reproductive effects from TPG and the high NOAEL for PG reproductive toxicity indicate that no reproductive effects are expected in animals exposed to DPG, in the absence of maternal toxicity. DPG is not a genetic toxicant based on in vitro (bacterial and mammalian cells in culture) and in vivo (micronucleus) studies.

Environment

Dipropylene glycol (DPG) is not volatile, but is miscible with water. Air monitoring data are not available, but concentrations of dipropylene glycol in the atmosphere are expected to be extremely low because of its low vapor pressure and high water solubility. Low levels of DPG (0.4 mg/l) in drinking water were reported in one study. It is biodegraded in water and expected to be biodegraded in soil, as indicated by >70% degradation after 28d in a Zahn-Wellens test. It is not expected to bioaccumulate, with measured BCFs between 0.3 and 4.6 in fish. Measured aquatic toxicity data on fish and amphibians report toxicity at >5,000 and 3,181 mg/L, respectively. Based on QSAR data for Daphnia and algal toxicity, and the measured data for fish and amphibians, DPG is not expected to be toxic to aquatic organisms except at very high concentrations. Using an assessment factor of 100 and the fish 96-hour LC₅₀, the PNEC is >50 mg/l; if the amphibian data are used, the PNEC is 32 mg/l.
**Exposure**

Dipropylene glycol is produced as a byproduct of the manufacture of propylene glycol. The US production capacity of DPG was 131 million pounds (59.5 kilotonnes) in 1998; the demand was 108 million pounds (49 kilotonnes). DPG is used (percent of demand) as follows: plasticizers, 38 percent; unsaturated polyester resins, 23 percent; cosmetics and fragrances, 10 percent; polyurethane polyols, 8 percent; alkyd resins, 7 percent; miscellaneous, including solvents and functional fluids (specialty de-icers, inks, lubricants), 14 percent.

**NATURE OF FURTHER WORK RECOMMENDED**

No further work is recommended.
## FULL SIDS SUMMARY

### PHYSICAL-CHEMICAL

<table>
<thead>
<tr>
<th>CAS NO: 25265-71-8, 110-98-5</th>
<th>SPECIES</th>
<th>PROTOCOL</th>
<th>RESULTS</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td><strong>2.1 Melting Point</strong></td>
<td></td>
<td>NG</td>
<td>&lt; -39°C</td>
</tr>
<tr>
<td><strong>2.2 Boiling Point</strong></td>
<td></td>
<td>NG</td>
<td>228-236°C</td>
</tr>
<tr>
<td><strong>2.3 Density</strong></td>
<td></td>
<td>NG</td>
<td>1.022 g/cm³ at 25°C</td>
</tr>
<tr>
<td><strong>2.4 Vapour Pressure</strong></td>
<td></td>
<td>NG</td>
<td>0.0128 hPa at 20°C</td>
</tr>
<tr>
<td><strong>2.5 Partition Coefficient (Log K&lt;sub&gt;ow&lt;/sub&gt;)</strong></td>
<td></td>
<td>NG</td>
<td>-1.486</td>
</tr>
<tr>
<td><strong>2.6 Water Solubility</strong></td>
<td></td>
<td>NG</td>
<td>Miscible at 25°C</td>
</tr>
<tr>
<td><strong>A. PH</strong></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td><strong>B. Pka</strong></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td><strong>2.12 Oxidation: Reduction Potential</strong></td>
<td></td>
<td></td>
<td></td>
</tr>
</tbody>
</table>

### ENVIRONMENTAL FATE AND PATHWAY

| **3.1.1 Photodegradation**    |         |          |         |
| **3.1.2 Stability in Water**  |         | NG       | No hydrolyzable groups |
| **3.2 Monitoring Data**       |         |          |         |
| **3.3 Transport and Distribution** | Fugacity model Level III | | 0.11% to air; 46.1% to water; 53.7% to soil; 0.08% to sediments. |
| **3.5 Biodegradation**        | a. Other TG: EEC-87/302 b. OECD TG 301C | | a. >70% after 28 days (pre-adapted) b. No degradation. |
| **3.6 BOD5, COD**             | Other TG: APHA 219 | | BOD5 = 92 mgO2/L COD = 1840 mg/g |
| **3.7 Bioaccumulation**       | Cyprinus carpio OECD TG 305C | | BCF of 0.3-1.4 (42 days at 3 mg/L) BCF of <2.2-4.6 (42 days at 3 mg/L) |

### ECOTOXICOLOGY

<p>| <strong>4.1 Acute/Prolonged Toxicity to Fish</strong> | Carassius auratus 24-hr lethality Other TG: APHA 1971 | | LC50 = &gt;5000 mg/L (static) |
| <strong>4.2 Acute Toxicity to Aquatic Invertebrates</strong> (Daphnia) | | | No data |
| <strong>4.3 Toxicity to Aquatic Plants e.g. Algae</strong> | | | No data |</p>
<table>
<thead>
<tr>
<th>CAS NO: 25265-71-8, 110-98-5</th>
<th>SPECIES</th>
<th>PROTOCOL</th>
<th>RESULTS</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>Toxicity to</td>
<td></td>
<td></td>
</tr>
<tr>
<td>4.4</td>
<td>Microorganisms</td>
<td></td>
<td></td>
</tr>
<tr>
<td></td>
<td>Pseudomonas putida</td>
<td>Other TG:</td>
<td>EC10 = 1000 mg/L (static, 18 hr)</td>
</tr>
<tr>
<td></td>
<td></td>
<td>LTwS-Nr 10</td>
<td></td>
</tr>
<tr>
<td>4.5.1</td>
<td>Chronic Toxicity</td>
<td></td>
<td></td>
</tr>
<tr>
<td></td>
<td>to Fish</td>
<td>--</td>
<td></td>
</tr>
<tr>
<td></td>
<td>--</td>
<td>--</td>
<td>No Data</td>
</tr>
<tr>
<td>4.5.2</td>
<td>Chronic Toxicity</td>
<td></td>
<td></td>
</tr>
<tr>
<td></td>
<td>to Aquatic</td>
<td>--</td>
<td></td>
</tr>
<tr>
<td></td>
<td>Invertebrates</td>
<td>--</td>
<td></td>
</tr>
<tr>
<td></td>
<td>(Daphnia)</td>
<td>--</td>
<td></td>
</tr>
<tr>
<td></td>
<td>--</td>
<td>--</td>
<td>No Data</td>
</tr>
<tr>
<td>4.6.1</td>
<td>Toxicity to Soil</td>
<td></td>
<td></td>
</tr>
<tr>
<td></td>
<td>Dwelling</td>
<td>--</td>
<td></td>
</tr>
<tr>
<td></td>
<td>Organisms</td>
<td>--</td>
<td></td>
</tr>
<tr>
<td></td>
<td>--</td>
<td>--</td>
<td>No Data</td>
</tr>
<tr>
<td>4.6.2</td>
<td>Toxicity to</td>
<td></td>
<td></td>
</tr>
<tr>
<td></td>
<td>Terrestrial</td>
<td>--</td>
<td></td>
</tr>
<tr>
<td></td>
<td>Plants</td>
<td>--</td>
<td></td>
</tr>
<tr>
<td></td>
<td>--</td>
<td>--</td>
<td>No Data</td>
</tr>
<tr>
<td>4.6.3</td>
<td>Toxicity to Other</td>
<td></td>
<td></td>
</tr>
<tr>
<td></td>
<td>Non-Mammalian</td>
<td></td>
<td></td>
</tr>
<tr>
<td></td>
<td>Terrestrial</td>
<td></td>
<td></td>
</tr>
<tr>
<td></td>
<td>Species (Including</td>
<td></td>
<td></td>
</tr>
<tr>
<td></td>
<td>Birds)</td>
<td></td>
<td></td>
</tr>
<tr>
<td></td>
<td>--</td>
<td>--</td>
<td>No Data</td>
</tr>
<tr>
<td>4.9</td>
<td>Additional</td>
<td></td>
<td></td>
</tr>
<tr>
<td></td>
<td>Remarks</td>
<td>Xenopus laevis</td>
<td>LC50 = 3,181 mg/L (48 hr, static)</td>
</tr>
</tbody>
</table>

### TOXICOLOGY

| 5.1.1 | Acute Oral Toxicity | a. Rat | Acute lethality NG | a. LD50 = 16,000 mg/kg (est. from 15.8 mL/kg) |
|       |                    | b. Mouse |                   | b. LD50 = >2,000 mg/kg                        |
| 5.1.2 | Acute Inhalation   | Rat, guinea pig | Acute lethality NG | Not lethal: 6000-8000 mg/m^3 (aerosol, nominal concentration) |
|       | Toxicity           | |                   |                                             |
| 5.1.3 | Acute Dermal Toxicity | Rabbit | Acute lethality NG | LD50 > 5000 mg/kg                           |

| 5.2.1 | Skin Irritation   | a. Rabbit | Draize test NG | a. Slightly irritating (500 mg, 24 hr, occluded) |
|       |                   | b. Human | NG | b. None to mild irritation (2 mL of 25% solution; 24 hr; semi-occluded) |
| 5.2.2 | Eye Irritation    | Rabbit | Draize test NG | Slightly irritating (0.1 mL of 100% solution; 4 hr) |

| 5.3 | Skin Sensitization | Human | NG | 1 of 503 was sensitized to synthesis grade |

| 5.4 | Repeated Dose Toxicity | Rat | 9-77 days, in drinking water. | NOAEL 5% |
|     |                      | Dog | 4 or 6 gavage doses | NOAEL 20 mL/kg |
|     |                      | NG |                          |                  |
|     |                      | Rat | 15 wk, oral | NOAEL = 12% in diet |

| 5.5 | Genetic Toxicity In Vitro | | | |

6 UNEP PUBLICATIONS
### CAS NO: 25265-71-8, 110-98-5

<table>
<thead>
<tr>
<th>SPECIES</th>
<th>PROTOCOL</th>
<th>RESULTS</th>
</tr>
</thead>
<tbody>
<tr>
<td><strong>A.</strong> Bacterial Test (Gene mutation)</td>
<td>Salmonella typhimurium</td>
<td>Mutagenicity; Ames test NG</td>
</tr>
<tr>
<td><strong>B.</strong> Non-Bacterial In Vitro Test</td>
<td>Mouse</td>
<td>Lymphoma assay NG</td>
</tr>
<tr>
<td><strong>5.6</strong> Genetic Toxicity In Vivo</td>
<td>Mice</td>
<td>Micronucleus assay OECD TG 474 EPA 870.5395</td>
</tr>
<tr>
<td><strong>5.7</strong> Carcinogenicity</td>
<td>--</td>
<td>--</td>
</tr>
<tr>
<td><strong>5.8</strong> Toxicity to Reproduction</td>
<td>--</td>
<td>--</td>
</tr>
<tr>
<td><strong>5.9</strong> Developmental Toxicity/Teratogenicity</td>
<td>Rat</td>
<td>Gavage Other TG: NTP</td>
</tr>
<tr>
<td></td>
<td>Rabbit</td>
<td>Gavage Other TG: NTP</td>
</tr>
<tr>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td><strong>5.10</strong> Other Relevant Information</td>
<td>Rat</td>
<td>104 wk, oral NG</td>
</tr>
<tr>
<td></td>
<td>Mouse</td>
<td>Drinking water; continuous from 7 days prior to mating NG</td>
</tr>
<tr>
<td></td>
<td>Rat</td>
<td>Gavage; premating through Day 3 of lactation OECD TG 422</td>
</tr>
<tr>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td><strong>5.11</strong> Experience with Human Exposure</td>
<td>ss</td>
<td></td>
</tr>
</tbody>
</table>
SIDIS Initial Assessment Report

1. Identity

Commercial dipropylene glycol (CAS # 25265-71-8; (CH₃-CHOH-CH₂O-CH₂-CHOH-CH₃) is composed of 3 isomers (2,2'-dihydroxydiisopropylether (CAS-No.: 108-61-2); 2,2'-dihydroxydipropylether (CAS-No.: 110-98-5); 2-hydroxypropyl-2'-hydroxyisopropylether (CAS-No.: 106-62-7) and is typically 98% pure. It is a liquid that possesses the following physico-chemical properties and characteristics:

<table>
<thead>
<tr>
<th>Property</th>
<th>Value</th>
</tr>
</thead>
<tbody>
<tr>
<td><strong>Chemical Formula</strong></td>
<td>(CH₃CHOCH₂)₂O</td>
</tr>
<tr>
<td>Molecular Weight</td>
<td>134.2</td>
</tr>
<tr>
<td>Purity</td>
<td>&gt;= 98% w/w</td>
</tr>
<tr>
<td>Impurities</td>
<td></td>
</tr>
<tr>
<td>Solubility</td>
<td>Miscible with water</td>
</tr>
<tr>
<td>Melting Point</td>
<td>&lt; -39 °C</td>
</tr>
<tr>
<td>Boiling Point</td>
<td>222.2 – 236 °C</td>
</tr>
<tr>
<td>Density</td>
<td>1022-1025 mg/ml at 20 °C</td>
</tr>
<tr>
<td>Vapor Pressure</td>
<td>0.01 hPa at 20°C</td>
</tr>
<tr>
<td></td>
<td>0.05 hPa at 21°C</td>
</tr>
<tr>
<td>Log Kow</td>
<td>-1.486</td>
</tr>
<tr>
<td>Synonyms</td>
<td>DPG; Di-1,2-propyleneglycol; oxybispropanol; Practical</td>
</tr>
</tbody>
</table>
2. General Information on Exposure

2.1 Production

Dipropylene glycol is produced as a byproduct of the manufacture of propylene glycol. The United States (US) production capacity of dipropylene glycol (i.e., byproduct production capacities of propylene glycol plants) was 131 million pounds (60 thousand tonnes) in 1998. The US Domestic demand in 1998 was 108 million pounds (49 thousand tonnes). In 1998, dipropylene glycol was produced in the US by The Dow Chemical Company, Eastman Chemical Company, Huntsman Corporation, Lyondell Chemical Company, and Olin Corporation. (ChemExpo Chemical Profile, 1998). Worldwide capacity is estimated at 235 million pounds (107 thousand tonnes).

2.2 Use

Dipropylene glycol is used primarily as an industrial intermediate, but is also used as a substance in consumer products, and as an ingredient in pesticidal formulations. Uses of dipropylene glycol as a substance capitalize on its superior performance as a plasticizer as well as properties (e.g., high solvency, high viscosity) that permit dipropylene glycol to act as a functional ingredient of fluids. The uses of dipropylene glycol are listed in the following table.

<table>
<thead>
<tr>
<th>USES</th>
<th>APPLICATION</th>
<th>FUNCTION</th>
<th>% PRODUCTION</th>
</tr>
</thead>
<tbody>
<tr>
<td>Intermediate</td>
<td>Specialty plasticizer</td>
<td>Plasticizer</td>
<td>38 %</td>
</tr>
<tr>
<td>Intermediate</td>
<td>Non-reinforced polyesters</td>
<td>Resin monomer</td>
<td>23 %</td>
</tr>
<tr>
<td>Substance</td>
<td>Cosmetics and Fragrances</td>
<td>Humectant, Emollient</td>
<td>10 %</td>
</tr>
<tr>
<td>Intermediate</td>
<td>Polyurethane polyols</td>
<td>Plasticizer</td>
<td>10 %</td>
</tr>
<tr>
<td>Intermediate</td>
<td>Alkyd resins</td>
<td>Resin monomer</td>
<td>7 %</td>
</tr>
<tr>
<td>Substance, Intermediate</td>
<td>Miscellaneous (e.g., dyes &amp; inks, paints &amp; coatings; functional uses in hydraulic brake fluids; cutting oils)</td>
<td>Solvent, functional fluid, coolant</td>
<td>14%</td>
</tr>
</tbody>
</table>

(Source: ChemExpo Chemical Profile (1998)).

The uses listed in the table above have been grouped and assigned a publicly available percentage of production. The Environmental Defense (ED) Scorecard website (Source: www.scorecard.org), for dipropylene glycol lists 6 consumer product types and 18 pesticidal products containing dipropylene glycol. Data available from the US EPA indicates that in 2001 there are two pesticidal products containing dipropylene glycol as an ingredient registered in the US, whereas 20 such products have been registered over time (Source: California Department of Pesticide Registration website, http://www.cdpr.ca.gov/cgi-bin/epa/chemdet.pl?pccode=068604).
2.3 Exposure

As most dipropylene glycol in the US is used in industrial applications, occupational exposures via the dermal or inhalation route present a potential for exposure. In the commercial service and consumer settings use as a functional fluid (e.g. in hydraulic brake fluids, cutting oils) or as an ingredient in pesticides present a potential for inhalation exposure in addition to dermal exposure.

In the consumer setting, dermal exposure, and to a lesser degree inhalation exposure are to be expected where dipropylene glycol is formulated into cosmetic or fragrance products.

**DIPROPYLENE GLYCOL EXPOSURE POTENTIAL**

<table>
<thead>
<tr>
<th>POPULATION</th>
<th>ROUTE(S) OF EXPOSURE</th>
<th>SOURCE(S)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Occupational Exposure</td>
<td>Dermal</td>
<td>Manufacturing</td>
</tr>
<tr>
<td></td>
<td>Inhalation</td>
<td>Industrial use as Intermediate</td>
</tr>
<tr>
<td></td>
<td></td>
<td>Commercial Service</td>
</tr>
<tr>
<td>Consumer</td>
<td>Dermal</td>
<td>Cosmetics, fragrances, germicidal products</td>
</tr>
<tr>
<td></td>
<td>Inhalation</td>
<td></td>
</tr>
</tbody>
</table>

The use concentration of dipropylene glycol in branded pesticidal products is also available. This is summarized in the following table.

**PESTICIDAL FORMULATIONS CONTAINING DIPROPYLENE GLYCOL**

<table>
<thead>
<tr>
<th>RANGE OF CONCENTRATION (%)</th>
<th>&lt;1</th>
<th>1-5</th>
<th>&gt;5</th>
</tr>
</thead>
<tbody>
<tr>
<td>No. of Brand Name Products</td>
<td>1</td>
<td>16</td>
<td>1</td>
</tr>
</tbody>
</table>

(Source: EDF Website: [www.scorecard.org](http://www.scorecard.org))

Only two of the 18 branded products shown in the above table currently (May, 2001) maintain active registrations in the US, and both contain dipropylene glycol in the 1 to 5 percent range.

Exposure of dipropylene glycol from environmental sources is expected to be low. Very low levels of dipropylene glycol have been reported in drinking water (0.2 and 0.4 ng/l) and 5 paper mill waste water treatment plants effluents (11 µg/l). Because of its low volatility detection of dipropylene glycol in air samples is not expected.
3. Environment

3.1 Environmental Exposure

Based on the available data, dipropylene glycol is expected to present a low hazard to the environment. If released into the environment, it will preferentially partition into water. Because of its low soil sorption coefficient ($K_{oc}$), dipropylene glycol is expected to mobilize if spilled on soil and will not adsorb to particles. Volatility is not expected to be a significant fate process for this material due to its low vapor pressure and high water solubility. Photodegradation of the material in air or water is also not expected to be a significant fate process. Once in the environment, dipropylene glycol is capable of being degraded under aerobic conditions by bacteria present in soil and water. As is typical for ethers and glycols, dipropylene glycol is hydrolytically stable. Results of the octanol/water partition coefficient (expressed as log $K_{ow}$) and from bioaccumulation studies with carp (MITI, 1992) indicate that dipropylene glycol is not expected to significantly accumulate in aquatic organisms (BCF 0.3 to 4.6).

3.2 Environmental Partitioning

Based on the EPIWIN Level III Fugacity Model (Mackay, 1991), dipropylene glycol is expected to partition primarily to water and soil.

<table>
<thead>
<tr>
<th>Compartment</th>
<th>Percent</th>
</tr>
</thead>
<tbody>
<tr>
<td>Air</td>
<td>0.107</td>
</tr>
<tr>
<td>Water</td>
<td>46.1</td>
</tr>
<tr>
<td>Soil</td>
<td>53.7</td>
</tr>
<tr>
<td>Sediment</td>
<td>0.0768</td>
</tr>
</tbody>
</table>

The results predicted by the model are consistent with study data generated on dipropylene glycol. Based on the low soil sorption coefficient ($K_{oc}$ 3 to 56.2), dipropylene glycol would be expected to mobilize following exposure to soil and would not bind to sediment or soil particles. Mathematical modeling of photodegradation gives an estimated atmospheric half-life of 13 hours, supporting the model results suggesting that the air compartment is not a significant fate process following release of dipropylene glycol. Hydrolysis of dipropylene glycol in water and soil is also not expected to be an important fate process.

Results from biodegradation studies show that dipropylene glycol is biodegraded in water and expected to be biodegraded in soil. Degradation (>70% in 28 days) was observed when the Zahn-Wellens test was used (BASF, 1989), but not when the same compound was evaluated in the MITI test (MITI, 1992). Differences in the technique used to add dipropylene glycol to the test flasks was an important variable in determining the potential degradation of the test substance under the conditions used in these tests. Additional biodegradation tests with *Corynebacterium* sp. under vigorous shaking conditions and with soil bacteria confirmed that dipropylene glycol is biodegradable when contact between the test substance and bacteria is enhanced (Kawai *et al.*, 1977, 1985). Evaluation of the BOD5/COD plants (Bridie *et al.*, 1979b) and the acute toxicity of dipropylene glycol to *Pseudomonas putida* (Claussen, 1993) demonstrate that dipropylene glycol is not expected to be toxic or to cause a significant oxidative load to wastewater treatment.
### 3.3 Effects on the Environment

Based on a combination of test data, quantitative structure activity relationship analysis (QSAR), and data on analogs (propylene glycol, tripropylene glycol), dipropylene glycol presents a low hazard concern for the environment. Representative results for dipropylene glycol include:

<table>
<thead>
<tr>
<th>Organism</th>
<th>Duration (hrs.)</th>
<th>LC$<em>{50}$/EC$</em>{50}$ (mg/l)</th>
<th>Source</th>
</tr>
</thead>
<tbody>
<tr>
<td><strong>Fish</strong></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td><em>Carassium auratus</em></td>
<td>24</td>
<td>&gt;5,000</td>
<td>Bridie <em>et al.</em>, 1979a</td>
</tr>
<tr>
<td><strong>Amphibians</strong></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td><em>Rana brevipoda porosa</em></td>
<td>48</td>
<td>5,300</td>
<td>Nishiuchi, 1984</td>
</tr>
<tr>
<td><em>Xenopus laevis</em></td>
<td>48</td>
<td>3,181</td>
<td>DeZwart and Slooff, 1987</td>
</tr>
<tr>
<td><strong>Bacteria</strong></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td><em>Pseudomonas putida</em></td>
<td>18</td>
<td>1,000 (EC$_{10}$)</td>
<td>Claussen, 1993</td>
</tr>
</tbody>
</table>

Toxicity predictions using QSAR support the general conclusions of low toxicity. QSAR predictions for dipropylene glycol based on SARs (chemical class = alcohol ether) for neutral organic chemicals include:

- Fish 96-hour LC$_{50}$ >10,000 mg/l
- Daphnid 48-hour LC$_{50}$ >10,000 mg/l
- Green algal 96-hour LC$_{50}$ >10,000 mg/l
- Fish chronic value (ChV) >1,000 mg/l
- Daphnid ChV 470 mg/l
- Algal ChV 340 mg/l

Toxicity values for the analogs propylene glycol and tripropylene glycol also support the expected lack of aquatic toxicity, with fish LC$_{50}$ values of >46,000 mg/l and >1,000 mg/l, invertebrate EC$_{50}$ values of 10,000 mg/l and >1,000 mg/l and aquatic plant EC$_{50}$ values of 19,000 mg/l and >1,000 mg/l for propylene glycol and tripropylene glycol, respectively. Although chronic study data for dipropylene glycol are not available, QSAR predictions suggest that this test substance will not cause chronic toxicity to aquatic organisms. Published studies are not available on the effects of dipropylene glycol on terrestrial organisms or plants, however, little effect is expected, since dipropylene glycol degrades in soil, has very low bioaccumulation potential, and is of low toxicity to aquatic organisms.

Based on the QSAR data for Daphnia and algal toxicity, and the measured data for fish and amphibians, dipropylene glycol is not expected to be toxic to aquatic organisms except at very high concentrations. Using an assessment factor of 100 and the goldfish (*Carassius auratus*) data (LC$_{50}$ >5,000 mg/l), a predicted no effect concentration (PNEC) of >50 mg/l is obtained. Using the clawed toad (*Xenopus laevis*) data (LC$_{50}$ 3,181 mg/l), and an assessment factor of 100, the PNEC is 32 mg/l.
4. Human Health

4.1 Toxicity Studies

4.1.1. Acute Toxicity

Dipropylene glycol is not acutely toxic by oral, dermal, or inhalation exposure. Acute oral toxicity has been examined in the rat, mouse, and guinea pig and the reported LD₅₀s were 15.8 ml/kg (16000 mg/kg), >2000 mg/kg and 17600 mg/kg, respectively. The study that provides the best documentation of design and results was reported by Spanjers and Til, 1980. The authors dosed 5 groups of animals with graded amounts of dipropylene glycol. There was no indication of clinical signs in the report, but the necropsy examination after death or at the end of the 14-day observation period was reported to have found no gross alterations of the internal organs. The authors determined an LD₅₀ for the study of 15.8 ml/kg (approximately 16000 mg/kg). In a study designed to assess micronuclei, 6 male mice received two consecutive daily doses of dipropylene glycol via oral gavage; survivors were terminated 24 hours later. There were no deaths in the study; therefore, the LD₅₀ for this study was >2000 mg/kg (Dow, 1999).

Dipropylene glycol vapor and aerosol has been examined for acute inhalation toxicity. An aerosol atmosphere of 6000 to 8000 mg/m³ dipropylene glycol was not lethal to rats or guinea pigs (Oettel and Hofmann, 1961), but vaporized degradation products produced by heating dipropylene glycol to 170 °C was lethal to 5 of 6 rats exposed for 8 hours. No mortality occurred from vapors generated at 120°C. Pathologic abnormalities were not observed in any of the animals (Oettel and Hofmann, 1961).

Dipropylene glycol did not produce deaths when administered to the skin of animals. Rabbits that were reportedly administered 5000 and 20000 mg/kg dipropylene glycol to their skin did not die from the treatment (BIBRA, 1991; Opdyke, 1978; Deichman and Gerarde, 1969). Details of these studies are not available.

Based on the above acute data, dipropylene glycol is practically non-toxic by the oral, inhalation and dermal routes.

4.1.2. Irritation and Sensitization

Dipropylene glycol is slightly irritating to the skin and eyes. Dipropylene glycol was described as slightly irritating to rabbit skin in a report that did not provide details on test conditions or results (BIBRA, 1991). Similar minimal skin effects were seen in a human volunteer study where 0.2 ml of 25% dipropylene glycol in water was applied semi-occlusively to 33 subjects for 24 hours. Nine subjects had mild erythema at either 30 minutes or 24 hours; two had mild erythema at both 30 minutes and 24 hours. Twenty-two subjects had no reaction after 30 minutes or 24 hours (Acklin and Plaza, 1995).

Dipropylene glycol was reported to be mildly irritating when it was placed full strength in rabbit eyes, but only transient eye irritation was reported from a formulation containing 7.2% dipropylene glycol (BIBRA, 1991). The details of the study were not provided.

Dipropylene glycol appears to have low potential to produce allergic skin reactions. There are no experimental animal studies reported, but there is a human clinical study. A
dermatology clinic tested 503 (212 men, 291 women) consecutive patients with eczema for sensitivity to dipropylene glycol, using 1 to 10% dipropylene glycol applied for 2 days. This was not a standard sensitization test because there was no specific induction phase and the subjects at the onset of the study were not clinically normal. One individual was found to be sensitized; 22 had questionable erythema; and 480 were unreactive (Johansen et al., 1995).

4.1.3. Repeated Dose Toxicity

The available information indicates that dipropylene glycol has a low order of repeated exposure toxicity. Dipropylene glycol repeatedly administered at very high doses produced lethality, kidney damage, and apparent neurobehavioral changes in laboratory animals. Dipropylene glycol was administered at 10% in the drinking water of twenty five rats for 9 to 68 days. Seven animals died between days 10 and 30; five had kidney lesions described as hydropic degeneration of renal epithelium. Four of the 18 animals that survived to termination also had kidney lesions. The severity of kidney lesions was not reported. No effects were seen in seven rats given dipropylene glycol in the drinking water at 1 to 5% for 33 to 77 days (Kesten et al., 1939). Interpretation of this study is difficult due to the very high doses administered (10%). Under current guidelines, materials are not normally administered at greater than 5% in diet or drinking water because of concerns about interference with normal nutrition. Thus in this study, it can not be discerned whether these effects were directly related to dipropylene glycol or to nutritional imbalance; however, the NOAEL was established to be >5% in the drinking water.

In another study, investigators administered dipropylene glycol at 12% of the diet to rats for 15 weeks. The purpose of the study was to compare effects of dipropylene glycol with propylene glycol on running behavior. Administration of dipropylene glycol at 12% in the diet resulted in decreased running activity (van Winkle and Kennedy, 1940). The magnitude and temporal nature (relative to daily dosing and over the course of the study) to the changes in running behavior were not reported. The lack of details on study design, assessment method, and results limit the interpretation of this study.

There is a study of dogs that received multiple doses of dipropylene glycol over the apparent course of a couple of days. Dipropylene glycol was administered to dogs by gavage as follows: one dog received 6 doses of 1.5 cc/kg/dose (9 cc/kg); two dogs received 6 doses of 2.0 cc/kg/dose (12 cc/kg), and one dog received 4 doses of 5 cc/kg/dose (20 cc/kg). Based on the density of dipropylene glycol, these administered dosages are approximately equivalent to total doses of 9400, 12500, and 20800 mg/kg. Frequency of administration was not stated, but it was probably 3 times per day. Emesis and recovery was observed in one dog of the middle dose. No deaths and no other signs of toxicity were observed in any of the dogs (Hanzlik et al., 1939). The design limitations of the study hamper its applicability to the repeated exposure hazard characterization for dipropylene glycol.

In a study that was designed to assess effects of dipropylene glycol on the development of offspring to dams, investigators exposed pregnant rats to dipropylene glycol on gestation days 6 to 15. At 2000 mg/kg/day and above, some rats died and there were signs of toxicity as well as reduced body weight, food consumption and increased water consumption. The NOAEL for maternal toxicity was 800 mg/kg/day. In pregnant rabbits exposed on gestation days 6-19, no maternal toxicity was seen; hence, the NOAEL for the study was 1200 mg/kg/day (Bates et al., 1992a; 1992b).
A repeat exposure/reproductive screen study was conducted on tripropylene glycol, which is rapidly converted to dipropylene glycol \textit{in vivo}. Males and females exposed by gavage to tripropylene glycol for 14 days at 1000 mg/kg/day had increased liver weights; thus the study NOAEL was 200 mg/kg/day (Tanaka et al., year not reported).

The available data from repeated dosing studies indicate toxic effects only at very high doses of dipropylene glycol. The studies are summarized below:

| Species | Treatment | NOAEL/LOAEL  
| (mg/kg bw/day) | Comments | Source |
|---------|-----------|-----------|--------|--------|
| Rat     | 1% - 10% in drinking water for 9-77 d | 5% in drinking water; data do not permit calculation in terms of mg/kg/day | Some animals from 10% exposure group died; some had hydropic degeneration of tubule epithelium in kidneys. | Kesten et al., 1939 |
| Rat     | 12% in feed for 15 weeks | Not established | Decreased running activity after 15 weeks. | van Winkle and Kennedy, 1940 |
| Dog     | Total dose of 9, 12 or 20 cc/kg divided among multiple doses | 9400 | One of two dogs dosed at 12 cc/kg (12500 mg/kg) had emesis, but recovered. No other indication of toxicity in study. | Hanzlik et al., 1939 |
| Rat     | 800, 2000, and 5000 mg/kg/day days 6-15 of gestation | 800 mg/kg/day | Deaths, reduced food consumption and body weight at two higher doses | Bates et al., 1992a |
| Rabbit  | 200, 400, 800, and 1200 mg/kg/day days 6-19 of gestation | 1200 | No adverse effects seen. | Bates et al., 1992b |

### 4.1.4. Genotoxicity

Dipropylene glycol is not toxic to genetic material based on \textit{in vitro} and \textit{in vivo} study findings. The National Cancer Institute (NCI) conducted Ames assays of dipropylene glycol using S9 from livers of rats and hamsters induced by feeding of Aroclor 1254, and without metabolic activation, using strains TA98, TA100, TA 1535 and TA1537. It was a standard plate assay with concentrations of dipropylene glycol up to 10000 µg/plate. Tables of revertants at each dose level were not provided, but the results were characterized by NCI as negative (CCRIS, 2001). Similarly NCI conducted a mouse lymphoma assay of dipropylene glycol using L5178 (TK+/TK-) cells. Testing was conducted with and without metabolic activation using S9 from livers of rats fed Aroclor 1254. Concentrations of dipropylene glycol were 29-60 µl/ml without metabolic activation and 30-50 µl/ml with activation. Detailed results were not provided, but were characterized by NCI as negative (CCRIS, 2001).

Male CD-1 mice (6 per treatment group) were treated by gavage with 0, 500, 1000, or 2000 mg/kg/day dipropylene glycol for 2 consecutive days. Cyclophosphamide (120 mg/kg) was administered as a positive control. Bone marrow was isolated and stained for examination of 2000 polychromatic erythrocytes (PCE) for the presence of micronuclei. The number of cells with micronuclei ranged from 1.3 to 2.8 cells/2000 PCE in the treated and negative controls groups (Dow, 1999). None of the treated groups had significantly
more micronuclei than the controls or a change in the percentage of PCEs. In contrast, cyclophosphamide caused 40 micronuclei/2000 PCE, indicating the assay was sufficiently sensitive to detect induction of micronuclei by a known clastogen.

4.1.5. Carcinogenicity

Dipropylene glycol has not been tested specifically for carcinogenicity; however, based on the results of a cancer bioassay of propylene glycol (to which dipropylene glycol is rapidly converted), dipropylene glycol is not expected to have carcinogenic potential. Propylene glycol was administered in the diet of male and female Crl:CD rats at concentrations of 0, 6250, 12500, 25000, and 50000 ppm for 104 weeks. No increases in tumor incidence were found in the treated rats compared to the controls. Daily exposure to propylene glycol in the high-dose group was calculated at 1700 to 2100 mg/kg/day (Gaunt et al., 1972).

4.1.6. Reproductive/Developmental Toxicity

No reproduction studies have been conducted for dipropylene glycol; however the structural analogues tripropylene glycol and propylene glycol have been tested for reproductive effects and no effects on fertility and reproductive performance were found at doses that did not result in maternal toxicity. The negative reproductive findings for tripropylene glycol and propylene glycol support an expected similar lack of reproductive toxicity for dipropylene glycol because the metabolism data demonstrates that tripropylene glycol is rapidly hydrolyzed to dipropylene glycol, which is further rapidly hydrolyzed to propylene glycol.

Tripropylene glycol was administered by gavage from premating through day 3 of lactation at doses of 0, 8, 40, 200, and 1000 mg/kg/day to groups of 12 male and 12 female Crj:CD rats. There was no effect on paternal body weight, food consumption, hematology, clinical chemistry, necropsy or histopathology. Males and females that received 1000 mg/kg/day had increased liver weights and males also had increased kidney weights. The NOAEL for parental effects was 200 mg/kg/day. There were no differences in estrus cycle, mating, pregnancy rate, length of gestation, gestation index, mean litter size, mean live litter size, pup survival through day 4, sex ratio, or external abnormalities. The NOAEL for reproduction and developmental effects was >1000 mg/kg/day (Tanaka et al., year not reported).

In a continuous breeding study in mice, propylene glycol was administered at 0, 1, 2.5 and 5% in the drinking water of male and female CD-1 mice for 7 days premating and through 98 additional days of cohabitation. Estimated daily doses were: 1800, 4800 and 10100 mg/kg/day. The outcome of the study was that no adverse effects were found in the F0, F1 or F2 generation animals or in their ability to reproduce (Morrissey et al., 1989).

Developmental toxicity has been studied for dipropylene glycol using rats and rabbits. These studies found no differences in developmental parameters between exposed and control animals when dipropylene glycol was administered to rats at up to 5000 mg/kg/day days 6-15 of gestation or to rabbits at up to 1200 mg/kg/day days 6-19 of gestation.

Dipropylene glycol was administered to groups of 26 or 27 pregnant Crl:CD rats by gavage at doses of 800, 2000 or 5000 mg/kg/day on days 6-15 of gestation. Dams were necropsied on gestation day 20. One dam at 2000 mg/kg/day and two dams at 5000 mg/kg/day died prior to termination. At 2000 and 5000 mg/kg/day ataxia, unstable gait
and piloerection were seen in the dams. At 5000 mg/kg/day, dams also had reduced body weight gain, reduced food consumption and increased water consumption. Dipropylene glycol did not affect resorptions, number of live fetuses, sex ratio, fetal weight, or skeletal or visceral abnormalities or variations (Bates et al., 1992a).

Dipropylene glycol was administered to groups of 24 pregnant New Zealand White rabbits by gavage at doses of 200, 400, 800, or 1200 mg/kg/day on days 6-19 of gestation. Dams were necropsied on gestation day 30. There was no evidence of maternal toxicity, although mortality was seen in a preliminary study at 800 and 1500 mg/kg/day. Dipropylene glycol did not affect resorptions, number of live fetuses, sex ratio, fetal weight, or skeletal or visceral abnormalities or variations (Bates et al., 1992b).

The following NOAELS and LOAELS were obtained:

<table>
<thead>
<tr>
<th>Specie</th>
<th>Treatment mg/kg bw/d</th>
<th>Treatment period</th>
<th>Maternal LOAEL</th>
<th>Maternal NOAEL</th>
<th>Fetal LOAEL</th>
<th>Fetal NOAEL</th>
</tr>
</thead>
<tbody>
<tr>
<td>Rat</td>
<td>800, 2000, 5000</td>
<td>GD 6-15</td>
<td>2000</td>
<td>800</td>
<td>NA</td>
<td>&gt;5000</td>
</tr>
<tr>
<td>Rabbit</td>
<td>200, 400, 800, 1200</td>
<td>GD 6-19</td>
<td>NA</td>
<td>&gt;1200</td>
<td>NA</td>
<td>&gt;1200</td>
</tr>
</tbody>
</table>

Thus based on reproduction studies with tripropylene glycol and propylene glycol and developmental toxicity studies in rats and rabbits with dipropylene glycol, dipropylene glycol represents a low hazard potential for developmental or reproductive effects in humans.

4.1.7. Toxicokinetics

Dipropylene glycol is readily absorbed from the gastrointestinal tract. Once absorbed it is readily converted into propylene glycol, which is further converted to lactic and pyruvic acids. These acids are normal body constituents and are further broken down to carbon dioxide and water or incorporated into glycogen. Propylene glycol that is not metabolized is excreted into the urine. These conclusions are drawn from a metabolism study of tripropylene glycol and propylene glycol (Dow, 1995). $^{14}$C-Labelled tripropylene glycol was administered by gavage to male F344 rats; exhaled air, urine and feces were collected for 24 hours. After 24 hours the animals were sacrificed for collection of blood, liver, kidney, fat, brain, muscle, and the remaining carcass. 73% of administered dose was recovered as CO$_2$ (21%) or urinary metabolites (53%); 10% remained in the tissues. The urine contained 13%, 8%, and 4% of administered dose as tripropylene glycol + tripropylene glycol conjugates, dipropylene glycol + dipropylene glycol conjugates, and propylene glycol + propylene glycol conjugates, respectively. When propylene glycol was similarly administered, only urine was collected. 12% of the administered dose was found in the urine. This study demonstrates that tripropylene glycol is rapidly metabolized to dipropylene glycol which is further metabolized to propylene glycol to lactic and pyruvic acids which either enter the citric acid cycle with exhaled CO$_2$ as a byproduct of energy production, or are incorporated into tissue components such as glycogen.
5. Conclusions and Recommendations

5.1 Conclusions

Commercial dipropylene glycol (CAS # 25265-71-8; (CH\textsubscript{3}-CHOH-CH\textsubscript{2}-CH\textsubscript{2}-CHOH-CH\textsubscript{3}) is composed of 3 isomers and is typically 98% pure. The commercial product is typically composed of up to 48% isomer 110-98-5. Dipropylene glycol is produced as a byproduct of the manufacture of propylene glycol. The US production capacity of dipropylene glycol was 131 million pounds (60 thousand tonnes) in 1998. Dipropylene glycol is used as both a reactive intermediate and as a solvent. Reactive intermediate end uses in the US include: plasticizers, unsaturated polyester resins, polyurethane polyols, and alkyd resins. Solvent end uses in the US include: cosmetics, pesticides and functional fluids: specialty deicers, inks, lubricants.

Dipropylene glycol is a liquid at room temperature, it has a low vapor pressure, and is miscible with water. Releases to the environment are expected to partition primarily to water and soil where they will be degraded. It is not expected to bioaccumulate due to BCFs measured at 0.3 to 4.6. Testing in aquatic species showed a low hazard concern. PNECs of >50 mg/L were obtained from a fish LC\textsubscript{50} of 5000 mg/l and 32 mg/l from a clawed toad (Xenopus laevis) LC\textsubscript{50} of 3181 mg/l. The mammalian acute toxicity of dipropylene glycol is low, with values around 16000 mg/kg reported in tests in rats. Dipropylene glycol is only minimally irritating to the skin and eye. It appears to have low potential to produce allergic skin reactions, as only 1 of 503 patients with eczema reacted to dipropylene glycol. Longer-term studies of dipropylene glycol and the structural analogues, propylene glycol and tripropylene glycol, demonstrate a low concern for chronic, reproductive, and developmental effects. Dipropylene glycol is not genotoxic and not expected to be carcinogenic.

5.2 Recommendations

The chemical is currently a low priority for further work.
6. References


CCRIS (Chemical Carcinogenesis Research Information System from National Cancer Institute). Dipropylene glycol; available online through ToxNet, listing on 4/25/01 at http://toxnet.nlm.nih.gov/cgi-bin/sis/htmlgen?CCRIS.


SIDS DOSSIER
Dipropylene glycol, mixed isomers and dominant isomer (DPG)
(CAS No. 25265-71-8 and 110-98-5)

Sponsor Country: U.S.A.
I. GENERAL INFORMATION

1.01 SUBSTANCE INFORMATION

A. CAS-Number: 25265-71-8 and 110-98-5

B. Name (IUPAC name): dipropylene glycol, mixed isomers and dominant isomer structural isomers (as mixture in the technical product):
   - (1) 2,2'-dihydroxydiisopropylether (CAS-No.: 108-61-2);
   - (2) 2,2'-dihydroxydipropylether (CAS-No.: 110-98-5);
   - (3) 2-hydroxypropyl-2'-hydroxyisopropylether (CAS-No.: 106-62-7);

C. Name (OECD name): DPG

D. CAS Descriptor

E. EINECS-Number

F. Molecular Formula: C₆H₁₄O₃

G. Structural Formula: CH₃-CHOH-CH₂O-CH₂-CHOH-CH₃

H. Substance Group

I. Substance Remark

J. Molecular Weight

1.02 OECD INFORMATION

A. Sponsor Country: U.S.A.

B. Lead Organisation:

Name of Sponsor Country: United States of America
Contact point: Oscar Hernandez
EPA/Office of Toxic Substances RAD (7403M)
1200 Pennsylvania Ave, NW
Washington, D.C. 20460
Telephone: (202) 564-7649
Fax: (202)-564-7450

Name of Lead Organisation: American Chemistry Council Propylene Glycol Ethers Panel
Contact person: Anne LeHuray.
Address: American Chemistry Council
1300 Wilson Blvd.
Arlington, VA 22209
U.S.A.
Tel: 703-741-5630
Fax: 703-741-6091
Email: LeHuray_Anne@americanchemistry.com
1.1 GENERAL SUBSTANCE INFORMATION

A. Type of Substance
   element [ ]; inorganic [ ]; natural substance [ ]; organic [ X ];
   organometalic [ ]; petroleum product [ ]

B. Physical State (at 20°C and 1.013 hPa)
   gaseous [ ]; liquid [ X ]; solid [ ]

C. Purity (indicate the percentage by weight/weight)

1.2 SYNONYMS:
   1,1'-Oxybis-2-propanol; 1,1'-Oxydi-2-propanol; 2,2'-
   Dihydroxydipropylether; Bis(2-hydroxypropyl)ether; Di-1,2-
   propyleneglycol; dipropylene glycol; dipropyleneglycol2-(2-
   hydroxypropoxy)-1-propanol (CAS No. 108-61-2)
   2,2’oxygenbis-1-propanol (CAS No. 106-62-7)

1.3 IMPURITIES

1.4 ADDITIVES

1.5 QUANTITY

Dipropylene glycol is produced as a byproduct of the manufacture of
propylene glycol. The United States (US) production capacity of
dipropylene glycol (i.e., byproduct production capacities of propylene
glycol plants) was 131 million pounds (60 thousand tonnes) in 1998. The
US Domestic demand in 1998 was 108 million pounds (49 thousand
 tonnes). In 1998, dipropylene glycol was produced in the US by The Dow
Chemical Company, Eastman Chemical Company, Huntsman Corporation,
Lyondell Chemical Company, and Olin Corporation. (ChemExpo Chemical
Profile, 1998). Worldwide capacity is estimated at 235 million pounds
(107 thousand tonnes).

1.6 LABELLING AND CLASSIFICATION

Labelling
Type:
Specific limits:
Symbols:
Nota:
R-phrases:
S-phrases:
Text of S-phrases:
Remarks:

Classification
Type:
Category of danger:
R-phrases:
Remarks:
1.7 USE PATTERN

A. General

Type of Use:

<table>
<thead>
<tr>
<th>USES</th>
<th>APPLICATION</th>
<th>FUNCTION</th>
<th>% PRODUCTION</th>
</tr>
</thead>
<tbody>
<tr>
<td>Intermediate</td>
<td>Specialty plasticizer</td>
<td>Plasticizer</td>
<td>38 %</td>
</tr>
<tr>
<td>Intermediate</td>
<td>Non-reinforced polyesters</td>
<td>Resin monomer</td>
<td>23 %</td>
</tr>
<tr>
<td>Substance</td>
<td>Cosmetics and Fragrances</td>
<td>Humectant, Emollient</td>
<td>10 %</td>
</tr>
<tr>
<td>Intermediate</td>
<td>Polyurethane polyols</td>
<td>Plasticizer</td>
<td>10 %</td>
</tr>
<tr>
<td>Intermediate</td>
<td>Alkyd resins</td>
<td>Resin monomer</td>
<td>7 %</td>
</tr>
<tr>
<td>Substance, Intermediate</td>
<td>Miscellaneous (e.g., dyes &amp; inks, paints &amp; coatings; functional uses in hydraulic brake fluids; cutting oils)</td>
<td>Solvent, functional fluid, coolant</td>
<td>14%</td>
</tr>
</tbody>
</table>

(Source: ChemExpo Chemical Profile (1998)).

B. Uses in Consumer Products

Cosmetics, fragrances, germicidal products

The use concentration of dipropylene glycol in branded pesticidal products is also available. This is summarized in the following table.

<table>
<thead>
<tr>
<th>PESTICIDAL FORMULATIONS CONTAINING DIPROPYLENE GLYCOL</th>
</tr>
</thead>
<tbody>
<tr>
<td>RANGE OF CONCENTRATION (%)</td>
</tr>
<tr>
<td>No. of Brand Name Products</td>
</tr>
</tbody>
</table>

(Source: EDF Website: www.scorecard.org)

1.8 OCCUPATIONAL EXPOSURE LIMIT VALUE

1.9 SOURCES OF EXPOSURE

(a)
Media of release:
Source:
Remarks:
Reference:

(b)
Media of release:
Source:
Remarks:
Reference:
1.10 ADDITIONAL REMARKS

A. Options for disposal

Remarks:
Reference:

B. Other remarks
2. PHYSICAL-CHEMICAL DATA

2.1 MELTING POINT

Value = -39 degree C – pour point
Method
GLP: No data

2.2 BOILING POINT

Value = 222.2 - 225.7 degree C
Pressure: 1013 hPa
Decomposition
GLP: No data

Value = 232 degree C
Pressure
Decomposition
GLP: No data

Value = 233 degree C
Pressure
Decomposition
GLP: No data

2.3 DENSITY

Value = 1.023 g/cm³
Temperature: 20 degree C
GLP: No data

Value = 1.0252 g/cm³
Temperature: 20 degree C
GLP: No data
2.4  VAPOUR PRESSURE

Value = <.01 hPa
Temperature: 20 degree C
GLP: No data


Value = <.013 hPa
Temperature: 20 degree C
GLP: No data


Value = .05 hPa
Temperature: 21 degree C
GLP: No data


2.5  PARTITION COEFFICIENT $\log_{10} P_{ow}$

$\log P_{ow} = -1.486$
Temperature degree C
GLP: No data


$\log P_{ow} = -.687$
Temperature degree C
GLP: No data


2.6  WATER SOLUBILITY

Described as miscible
GLP: No data


2.7 **FLASH POINT** (*liquids*)

Value = 118 degree C  
Type: other  
GLP: No data  
Method: ASTM D56.


Value = 120 degree C  
Type: closed cup  
GLP: No data  
Method: DIN 51758


Value = 121 degree C  
Type: closed cup  
GLP: No data  
Method: Pensky Martens


Value = 138 degree C  
Type: other  
GLP: No data  
Method: other


2.8 **AUTO FLAMMABILITY** (*solid/gases*)

Value = 371 degree C  
Pressure  
GLP: No data

Reference: Material Safety Data Sheet, Arco Chemical Company, 1993

2.9 **FLAMMABILITY**

No data

2.10 **EXPLOSIVE PROPERTIES**

No data
2.11 OXIDIZING PROPERTIES

No data

2.12 ADDITIONAL REMARKS

Remarks: No additional remarks

2.13 ADDITIONAL DATA

A. Partition co-efficient between soil/sediment and water (Kd)

Value:
Method:
GLP:
Remarks: No studies located
Reference:

B. Other data

Results: No studies located
Remarks:
Reference:
3. ENVIRONMENTAL FATE AND PATHWAYS

3.1 STABILITY

3.1.1 PHOTODEGRADATION

(a) Type: Air [X]; Water []; Soil []; Other []
Light source: Sun light []; Xenon lamp []; Other [X]
Light spectrum:
Relative intensity:
Concentration of Substance: molecule/cm³

Temperature:
Direct photolysis:
Half life: 13 hours
Degradation:
Quantum yield:

Method: calculated []; measured []
Other
GLP: Yes [] No [] ? [x]
Test substance:
Remarks:

Result: The rate constant for the reaction of DPG with photochemically induced OH radicals was calculated by a mathematical model (AOP) to a value of 29.7 x 10⁻¹¹ cm³/molecule x sec at a temperature of 25 degrees C. This results in an atmospheric half life of ca. 13 hours at an atmospheric concentration of 5 x 10⁵ OH radicals per cm³.


Type: Air [ ]; Water [X]; Soil []; Other []
Rel. Intens. based on Intensity of Sunlight
Spectrum of Substance
λ (max) nm ε (max)
Test condition:
Rate Constant cm³/(molecule*sec)
GLP: no data
Rate constant:
Remark: According to the λ-spectrum of DPG (i.e., no relevant UV absorption above 290 nm) only a minimal tendency for direct photolysis can be expected.


3.1.2 STABILITY IN WATER

Remark: Ethers and glycols are generally regarded as resistant towards hydrolysis. Therefore, DPG is not expected to undergo hydrolysis under environmentally relevant conditions.

3.1.3 STABILITY IN SOIL

No data available

3.2 MONITORING DATA (ENVIRONMENT)

Medium: other: drinking water

Remark: In a study on drinking water and waste water in Ohio (USA) during 1978-1980, DPG was found in samples of drinking water at concentrations between 0.2 and 0.4 ng/L.


Remark: A study on the priority pollutants found in the effluent of 5 paper mill waste water treatment plants in USA and Canada revealed a DPG concentration of 11 µg/l in the effluent of one of the plants. It was not detected in the influent stream nor in the untreated waste stream.


Medium: other: ground water

Remark: The occurrence of DPG in ground water samples taken in the vicinity of a landfill in Norman, Oklahoma (USA) was demonstrated qualitatively in a study performed during 1972-1973.


3.3 TRANSPORT AND DISTRIBUTION BETWEEN ENVIRONMENTAL COMPARTMENTS INCLUDING ESTIMATED ENVIRONMENTAL CONCENTRATIONS AND DISTRIBUTION PATHWAYS

3.3.1 TRANSPORT BETWEEN ENVIRONMENTAL COMPARTMENTS

| Type: Adsorption | [X]; Desorption | [ ]; Volatility | [ ]; Other | [ ] |
| Media: Water-Soil |
| Method: Other |
| Remarks: Due to lack of valid data, a rough estimation was made on the basis of a calculated range for the log $P_{ow}$ value of -0.69 to -1.486. These theoretical considerations lead to an expected range for the $K_{oc}$ value of DPG = 3.0 - 56.2. |

The tendency of DPG to adsorb onto soil must be regarded as very low.

Type: Adsorption [ ]; Desorption [ ]; Volatility [ ]; Other [ X]
Media: Water-Air
Method: Other
Remarks: Henry Law Constant: \(3.6 \times 10^{-4}\) Pa*m\(^3\)/mole.
The volatility of DPG from water can, therefore, be considered as minimal.


3.3.2 THEORETICAL DISTRIBUTION (FUGACITY CALCULATION)

No data available

3.4 IDENTIFICATION OF MAIN MODE OF DEGRADABILITY IN ACTUAL USE

No data available

3.5 BIODEGRADATION

Type: aerobic
Inoculum: activated sludge, adapted
Concentration:
GLP: No data
Results: > 70 % degraded after 28 day

Type: aerobic
Inoculum: activated sludge
Method: not given; 30 mg/l substance; 100 mg/l sludge
GLP: No data
Results: under test conditions no biodegradation observed

Type: aerobic
Inoculum: Corynebacterium sp. (Bacteria)
Concentration: 5 g/l
GLP: No data
Results: Degradation: > 90 % after 23 hour
Remark Degradation of DPG to propylene glycol and ketones depended on the shaking conditions. In stationary culture, DPG was scarcely degraded. With vigorous shaking (about 120 rpm), over 90% of DPG was consumed within 23 h, but traces
of metabolites accumulated in the reaction mixture. The metabolites were characterized by GC-MS analysis using a capillary column.


**Type:** aerobic  
**Inoculum:** other: soil bacterium (aerobic, gram-negative)  
**Concentration:** 671 ug/l  
**Results:** Degradation: >70% after 28 day; 2% degradation after 4 hours related to oxygen uptake.  
**GLP:** No data  
**Remark:** The bacterium was isolated from soil enriched with triethylene glycol. Cells to be assayed for oxidative activity were cultured at 30 degree C on a shaker for 72 to 96 hr in basal medium.


### 3.6 BOD$_5$, COD OR RATIO BOD$_5$/COD

**BOD5/COD**

**Method:** APHA standard methods: No. 219  
**GLP:** No data  
**Results:**  
BOD$_5$ = 92 mg/l  
COD = 1840 mg/g  
BOD$_5$/COD = .049  

### 3.7 BIOACCUMULATION

**Species:** *Cyprinus carpio* (Fish, fresh water)  
**Method** OECD Guideline 305C “Bioaccumulation: Test for the Degree of Bioconcentration in Fish”

**Exposure Period:** 42 days; Temperature 25 degree C; Concentration: 3 mg/l  
**Elimination:** No data  
**GLP:** No data  
**BCF:** 0.3 to 1.4  

**Species:** *Cyprinus carpio* (Fish, fresh water)  
**Method** OECD Guideline 305C “Bioaccumulation: Test for the Degree of Bioconcentration in Fish”

**Exposure Period:** 42 days; Temperature 25 degree C; Concentration: 3 mg/l  
**Elimination:** No data  
**GLP:** No data  
**BCF:** <2.2 to 4.6  
3.8 ADDITIONAL REMARKS

A. Sewage Treatment
   Remarks: No additional remarks

B. Other
   Remarks: No additional remarks
4. **ECOTOXICOLOGICAL DATA**

4.1 **ACUTE/PROLONGED TOXICITY TO FISH**

Type: static  
Species: *Carassius auratus* (Fish, fresh water)  
Method: Static-tank acute toxicity test. Standard methods for the examination of water and wastewater. APHA method No. 231.; Exposure Period 96 hour  
Analyt. Monitoring: no data  
GLP: No data  
Results: LC50 = >5000 mg/l  

4.2 **ACUTE TOXICITY TO AQUATIC INVERTEBRATES**

No data available

4.3 **TOXICITY TO AQUATIC PLANTS e.g. Algae**

No data available

4.4 **TOXICITY TO BACTERIA**

Species: Bacterium *Pseudomonas putida*  
Method: UBA Richtlinie LTWS - Nr. 10; DIN 38412L8; Exposure Period: 18 hr  
GLP: No data  
Results: Bacterial growth was inhibited (13.5%) at a concentration of 100 mg/l DPG.  

Species: Bacterium *Pseudomonas putida*  
Method: Conformed to German Water Hazard classification; Exposure Period 18 hour  
Analyt. Monitoring: yes  
GLP: Yes  
Results: The mean concentration of dipropylene glycol which demonstrated a 10% growth inhibition (EC10) of *Pseudomonas putida* was 1,000 mg/L.  

4.5 **CHRONIC TOXICITY TO AQUATIC ORGANISMS**

4.5.1. **CHRONIC TOXICITY TO FISH**

No data available
4.5.2. CHRONIC TOXICITY TO AQUATIC INVERTEBRATES

No data available

4.6 TOXICITY TO TERRESTRIAL ORGANISMS

4.6.1 TOXICITY TO SOIL DWELLING ORGANISMS

No data available

4.6.2 TOXICITY TO TERRESTRIAL PLANTS

No data available

4.6.3 TOXICITY TO OTHER NON MAMMALIAN TERRESTRIAL SPECIES (INCLUDING AVIAN)

No data available

4.7 BIOLOGICAL EFFECTS MONITORING (INCLUDING BIOMAGNIFICATION)

No data available

4.8 BIOTRANSFORMATION AND KINETICS

No data available

4.9 ADDITIONAL REMARKS

Toxicity to Aquatic Animals:

Species: *Rana brevipoda porosa*
Method: not specified; Exposure Period: 48 hours
GLP: No data
Results: $LC_{50} = 5300 \text{ mg/l}$

Species: *Xenopus laevis*
Method: not specified; Exposure Period 48 hour
GLP: No data
Results: $LC_{50} = 3181 \text{ mg/l}$
5. **TOXICITY**

5.1 **ACUTE TOXICITY**

5.1.1 **ACUTE ORAL TOXICITY**

Species: rat  
Method: details not given  
GLP: no data  
Results: LD$_{50}$ = 14800 mg/kg  
Effects: Details not Reported.  


Species: rat  
Method: details not given  
GLP: No data  
Results: LD$_{50}$ = 14850 mg/kg  

Species: rat  
Method: details not given  
GLP: No data  
Results: LD$_{50}$ = 15000 mg/kg  


Species: rat  
Method: single oral dose; details not given  
GLP: No data  
Results: LD$_{50}$ = 15.0 and 8.0 ml/kg-bw for males and females, respectively  
Reference: Dow Chemical Europe, Horgen, Switzerland. Determination of the acute oral toxicity of dipropylene glycol in rats with cover letter dated 03/28/94 (sanitized). Doc ID# 869400000276S.
Species: Rat
Method: details not given
GLP: No data
Results: LD0 = 3000 mg/kg; LD100 = 15000 mg/kg

Species: Guinea pig
Method: details not given
GLP: No data
Results: LD50 = 17600 mg/kg
BIBRA Toxicology International - British Industrial Biological Research Association, 1-4.

5.1.2 ACUTE INHALATION TOXICITY

Species: Rat
Method: details not given
GLP: No data
Results: LC50 = 6000 mg/m³

5.1.3 ACUTE DERMAL TOXICITY

Species: rabbit
Method: not specified
GLP: No data
Results: LD50 = >5000 mg/kg
BIBRA Toxicology International - British Industrial Biological Research Association, 1-4.


Species: rabbit
Method: not specified
GLP: No data
Results: LD50 = >20000 mg/kg

5.1.4 ACUTE TOXICITY, OTHER ROUTES OF ADMINISTRATION

Species: rat
Route: i.p.
Method: details not given
GLP: No data
Results: LD50 = 10000 mg/kg.
Method: details not given
GLP: No data
Results: LD50 = 10300 mg/kg.


Species: rat
Route: i.p
Method: details not given
GLP: No data
Results: LD50 = 10590 mg/kg

Species: mouse
Route: i.p.
Method: details not given
GLP: No data
Results: LD50 = 4600 mg/kg


Species: mouse
Route: i.p.
Method: details not given
GLP: No data
Results: LD50 = 4500 mg/kg

Species: dog
Route: i.p.
Method: details not given
GLP: No data
Results: LD50 = 11500 mg/kg

Species: rat
Route: i.v.
Method: details not given
GLP: No data
Results: LD50 = 5800 mg/kg

Species: dog
5.2 CORROSIVENESS/IRRITATION

5.2.1 SKIN IRRITATION/CORROSION

Species: rabbit
Method: other
GLP: No data
Results: 10 applications of DPG over 12 days produced negligible irritation of rabbit skin.

Species: rabbit
Method: uncovered applications, 5 days/w for 6 w.
GLP: no data
Results: neat material showed only slight irritation, while a 10% aqueous solution did not cause any local reactions in rabbits.

Species: rabbit
Method: not specified
GLP: No data
Results: slightly irritating; Application (generally covered) of DPG caused slight irritation for 24 h to intact or abraded rabbit skin.

Species: rabbit
Method: 500 mg undiluted DPG was applied to rabbit skin for 24 hours.
GLP: No data
Results: slightly irritating
5.2.2 EYE IRRITATION/CORROSION

Species: rabbit
Method: not specified
GLP: No data
Results: not irritating.

Species: rabbit
Method: Dose: 500 mg DPG for 24 hours or a formulation containing 7.2% of DPG
GLP: No data
Results: from neat DPG - mild irritation. No more than minimal transient irritation from 7.2% DPG formulation

5.3 SKIN SENSITISATION

No animal data available. Please see section 5.11.

5.4 REPEATED DOSE TOXICITY

Oral:

Species: male/female rats, strain not reported
Method: not reported
Exposure Period: 9-77 days
Doses: 1-10% in drinking water
GLP: No data
Results: No adverse effects in rats exposed up to 5%. In rats exposed to 10% DPG in drinking water 9-68 days, 36% developed kidney lesions; 7 of 25 died.

Species: dog, strain and sex not reported
Method: By gavage. Dosing described as “divided doses”, no specifications on how frequently doses were administered; in another part of the study doses of diethylene glycol were administered 3 times daily.
Exposure Period: not stated
Doses: 20 cc/kg bw in 4 applications; 9 cc/kg bw in 6 applications; 12 cc/kg bw in 6 applications
OECD SIDS

DIPROPYLENE GLYCOLE

GLP: No data
Results: No effects in 3 of 4 dogs; 1 at 12 cc/kg – emesis and recovery; minimal liver damage; moderate degeneration of kidneys in 2 of 4 dogs.

Species: rat; strain and sex not specified
Method: not reported
Exposure Period: 15 weeks
Feed Concentrations: 0, 12%
GLP: No data
Results: No effects in 3 of 4 dogs; 1 at 12 cc/kg – emesis and recovery; minimal liver damage; moderate degeneration of kidneys in 2 of 4 dogs.

Species: chicks; strain and sex not specified
Method: not reported
Exposure Period: 27 days
Feed Concentrations: 5% (approximately 6.25 g/kg bw/day)
GLP: No data
Results: No adverse effects

Intravenous Studies:

Species: rabbit; strain and sex not specified
Method: not given
Duration: 1-21 days
Doses: 2-4 cc/kg
GLP: No data
Results: Dosing at 4 cc/kg/day resulted in 50% incidence of renal changes (not specifically described) and 40% mortality. No effect was seen at 2 cc/kg.

5.5 GENETIC TOXICITY IN VITRO

A. BACTERIAL IN VITRO TEST

Type: Ames test
System of Testing: Salmonella typhimurium TA 98, TA 100, TA 1535, TA 1537
Concentration: 100 – 10,000 µg/plate
GLP: No data
Result: Negative

Type: Ames test
System of Testing: TA98, TA100, TA1535, TA1537, TA1538 with and without metabolic activation
Concentration: 0.102 - 102 µg/plate
Method: OECD Guideline 471 “Genetic Toxicology: Salmonella typhimurium Reverse Mutation Assay”
GLP: No data
Result: Negative

B. NON-BACTERIAL IN VITRO TEST

Type: Mouse lymphoma assay
Method: OECD Guideline 473
System of Testing: L5178Y (Tk+/TK-) with and without metabolic activation
Concentration: 30 - 60 µl/ml
GLP: Yes
Result: Negative.

5.6 GENETIC TOXICITY IN VIVO

Test type: Micronucleus assay
Test Species: CD-1 mouse, male
Test Substance: Dipropylene glycol
Test Method: OECD #474 (1997); EPA OPPTS 870.5395 (1998)
Route of Exposure: oral gavage
Frequency of Exposure: once per day
Duration of Exposure: 2 consecutive days
Post Exposure Observation Period: none
Doses: 0, 500, 1000, 2000 mg/kg/day
GLP: Yes
Test Results: The study was negative.

5.7 CARCINOGENICITY

No data available.

5.8 TOXICITY TO REPRODUCTION

No data available

5.9 DEVELOPMENTAL TOXICITY/TERATOGENICITY

Type: Teratology
Method: not specified
Species: female Sprague-Dawley rat
Route of Administration: gavage
Doses: 0, 800, 2000, 5000 mg/kg of DPG daily days 6-15 of gestation period
GLP: No data
Results: Maternal Effects: Maternal toxicity and lethality were observed at 2000 and 5000 mg/kg/day (mortality rate: 4% and 9%), establishing the maternal NOAEL as 800 mg/kg/day.

Embryo/Fetal Effects: There were no significant differences between the DPG exposed groups and the control. NOAEL was 5000 mg/kg/day.


Type: Teratology
Method: not specified
Species: female New Zealand White rabbit
Route of Administration: gavage
Doses: 0, 200, 400, 800, 1200 mg/kg of DPG daily days 6-19 of gestation period
GLP: No data
Results: Maternal Effects: No maternal toxicity or lethality was observed at any doses level.

Embryo/Fetal Effects: There were no significant differences between the DPG exposed groups and the control. NOAEL was 1200 mg/kg/day.


5.10 OTHER RELEVANT INFORMATION

Type: metabolism
Remark: Tripropylene glycol was administered by oral gavage to male F344 rats. TPG was readily converted to DPG and then to PG, which was further metabolized via the citric acid cycle to CO₂ or excreted in the urine.


Type: Biochemical or cellular interactions
Remark: In experiments with mice the exposure of DPG (contained in a commercial deodorant spray) for 2 hours caused a rise in the ozone sensibility of the liver (measured as an increase of the reduced form of ascorbic acid in the liver).


Type: Excretion of Bile
Remark: The application of 1 ml/kg DPG (intraduodenal) to rats increased the excretion of bile.


5.11 EXPERIENCE WITH HUMAN EXPOSURE

Skin Irritation:
Species: Human
Method: 0.2 ml of a 25% solution of dipropylene glycol in distilled water for 24 hours via semi-occluded patches. 33 subjects
GLP: No data
Results: 2 subjects no effects at 30 min. but mild erythema at 24 hrs.
2 subjects mild to moderate erythema at 30 min. and mild erythema at 24 hrs.
7 subjects mild to moderate erythema at 30 min. but no effects at 24 hrs.
22 subjects no reaction at 30 min or 24 hr.

Species: Human
Method: Covered 48-h application of a 20% or 50% solution of DPG in petrolatum.
GLP: No data
Results: 50% solution caused irritation in 14 of 34 persons and was equivocally an irritant in a further 17.
20% solution caused no irritation number tested not stated.
7.2% shaving prep no irritation open application, mild irritation in 6 of 101 subjects in closed patch.

Sensitization:
Species: Human
Method: Not specified
GLP: No data
Results: Only 1 patient out of 503 eczema patients displayed a positive patch test to dipropylene glycol (96% purity) in an eczematous population.

Species: Human
Method: 24/48-h (presumably covered) patches, 3 days/week for 3 weeks, followed by a challenge patch after a 2-w rest period. Supplemental exposure to UV light after application of the first four patches and the challenge patch. 50 subjects
GLP: No data
Results: No indication of sensitization or photosensitization.
6. REFERENCES


BUA report (1994): Dipropylenglykol


Dow Chemical Europe, Horgen, Switzerland. Determination of the acute oral toxicity of dipropylene glycol in rats with cover letter dated 03/28/94 (sanitized). Doc ID# 8694000276S.


Material Safety Data Sheet, Arco Chemical Company, 1993


Spencer, H.C. (1946): A summary of the toxicological information on dipropylene glycol and a discussion of the hazards associated with its handling and use. Dow Chemical Company - Biochemical Research Laboratory.


**I U C L I D Data Set**

**Existing Chemical**
- **ID:** 25265-71-8
- **CAS No.:** 25265-71-8
- **EINECS Name:** oxydipropanol
- **EINECS No.:** 246-770-3
- **TSCA Name:** Propanol, oxybis-
- **Molecular Formula:** C₆H₁₄O₃

**Producer Related Part**
- **Company:** ACC Propylene Oxide/Propylene Glycol Panel
- **Creation date:** 23.05.2001

**Substance Related Part**
- **Company:** ACC Propylene Oxide/Propylene Glycol Panel
- **Creation date:** 23.05.2001

**Memo**

**Printing date:** 30.05.2001

**Revision date:**

**Date of last Update:** 30.05.2001

**Number of Pages:** 3

**Chapter (profile):** Chapter: 1, 2, 3, 4, 5, 7

**Reliability (profile):** Reliability: without reliability, 1, 2, 3, 4

**Flags (profile):** Flags: without flag, confidential, non confidential, WGK (DE), TA-Luft (DE), Material Safety Dataset, Risk Assessment, Directive 67/548/EEC, SIDS
1.0.1 OECD AND COMPANY INFORMATION

Type : 
Name : Lyondell Chemical Company
Partner : 
Date : 
Street : 1221 McKinney Street Suite 1600
Town : 77010 Houston Texas
Country : United States
Phone : 713-652-7200
Telefax : 
Telex : 
Cedex : 
Source : Lyondell Chemical Co. Houston, Texas
11.05.2001

Type : 
Name : Huntsman Corporation
Partner : 
Date : 
Street : 500 Huntsman Way
Town : 84108 Salt Lake City, Utah
Country : United States
Phone : 1-800-421-2411
Telefax : 801-584-5781
Telex : 
Cedex : 
Source : Lyondell Chemical Co. Houston, Texas
30.05.2001

Type : 
Name : The Dow Chemical Company
Partner : 
Date : 
Street : 2030 Dow Center
Town : 48674 Midland, MI
Country : United States
Phone : 517-636-1000
Telefax : 517-636-4033
Telex : 
Cedex : 
Source : Lyondell Chemical Co. Houston, Texas
29.05.2001

1.0.2 LOCATION OF PRODUCTION SITE

1.0.3 IDENTITY OF RECIPIENTS

1.1 GENERAL SUBSTANCE INFORMATION

Substance type : organic
Physical status : liquid
<table>
<thead>
<tr>
<th>Purity</th>
<th>&gt;= 98 % w/w</th>
</tr>
</thead>
<tbody>
<tr>
<td>Source</td>
<td>Lyondell Chemical Co. Houston, Texas</td>
</tr>
<tr>
<td>Date</td>
<td>30.04.2001</td>
</tr>
</tbody>
</table>

### 1.1.0 DETAILS ON TEMPLATE

#### 1.1.1 SPECTRA

#### 1.2 SYNONYMS

Di-1,2-propyleneglycol

**Remark**
common name: dipropyleneglycol

structural isomers (as mixture in the technical product):
1. 2,2-Dihydroxydiisopropylether (CAS-No.: 108-61-2);
2. 2,2-Dihydroxydipropylether (CAS-No.: 110-98-5);
3. 2-Hydroxypropyl-2'-hydroxyisopropylether (CAS-No.: 106-62-7);

**Source**
Lyondell Chemical Co. Houston, Texas

**Flag**
Critical study for SIDS endpoint

21.05.2001

### 1.3 IMPURITIES

### 1.4 ADDITIVES

### 1.5 QUANTITY

### 1.6.1 LABELLING

### 1.6.2 CLASSIFICATION

### 1.7 USE PATTERN

**Type**
type

**Category**
Non dispersive use

**Source**
Lyondell Chemical Co. Houston, Texas

09.05.2001

**Type**
type

**Category**
Use resulting in inclusion into or onto matrix

**Source**
Lyondell Chemical Co. Houston, Texas

09.05.2001

**Type**
type
<table>
<thead>
<tr>
<th>Category</th>
<th>Source</th>
<th>Date</th>
</tr>
</thead>
<tbody>
<tr>
<td>Wide dispersive use</td>
<td>Lyondell Chemical Co. Houston, Texas</td>
<td>09.05.2001</td>
</tr>
<tr>
<td>Basic industry: basic chemicals</td>
<td>Lyondell Chemical Co. Houston, Texas</td>
<td>09.05.2001</td>
</tr>
<tr>
<td>Chemical industry: used in synthesis</td>
<td>Lyondell Chemical Co. Houston, Texas</td>
<td>09.05.2001</td>
</tr>
<tr>
<td>Electrical/electronic engineering industry</td>
<td>Lyondell Chemical Co. Houston, Texas</td>
<td>09.05.2001</td>
</tr>
<tr>
<td>Leather processing industry</td>
<td>Lyondell Chemical Co. Houston, Texas</td>
<td>09.05.2001</td>
</tr>
<tr>
<td>Personal and domestic use</td>
<td>Lyondell Chemical Co. Houston, Texas</td>
<td>09.05.2001</td>
</tr>
<tr>
<td>Public domain</td>
<td>Lyondell Chemical Co. Houston, Texas</td>
<td>09.05.2001</td>
</tr>
<tr>
<td>Textile processing industry</td>
<td>Lyondell Chemical Co. Houston, Texas</td>
<td>09.05.2001</td>
</tr>
<tr>
<td>Cleaning/washing agents and disinfectants</td>
<td>Lyondell Chemical Co. Houston, Texas</td>
<td>09.05.2001</td>
</tr>
<tr>
<td>Cosmetics</td>
<td>Lyondell Chemical Co. Houston, Texas</td>
<td>09.05.2001</td>
</tr>
<tr>
<td>Intermediates</td>
<td>Lyondell Chemical Co. Houston, Texas</td>
<td>09.05.2001</td>
</tr>
<tr>
<td>Odour agents</td>
<td>Lyondell Chemical Co. Houston, Texas</td>
<td>21.05.2001</td>
</tr>
<tr>
<td>Use</td>
<td>Lyondell Chemical Co. Houston, Texas</td>
<td>09.05.2001</td>
</tr>
</tbody>
</table>

**Id:** 25265-71-8  
**Date:** 30.05.2001
## 1. General Information

<table>
<thead>
<tr>
<th>Category</th>
<th>Solvents</th>
</tr>
</thead>
<tbody>
<tr>
<td>Source</td>
<td>Lyondell Chemical Co. Houston, Texas</td>
</tr>
<tr>
<td>Date</td>
<td>30.05.2001</td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>Type</th>
<th>use</th>
</tr>
</thead>
<tbody>
<tr>
<td>Category</td>
<td>Stabilizers</td>
</tr>
<tr>
<td>Source</td>
<td>Lyondell Chemical Co. Houston, Texas</td>
</tr>
<tr>
<td>Date</td>
<td>09.05.2001</td>
</tr>
</tbody>
</table>

### 1.7.1 Technology Production/Use

### 1.8 Occupational Exposure Limit Values

### 1.9 Source of Exposure

### 1.10.1 Recommendations/Precautionary Measures

### 1.10.2 Emergency Measures

### 1.11 Packaging

### 1.12 Possible of Rendering Subst. Harmless

### 1.13 Statements Concerning Waste

### 1.14.1 Water Pollution

### 1.14.2 Major Accident Hazards

### 1.14.3 Air Pollution

### 1.15 Additional Remarks

### 1.16 Last Literature Search

### 1.17 Review

### 1.18 Listings E.G. Chemical Inventories
## 2. Physico-Chemical Data

<table>
<thead>
<tr>
<th>Section</th>
<th>Value</th>
<th>Source</th>
</tr>
</thead>
<tbody>
<tr>
<td><strong>2.1 MELTING POINT</strong></td>
<td>Value: &lt; -39 °C</td>
<td>Lyondell Chemical Co. Houston, Texas</td>
</tr>
<tr>
<td></td>
<td>Sublimation:</td>
<td></td>
</tr>
<tr>
<td></td>
<td>Method:</td>
<td></td>
</tr>
<tr>
<td></td>
<td>Year: 1980</td>
<td></td>
</tr>
<tr>
<td></td>
<td>GLP:</td>
<td></td>
</tr>
<tr>
<td></td>
<td>Test substance:</td>
<td></td>
</tr>
<tr>
<td></td>
<td>Source: Lyondell Chemical Co. Houston, Texas</td>
<td></td>
</tr>
<tr>
<td></td>
<td>Flag: Critical study for SIDS endpoint</td>
<td></td>
</tr>
<tr>
<td></td>
<td>Date: 22.05.2001</td>
<td></td>
</tr>
<tr>
<td><strong>2.2 BOILING POINT</strong></td>
<td>Value: = 222.2 - 233 °C at</td>
<td>Lyondell Chemical Co. Houston, Texas</td>
</tr>
<tr>
<td></td>
<td>Source: Lyondell Chemical Co. Houston, Texas</td>
<td></td>
</tr>
<tr>
<td></td>
<td>Date: 29.05.2001</td>
<td></td>
</tr>
<tr>
<td></td>
<td>Value: = 228 - 236 °C at</td>
<td></td>
</tr>
<tr>
<td></td>
<td>Source: Lyondell Chemical Co. Houston, Texas</td>
<td></td>
</tr>
<tr>
<td></td>
<td>Flag: Critical study for SIDS endpoint</td>
<td></td>
</tr>
<tr>
<td></td>
<td>Date: 29.05.2001</td>
<td></td>
</tr>
<tr>
<td><strong>2.3 DENSITY</strong></td>
<td>Type:</td>
<td></td>
</tr>
<tr>
<td></td>
<td>Value: = 1.025 g/cm3 at 20°C</td>
<td>Lyondell Chemical Co. Houston, Texas</td>
</tr>
<tr>
<td></td>
<td>Source: Lyondell Chemical Co. Houston, Texas</td>
<td></td>
</tr>
<tr>
<td></td>
<td>Date: 29.05.2001</td>
<td></td>
</tr>
<tr>
<td></td>
<td>Type:</td>
<td></td>
</tr>
<tr>
<td></td>
<td>Value: = 1.022 g/cm3 at 25°C</td>
<td>Lyondell Chemical Co. Houston, Texas</td>
</tr>
<tr>
<td></td>
<td>Source: Lyondell Chemical Co. Houston, Texas</td>
<td></td>
</tr>
<tr>
<td></td>
<td>Flag: Critical study for SIDS endpoint</td>
<td></td>
</tr>
<tr>
<td></td>
<td>Date: 29.05.2001</td>
<td></td>
</tr>
<tr>
<td><strong>2.3.1 GRANULOMETRY</strong></td>
<td></td>
<td></td>
</tr>
<tr>
<td><strong>2.4 VAPOUR PRESSURE</strong></td>
<td>Value: .05 hPa at 21°C</td>
<td>Lyondell Chemical Co. Houston, Texas</td>
</tr>
<tr>
<td></td>
<td>Source: Lyondell Chemical Co. Houston, Texas</td>
<td></td>
</tr>
<tr>
<td></td>
<td>Date: 29.05.2001</td>
<td></td>
</tr>
<tr>
<td></td>
<td>Value: .0128 hPa at 20°C</td>
<td>Lyondell Chemical Co. Houston, Texas</td>
</tr>
<tr>
<td></td>
<td>Source: Lyondell Chemical Co. Houston, Texas</td>
<td></td>
</tr>
<tr>
<td></td>
<td>Flag: Critical study for SIDS endpoint</td>
<td></td>
</tr>
<tr>
<td></td>
<td>Date: 29.05.2001</td>
<td></td>
</tr>
<tr>
<td><strong>2.5 PARTITION COEFFICIENT</strong></td>
<td>Log pow: = -1.486 at °C</td>
<td></td>
</tr>
</tbody>
</table>
### 2. Physico-Chemical Data

**Method**
- Year: 1993
- GLP:
- Test substance:
- Source: Lyondell Chemical Co. Houston, Texas
- Flag: Critical study for SIDS endpoint
- 23.05.2001

#### 2.6.1 WATER SOLUBILITY

**Value**
- at °C

**Qualitative**
- other: Described as miscible

**pKa**
- at 25 °C

**PH**
- at and °C

**Source**
- Lyondell Chemical Co. Houston, Texas

**Flag**
- Critical study for SIDS endpoint
- 29.05.2001

#### 2.6.2 SURFACE TENSION

**2.7 FLASH POINT**

**Value**
- = 118 °C

**Type**
- other

**Method**
- Year: 1980
- GLP:
- Test substance:
- Method: ASTM D56

**Source**
- Lyondell Chemical Co. Houston, Texas

**Flag**
- Critical study for SIDS endpoint
- 22.05.2001

#### 2.8 AUTO FLAMMABILITY

**Value**
- = 371 °C at

**Method**
- Year: 1993
- GLP:
- Test substance:
- Method: ASTM D56

**Source**
- Lyondell Chemical Co. Houston, Texas

**Flag**
- Critical study for SIDS endpoint
- 22.05.2001

#### 2.9 FLAMMABILITY

#### 2.10 EXPLOSIVE PROPERTIES

#### 2.11 OXIDIZING PROPERTIES

#### 2.12 ADDITIONAL REMARKS
### 3.1.1 PHOTODEGRADATION

### 3.1.2 STABILITY IN WATER

| Remark | no hydrolyzable groups |
| Source | Lyondell Chemical Co. Houston, Texas |
| Flag   | Critical study for SIDS endpoint |

22.05.2001

### 3.1.3 STABILITY IN SOIL

### 3.2 MONITORING DATA

#### 3.3.1 TRANSPORT BETWEEN ENVIRONMENTAL COMPARTMENTS

<table>
<thead>
<tr>
<th>Type</th>
<th>fugacity model level III</th>
</tr>
</thead>
<tbody>
<tr>
<td>Media</td>
<td></td>
</tr>
<tr>
<td>Air (level I)</td>
<td></td>
</tr>
<tr>
<td>Water (level I)</td>
<td></td>
</tr>
<tr>
<td>Soil (level I)</td>
<td></td>
</tr>
<tr>
<td>Biota (level II / III)</td>
<td></td>
</tr>
<tr>
<td>Soil (level II / III)</td>
<td></td>
</tr>
<tr>
<td>Method</td>
<td>other: calculated</td>
</tr>
<tr>
<td>Year</td>
<td></td>
</tr>
<tr>
<td>Method</td>
<td>EPIWIN (Estimation Program Interface for Windows) Version 3.05</td>
</tr>
<tr>
<td></td>
<td>LEVEL III FUGACITY MODEL</td>
</tr>
</tbody>
</table>

- Melting Point (deg C): -39
- Boiling Point (deg C): 222.20
- Vapor Pressure (mm Hg): 0.05
- Log Kow (octanol-water): -1.49

Default Emissions of 1000 Kg/h for air, water and soil (provided by EPIWIN)

<table>
<thead>
<tr>
<th>Result</th>
<th>Concentration (Percent)</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>Half Life (hours)</td>
</tr>
<tr>
<td></td>
<td>Emissions (Kg/hr)</td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>Medium</th>
<th>Concentration (%)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Air</td>
<td>0.107</td>
</tr>
<tr>
<td></td>
<td>8.19</td>
</tr>
<tr>
<td></td>
<td>1000</td>
</tr>
<tr>
<td>Water</td>
<td>46.1</td>
</tr>
<tr>
<td></td>
<td>360</td>
</tr>
<tr>
<td></td>
<td>1000</td>
</tr>
<tr>
<td>Soil</td>
<td>53.7</td>
</tr>
<tr>
<td></td>
<td>360</td>
</tr>
<tr>
<td></td>
<td>1000</td>
</tr>
</tbody>
</table>
3. Environmental Fate and Pathways

Sediment
0.0768
1.44e+003
0

Henry’s Law Constant: 3.58E-009 atm-m3/mole (EPIWIN estimate)
Volatization From Water (EPIWIN estimate)

River
Lake

Water depth (meters)
1
1

Wind Velocity (m/sec)
5
0.5

Current Velocity (m/sec)
1
0.05

HALF-LIFE (hours)
1.894E+005
2.067+006

HALF-LIFE (days)
7893
8.611E+004

HALF-LIFE (years)
21.61
235.8

Source: Lyondell Chemical Co. Houston, Texas

Conclusion: According to EPIWIN, assuming equal emissions to air, water, and soil, this chemical will concentrate mostly in water (46.1 %) and soil (53.7 %).

This chemical will take about 21.61 years to volatilize from a model river, and 235.8 years to volatilize from a model lake.

Flag: Critical study for SIDS endpoint

3.3.2 DISTRIBUTION

3.4 MODE OF DEGRADATION IN ACTUAL USE

3.5 BIODEGRADATION

Type: aerobic
Inoculum: activated sludge, adapted
Contact time:
Degradation: > 70 % after 28 day
Result:
### Deg. Product

### Year
1989

### GLP
no data

### Test substance
no data

### Method
Flasks containing test solution shaken continuously during the study period. Carbon dioxide production measured at intervals during the test. No additional study details given. Extent of pre-adaptation of bacteria not given.

### Remark
Original reference not available for review, so methods and results were described based on a reputable secondary source.

### Result
Following pre-adaptation of bacteria, significant degradation of the test substance was observed during the study.

### Source
Lyondell Chemical Co. Houston, Texas

### Conclusion
>70% degradation after 28 days

### Reliability
(2) valid with restrictions
Details on the test procedure and study results are not provided. Details on the results of analytical measurement of test substance concentrations in the test solutions and sample purity is not provided.

### Flag
Critical study for SIDS endpoint

### Type
aerobic

### Inoculum
activated sludge

### Deg. Product

### Method
OECD Guide-line 301 C “Ready Biodegradability: Modified MITI Test (I)”

### Year
1992

### GLP
no data

### Test substance
no data

### Method
Concentration: 30 mg/l substance and 100 mg/l sludge. Initial inoculation of 30 mg/l test substance to each flask. Respirometer used to evaluate test substance degradation. No agitation of exposure solutions during the study period. Carbon dioxide production measured continuously during the test. No additional study details given.

### Remark
Original reference not available for review, so methods and results were described based on a reputable secondary source.

### Result
no measurable degradation during the study period

### Source
Lyondell Chemical Co. Houston, Texas

### Conclusion
no degradation

### Reliability
(2) valid with restrictions
Details on the test procedure and study results are not provided. Details on the results of analytical measurement of test substance concentrations in the test solutions and sample purity is not provided.

### Flag
Critical study for SIDS endpoint

### 3.6 BOD5, COD OR BOD5/COD RATIO

#### BOD5
Method: other: APHA Standard Methods, Number 219

#### Year
1979

#### GLP
no

#### Concentration
related to BOD5 = 92 mgO2/l

#### COD
Method:

#### Year

#### GLP
### 3. Environmental Fate and Pathways

<table>
<thead>
<tr>
<th>COD</th>
<th>= 1840 mg/g substance</th>
</tr>
</thead>
<tbody>
<tr>
<td>RATIO BOD5 / COD</td>
<td>= 49</td>
</tr>
<tr>
<td>Method</td>
<td>Test conducted in special BOD bottles containing test substance and nutrient media. Oxygen depletion measured at initiation and after 5 days of incubation. Degradation of test substance calculated from oxygen depletion. No additional study details given.</td>
</tr>
</tbody>
</table>

Remarks: Because this study was conducted before any standardized guidelines were established, the question of guideline methodology and GLP conduct is not applicable.

<table>
<thead>
<tr>
<th>Result</th>
<th>BOD5 = 92 mg/l; COD = 1840 mg/g; BOD5/COD = 0.49</th>
</tr>
</thead>
<tbody>
<tr>
<td>Source</td>
<td>Lyondell Chemical Co. Houston, Texas</td>
</tr>
<tr>
<td>Test substance</td>
<td>no data</td>
</tr>
<tr>
<td>Conclusion</td>
<td>Low oxygen demand from test substance</td>
</tr>
<tr>
<td>Reliability</td>
<td>(2) valid with restrictions</td>
</tr>
</tbody>
</table>

Details on the test procedure and study results are not provided. Details on the results of analytical measurement of test substance concentrations in the test solutions and sample purity is not provided.

Flag: Critical study for SIDS endpoint

---

### 3.7 BIOACCUMULATION

<table>
<thead>
<tr>
<th>Species</th>
<th>Cyprinus carpio (Fish, fresh water)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Exposure period</td>
<td>42 day at 25 degree C</td>
</tr>
<tr>
<td>Concentration</td>
<td>3 mg/l</td>
</tr>
<tr>
<td>BCF</td>
<td>= 0.3 - 1.4</td>
</tr>
<tr>
<td>Elimination</td>
<td>no data</td>
</tr>
<tr>
<td>Method</td>
<td>OECD Guide-line 305 C &quot;Bioaccumulation: Test for the Degree of Bioconcentration in Fish&quot;</td>
</tr>
<tr>
<td>Year</td>
<td>1992</td>
</tr>
<tr>
<td>GLP</td>
<td>no data</td>
</tr>
<tr>
<td>Test substance</td>
<td>no data</td>
</tr>
<tr>
<td>Method</td>
<td>Fish exposed to test substance at concentration of 3 mg/l for exposure period of 42 days. Temperature of test solutions was maintained at 25 °C. No additional details given on the extent of the depuration period or other test conditions.</td>
</tr>
<tr>
<td>Remark</td>
<td>Original reference not available for review, so methods and results were described based on a reputable secondary source.</td>
</tr>
<tr>
<td>Source</td>
<td>Lyondell Chemical Co. Houston, Texas</td>
</tr>
<tr>
<td>Conclusion</td>
<td>The BCF for carp after 42 days of exposure was 0.3 to 1.4.</td>
</tr>
<tr>
<td>Reliability</td>
<td>(2) valid with restrictions</td>
</tr>
</tbody>
</table>

Details on the test procedure and study results are not provided. Details on the results of analytical measurement of test substance concentrations in the test solutions and sample purity is not provided.

Flag: Critical study for SIDS endpoint

---

### Species

<table>
<thead>
<tr>
<th>Species</th>
<th>Cyprinus carpio (Fish, fresh water)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Exposure period</td>
<td>42 day at 25 degree C</td>
</tr>
<tr>
<td>Concentration</td>
<td>3 mg/l</td>
</tr>
<tr>
<td>BCF</td>
<td>&lt; 2.2 - 4.6</td>
</tr>
<tr>
<td>Elimination</td>
<td>no data</td>
</tr>
<tr>
<td>Method</td>
<td>OECD Guide-line 305 C &quot;Bioaccumulation: Test for the Degree of Bioconcentration in Fish&quot;</td>
</tr>
<tr>
<td>Year</td>
<td>1992</td>
</tr>
<tr>
<td>GLP</td>
<td>no data</td>
</tr>
<tr>
<td>Test substance</td>
<td>no data</td>
</tr>
</tbody>
</table>

Details on the test procedure and study results are not provided. Details on the results of analytical measurement of test substance concentrations in the test solutions and sample purity is not provided.
### Method
Fish exposed to test substance at concentration of 3 mg/l for exposure period of 42 days. Temperature of test solutions was maintained at 25 °C. No additional details given on the extent of the depuration period or other test conditions.

### Remark
Original reference not available for review, so methods and results were described based on a reputable secondary source.

### Source
Lyondell Chemical Co., Houston, Texas

### Conclusion
The BCF for carp after 42 days of exposure was <2.2 to 4.6.

### Reliability
(2) valid with restrictions
Details on the test procedure and study results are not provided. Details on the results of analytical measurement of test substance concentrations in the test solutions and sample purity is not provided.

### Flag
Critical study for SIDS endpoint

---

#### 3.8 ADDITIONAL REMARKS
### 4.1 ACUTE/PROLONGED TOXICITY TO FISH

<table>
<thead>
<tr>
<th>Type</th>
<th>other: static acute LC50 (TLm)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Species</td>
<td>Carassius auratus (Fish, fresh water)</td>
</tr>
<tr>
<td>Exposure period</td>
<td>24 hour(s)</td>
</tr>
<tr>
<td>Unit</td>
<td>mg/l</td>
</tr>
<tr>
<td>Analytical monitoring</td>
<td>yes</td>
</tr>
<tr>
<td>LC50</td>
<td>&gt; 5000</td>
</tr>
<tr>
<td>Method</td>
<td></td>
</tr>
<tr>
<td>Year</td>
<td>1978</td>
</tr>
<tr>
<td>GLP</td>
<td>no</td>
</tr>
<tr>
<td>Test substance</td>
<td>no data</td>
</tr>
<tr>
<td>Method</td>
<td>APHA (American Public Health Association) 1971</td>
</tr>
</tbody>
</table>

Fish exposed to serial dilutions of TS with local tap water as the dilution water source.

Study conducted in 25 L all-glass tanks measuring 42 x 28 x 28 cm at 20 ± 1 °C.

Fish in the study had an average length of 6.2 ± 0.7 cm and an average weight of 3.3 ± 1.0 g.

Calculation of LC50 (TLm)
Interpolation from graph of logarithm of concentration vs. percentage mortality

Remarks: Because this study was conducted before any standardized guidelines were established, the question of guideline methodology and GLP conduct is not applicable.

Remark: Original reference not available for review, so methods and results were described based on a reputable secondary source.

Result: Since the 24-hour LC50 (TLm) was >5,000 mg/L, the study was terminated at this point.

Source: Lyondell Chemical Co. Houston, Texas

Conclusion: The 24-hour LC50 (TLm) for goldfish was determined to be >5,000 mg/L.

Reliability: (2) valid with restrictions

Details on the test procedure and study results are not provided. Details on the results of analytical measurement of test substance concentrations in the test solutions and sample purity is not provided.

Flag: Critical study for SIDS endpoint

### 4.2 ACUTE TOXICITY TO AQUATIC INVERTEBRATES

### 4.3 TOXICITY TO AQUATIC PLANTS E.G. ALGAE

### 4.4 TOXICITY TO MICROORGANISMS E.G. BACTERIA
OECD SIDS                           PROPYLENE GLYCOL
4. Ecotoxicity

Id 25265-71-8
Date 30.05.2001

Method :  
Year : 1992
GLP : yes
Test substance : as prescribed by 1.1 - 1.4
Method : German Water Hazard Classification (Umweltbundesamt, Sept. 1979, LTwS-Nr.10)

There were 3 flasks per exposure concentration and 10 flasks for control.

Bacteria exposed to serial dilutions of TS for 18 ± 2 hours with concentrations of 1.95, 3.91, 7.81, 15.63, 31.25, 62.5, 125, 250, 500 and 1000 mg/L.

Solutions evaluated with infrared spectroscopy (IR). Extinction of cultures measured at 436 nm.

Result : Based on measurements made in all exposure concentrations, the EC10 was 1,000 mg/L.
Source : Lyondell Chemical Co. Houston, Texas
Test substance : sample >99.9% purity
Conclusion : The 18-hour EC10 for Pseudomonas putida was determined to be 1,000 mg/L.
Reliability : (1) valid without restriction
Flag : Critical study for SIDS endpoint

4.5.1 CHRONIC TOXICITY TO FISH

4.5.2 CHRONIC TOXICITY TO AQUATIC INVERTEBRATES

4.6.1 TOXICITY TO SOIL DWELLING ORGANISMS

4.6.2 TOXICITY TO TERRESTRIAL PLANTS

4.6.3 TOXICITY TO OTHER NON-MAMM. TERRESTRIAL SPECIES

4.7 BIOLOGICAL EFFECTS MONITORING

4.8 BIOTRANSFORMATION AND KINETICS

4.9 ADDITIONAL REMARKS

Memo : Toxicity to clawed toad (Xenopus laevis)
Method : Type: static acute LC50
Species: clawed toad (Xenopus laevis)
Number of animals: 10 per exposure concentration
Value: = 3.181 mg/L
Year: 1987
GLP: No
Test substance: no compositional details given

Exposure Period: 48 hrs

Method: No methodology given.

Clawed toads exposed to serial dilutions of test substance with Dutch standard water as the dilution water source.

Study conducted in 1 L all-glass covered aquaria with 3-4 week old larvae at a study temperature of 20 ± 1 ºC.

Calculation of LC50
Calculated as projection from least square linear regression on log transformed nominal concentration data and probit transformed percent effect data.

Analytical monitoring: None

Result: Mortality measured after 48 hours of exposure resulted in determination of an LC50 of 3,181 mg/L.

Source: Lyondell Chemical Co. Houston, Texas

Conclusion: The 48-hour LC50 (TLm) for the clawed toad was determined to be 3,181 mg/L.

Reliability: (2) valid with restrictions
Details on the test procedure and study results are not provided. No analytical measurement of test substance concentrations in the test solutions or sample purity were conducted.

Flag: Critical study for SIDS endpoint

23.05.2001
5.1.1 ACUTE ORAL TOXICITY

Type : LD50  
Species : rat  
Strain : Wistar  
Sex : male/female  
Number of animals : 50  
Vehicle : other: none  
Value : = 16000 mg/kg bw  
Method :  
Year : 1980  
GLP : no  
Test substance : no data  
Method : Guideline not specified.  

Animals and treatments  
Rats were selected with a body weight range of 182 to 283 g for males and 110 to 150 g for females. Dipropylene glycol was administered by stomach tube following an overnight fast to 5 males and 5 females per dose at doses of 8.6, 10.4, 12.4, 14.0 or 17.9 ml per kg of body weight. The animals were observed for signs of intoxication for 14 days following dosing.

Postmortem examination  
All animals were necropsied after death or 14 days after dosing. There was no indication of histopathologic examination of any tissues.

Calculation of LD50  
Based on method of Weil et al, Biometrics (195X)=missing year; 3: 247-263.

General signs of toxicity were not reported. Examination of the internal organs revealed no gross alterations.

The LD50 in the rat was determined to be 15.8 ml/kg, with standard errors of 19.2 - 23.0 ml/kg (equivalent to 16000 mg/kg, standard error 15000-17500 mg/kg).(Reviewer)

Remarks: Because this study was conducted before any standardized guidelines were established, the question of guideline methodology and GLP conduct is not applicable.

Source : Lyondell Chemical Co. Houston, Texas  
Conclusion : Based on the results of this study, an LD50 of 16000 mg/kg was determined for dipropylene glycol in the rat.  
Reliability : (2) valid with restrictions  
Test material characterization not documented; GLP compliance not indicated.

Flag : Critical study for SIDS endpoint  
29.05.2001

Type : LD50  
Species : mouse  
Strain : CD-1  
Sex : male  
Number of animals : 24  
Vehicle : water  
Value : > 2000 mg/kg bw  
Method :  
Year : 1999
### 5. Acute Inhalation Toxicity

<table>
<thead>
<tr>
<th>Type</th>
<th>LC50</th>
</tr>
</thead>
<tbody>
<tr>
<td>Species</td>
<td>other: Rat &amp; Guinea Pig</td>
</tr>
<tr>
<td>Strain</td>
<td>other: not specified</td>
</tr>
<tr>
<td>Sex</td>
<td>no data</td>
</tr>
<tr>
<td>Number of animals</td>
<td></td>
</tr>
<tr>
<td>Vehicle</td>
<td></td>
</tr>
<tr>
<td>Exposure time</td>
<td></td>
</tr>
<tr>
<td>Method</td>
<td></td>
</tr>
<tr>
<td>Year</td>
<td>1961</td>
</tr>
<tr>
<td>GLP</td>
<td>no</td>
</tr>
<tr>
<td>Test substance</td>
<td></td>
</tr>
<tr>
<td>Method</td>
<td>Dose Level-Vapor concentrations not reported, aerosol nominal concentrations were 6000-8000 mg/m3.</td>
</tr>
</tbody>
</table>

Dipropylene glycol was heated to 170°C for 8 hours, 6 rats individually exposed to vapors in small chambers.

Dipropylene glycol was heated to 120°C for 4 hours, 6 rats individually exposed to vapors in small chambers.

Dipropylene glycol was heated to 170°C for 5 hours with the vapors being drawn off, allowed to cool; then reheated to 120°C for 5 hours, 6 rats individually exposed to vapors in small chambers.

Dipropylene glycol was heated to 170°C for 5 hours with the vapors being drawn off, allowed to cool; 6 rats were individually exposed in small chambers to an aerosol generated at room temperature.

An aerosol of dipropylene glycol was generated at room temperature, 6 rats and 6 guinea pigs were individually exposed in small chambers. (duration not reported)

Remarks: Because this study was conducted before any standardized guidelines were established, the question of guideline methodology and GLP conduct is not applicable.
Result: Five of 6 rats exposed to vapors generated at 170°C died; there were no pathologic abnormalities.

Vapors generated at 120°C were not lethal.

When rats were exposed to vapors generated at 120°C after removal of vapors generated by heating dipropylene glycol to 170°C for 5 hours, there were no deaths.

There was no lethality to rats or guinea pigs from aerosol of dipropylene glycol generated at room temperature.

Source: Lyondell Chemical Co. Houston, Texas

Conclusion: Heating Dipropylene Glycol to 170°C resulted in toxic degradation products that did not occur from heating to 120°C and were not present in aerosols generated at room temperature or at 120°C. (Reviewer)

Reliability: (4) not assignable

Flag: Critical study for SIDS endpoint

5.1.3 ACUTE DERMAL TOXICITY

Type: LD50
Species: rabbit
Strain:
Sex: no data
Number of animals:
Vehicle: other: none
Value: > 5000 mg/kg bw
Remark: Original reference not available for review, so methods and results were described based on a reputable secondary source.

Result: No deaths; LD50 >5000 mg/kg

Source: Lyondell Chemical Co. Houston, Texas

Reliability: (4) not assignable
details not provided in secondary source

Flag: Critical study for SIDS endpoint

5.1.4 ACUTE TOXICITY, OTHER ROUTES

5.2.1 SKIN IRRITATION

Species: rabbit
Concentration: other: undiluted
Exposure:
Exposure time: 24 hour(s)
Number of animals: 6
PDII:
Result: slightly irritating
EC classification:
Method: other: Not stated
Year: 1974
GLP: no
Test substance: no data
Method: Limited experimental details are available for this study.
500 mg undiluted dipropylene glycol was applied to intact and abraded skin of rabbits and left occluded for 24 hours. Reactions were recorded at 24, 48, and 72 hr after application, using the scoring system of Draize et al (1944, J Pharmac. Exp Ther, 82, 377 - 390).

Remarks: Because this study was conducted before any standardized guidelines were established, the question of guideline methodology and GLP conduct is not applicable.

Result:  The results from this study demonstrate that undiluted dipropylene glycol causes mild irritation of the rabbit skin. The response is insufficient to trigger classification of dipropylene glycol as an skin irritant. (Reviewer)

Reliability:  (4) not assignable

Flag:  Critical study for SIDS endpoint

Species:  human
Concentration:  25 %
Exposure:  : 
Exposure time:  24 hour(s)
Number of animals:  33
PDII:  : 
Result:  : 
EC classification:  : 
Method:  : 
Year:  1995
GLP:  : 
Test substance:  : 
Method:  : 

Limited experimental details are available for this study.
0.2 ml of a 25% solution of dipropylene glycol in distilled water for 24 hours via semi-occluded patches. Reactions were recorded at 4 and 24 hr after application.

Result:  2 subjects demonstrated no effects at 30 min. but mild erythema at 24 hrs. 2 subjects demonstrated mild to moderate erythema at 30 min. and mild erythema at 24 hrs. 7 subjects demonstrated mild to moderate erythema at 30 min. but no effects at 24 hrs. 22 subjects no reaction at 30 min or 24 hr.

Source:  Lyondell Chemical Co. Houston, Texas

Conclusion:  The results from this study demonstrated that 25% dipropylene glycol causes none to mild irritation of human skin.

Reliability:  (2) valid with restrictions

Flag:  Critical study for SIDS endpoint

Species:  rabbit
Concentration:  undiluted
Dose:  .1 ml
Exposure Time:  4 hour(s)
Comment:  other: Tested as 100% and formulation containing 7.2% dipropylene glycol
5. Toxicity

<p>| | |</p>
<table>
<thead>
<tr>
<th></th>
<th></th>
</tr>
</thead>
<tbody>
<tr>
<td>Number of animals</td>
<td>:</td>
</tr>
<tr>
<td>Result</td>
<td>: slightly irritating</td>
</tr>
<tr>
<td>EC classification</td>
<td>:</td>
</tr>
<tr>
<td>Method</td>
<td>: Draize Test</td>
</tr>
<tr>
<td>Year</td>
<td>: 1969</td>
</tr>
<tr>
<td>GLP</td>
<td>: no</td>
</tr>
<tr>
<td>Test substance</td>
<td>: no data</td>
</tr>
<tr>
<td>Method</td>
<td>: Animals and treatments</td>
</tr>
<tr>
<td></td>
<td>Limited experimental details are available for this study, which followed Draize method.</td>
</tr>
<tr>
<td></td>
<td>0.1 ml undiluted dipropylene glycol was applied to the lower conjunctival sac of rabbits. Reactions were recorded at 4, 24, 48, 72 and 96 hr post-instillation, using the scoring system of Draize et al (1944, J Pharmac. Exp Ther, 82, 377 - 390).</td>
</tr>
<tr>
<td>Remarks:</td>
<td>Because this study was conducted before any standardized guidelines were established, the question of guideline methodology and GLP conduct is not applicable.</td>
</tr>
<tr>
<td>Original reference</td>
<td>: Original reference not available for review, so methods and results were described based on a reputable secondary source.</td>
</tr>
<tr>
<td>Result</td>
<td>: Detailed results were not presented.</td>
</tr>
<tr>
<td>Source</td>
<td>: Lyondell Chemical Co. Houston, Texas</td>
</tr>
<tr>
<td>Conclusion</td>
<td>: The results from this study demonstrate that undiluted dipropylene glycol causes mild, fully reversible irritation of the rabbit eye. The response is insufficient to trigger classification of dipropylene glycol as an eye irritant. (Reviewer)</td>
</tr>
<tr>
<td>Reliability</td>
<td>: (4) not assignable</td>
</tr>
<tr>
<td>Flag</td>
<td>: Critical study for SIDS endpoint</td>
</tr>
</tbody>
</table>

5.3 SENSITIZATION

<p>| | |</p>
<table>
<thead>
<tr>
<th></th>
<th></th>
</tr>
</thead>
<tbody>
<tr>
<td>Type</td>
<td>: Patch-Test</td>
</tr>
<tr>
<td>Species</td>
<td>: human</td>
</tr>
<tr>
<td>Number of animals</td>
<td>: 503</td>
</tr>
<tr>
<td>Vehicle</td>
<td>: water</td>
</tr>
<tr>
<td>Result</td>
<td>:</td>
</tr>
<tr>
<td>Classification</td>
<td>: not sensitizing</td>
</tr>
<tr>
<td>Method</td>
<td>: other: repeated insult patch test</td>
</tr>
<tr>
<td>Year</td>
<td>: 1995</td>
</tr>
<tr>
<td>GLP</td>
<td>: no</td>
</tr>
<tr>
<td>Test substance</td>
<td>: as prescribed by 1.1 - 1.4</td>
</tr>
<tr>
<td>Method</td>
<td>: Subjects</td>
</tr>
<tr>
<td></td>
<td>503 consecutive patients (212 men, 291 women) who entered the dermatology clinic with eczema were challenged with both cosmetic and synthesis grade dipropylene glycol.</td>
</tr>
<tr>
<td>Challenge phase</td>
<td>Patients received a challenge patch of 1, 2, 5, or 10% dipropylene glycol in water using a Finn chamber and Scanpor tape. The patch was left on the skin for 2 days. Reaction was graded after 2, 3, and 5-7 days.</td>
</tr>
<tr>
<td>Scoring system</td>
<td>A four point scale was used to record skin responses during induction and</td>
</tr>
</tbody>
</table>
5. Toxicity

challenge: -, TR, ?, +
(Definition of -, TR, ?, and + were not given in the report)

Statistics
No statistical methods were applied to the data.

Result: One patient was sensitized to dipropylene glycol

<table>
<thead>
<tr>
<th>Score</th>
<th>Number of responses</th>
</tr>
</thead>
<tbody>
<tr>
<td>-</td>
<td>cosmetic DPG = 488 patients; synthesis DPG = 480 patients</td>
</tr>
<tr>
<td>TR</td>
<td>cosmetic DPG = 2 patients; synthesis DPG = 5 patients</td>
</tr>
<tr>
<td>?+</td>
<td>cosmetic DPG = 13 patients; synthesis DPG = 17 patients</td>
</tr>
<tr>
<td>+</td>
<td>cosmetic DPG = 0 patients; synthesis DPG = 1 patient</td>
</tr>
</tbody>
</table>

Source: Lyondell Chemical Co. Houston, Texas
Test substance: >97% pure from E.Merck (synthesis grade) and cosmetic grade (>96%)
Conclusion: The authors concluded that sensitization to dipropylene glycol was rare.
Reliability: (2) valid with restrictions
Flag: not standard guideline test

5.4 REPEATED DOSE TOXICITY

Species: rat
Sex: no data
Strain: no data
Route of admin.: drinking water
Exposure period: 9 to 77 days
Frequency of treatment: continuous
Post obs. period: none
Doses: 1 to 10% dipropylene glycol in drinking water
Control group: no
NOAEL: = 5%
LOAEL: = 10%
Method:
Year: 1939
GLP: no
Test substance: no data
Method: Animals
32 rats; age, weight, sex and strain not specified.

Exposure
Dipropylene glycol was mixed in the drinking water of 7 rats at concentrations from 1 to 5% for 33 to 77 days. Dipropylene glycol was mixed in the drinking water of 25 rats at a concentration of 10% for 9 to 68 days. Further details of exact doses, number of animals at each dose level, or duration for individual animals not provided.

Terminal observations
Kidneys were examined microscopically for pathology. Report indicated that the liver and kidney were examined. No indication that other tissues were examined.

Statistical methods
None

Remarks: Because this study was conducted before any standardized guidelines were established, the question of guideline methodology and
GLP conduct is not applicable.

**Result**

Seven of 25 rats exposed to 10% dipropylene glycol in the drinking water died between days 10 and 30. Five had kidney lesions. Four of the 18 survivors had kidney lesions. The kidney lesions were described as hydropic degeneration of the tubular epithelium. No effects on mortality or the kidney and liver were seen in rats exposed to 5% dipropylene glycol in their drinking water.

**Source**

Lyondell Chemical Co. Houston, Texas

**Reliability**

(2) valid with restrictions

Not guideline, pre-GLP, methods and results only briefly described but generally acceptable.

**Flag**

Critical study for SIDS endpoint

---

**Species**

dog

**Sex**

no data

**Strain**

no data

**Route of admin.**
gavage

**Exposure period**

not specified

**Frequency of treatment**

daily.

**Post obs. period**

None

**Doses**

9 cc/kg as 1.5 cc/kg administered for 6 doses, 12 cc/kg as 2.0 cc/kg administered for 6 doses, or 20 cc/kg as 5.0 cc/kg administered for 4 doses

**Control group**

no

**NOAEL**

= 20 ml/kg bw

**Method**

Animals

4 dogs; age, weight, sex and strain not specified

General

Divided gavage doses given as follows: 9 cc/kg as 1.5 cc/kg administered for 6 doses, 12 cc/kg as 2.0 cc/kg administered for 6 doses, or 20 cc/kg as 5.0 cc/kg administered for 4 doses.

Details on evaluations conducted were not provided in the report.

Remarks: Because this study was conducted before any standardized guidelines were established, the question of guideline methodology and GLP conduct is not applicable.

**Result**

1 dog dosed at 9 cc/kg - no evidence of toxicity
1 dog dosed at 12 cc/kg - no evidence of toxicity
1 dog dosed at 12 cc/kg - emesis; recovery
1 dog dosed at 20 cc/kg - no evidence of toxicity

**Source**

Lyondell Chemical Co. Houston, Texas

**Conclusion**

No evidence of systemic toxicity was detected under the conditions of this study.

**Reliability**

(2) valid with restrictions

Not guideline, pre-GLP, methods and results only briefly described but generally acceptable.

**Flag**

Critical study for SIDS endpoint
### 5. Toxicity

<table>
<thead>
<tr>
<th>Exposure period</th>
<th>15 weeks</th>
</tr>
</thead>
<tbody>
<tr>
<td>Frequency of treatment</td>
<td>daily</td>
</tr>
<tr>
<td>Post obs. period</td>
<td>none</td>
</tr>
<tr>
<td>Doses</td>
<td>0, 12% of diet</td>
</tr>
<tr>
<td>Control group</td>
<td>yes, concurrent vehicle</td>
</tr>
<tr>
<td>NOAEL</td>
<td>= 12 %</td>
</tr>
<tr>
<td>Method</td>
<td>other: investigative study</td>
</tr>
<tr>
<td>Year</td>
<td>1940</td>
</tr>
<tr>
<td>GLP</td>
<td>no</td>
</tr>
<tr>
<td>Test substance</td>
<td>no data</td>
</tr>
<tr>
<td>Method</td>
<td>Limited study details available</td>
</tr>
</tbody>
</table>

Dipropylene glycol mixed in diet at 12% and fed ad libitum.

Observations
The rats were observed for running activity after 15 weeks. Details on assessment of running activity not provided.

Remarks: Because this study was conducted before any standardized guidelines were established, the question of guideline methodology and GLP conduct is not applicable.

Result: Running activity was decreased in rats fed 12% dipropylene glycol compared to control rats. The magnitude of the changes in running activity were not given in the report.

Source: Lyondell Chemical Co. Houston, Texas

Reliability: (2) valid with restrictions
Not guideline, pre-GLP, methods and results only briefly described but generally acceptable.

Flag: Critical study for SIDS endpoint

#### 5.5 GENETIC TOXICITY 'IN VITRO'

| Type | Ames test |
| System of testing | Salmonella typhimurium TA98, TA100, TA1535, TA1537 |
| Concentration | up to 10 mg/plate (lower concentrations not specified) |
| Cycotoxic conc. | >10 mg/plate |
| Metabolic activation | with and without |
| Result | negative |
| Method | other: Ames's standard plate assay |
| Year | 1984 |
| GLP | no data |
| Test substance | no data |
| Method | Study details not available. |
| Result | No increase in revertants was recorded for any of the strains exposed to dipropylene glycol with or without metabolic activation. |
| Source | Lyondell Chemical Co. Houston, Texas |
| Conclusion | Dipropylene glycol was not mutagenic in Salmonella typhimurium TA98, TA100, TA1535 and TA 1537, in the presence or absence of an S-9 fraction, under the conditions of this test. |
| Reliability | (2) valid with restrictions summary data only available from NCI program. |
| Flag | Critical study for SIDS endpoint |

Type: Mouse lymphoma assay
System of testing: L5178Y (TK+/TK-)

29.05.2001
5. Toxicity

**Concentration**: 29-60 ul/ml without activation; 30 - 50 ul/ml with activation

**Cytotoxic conc.**: not reported

**Metabolic activation**: with and without

**Result**: negative

**Method**: other: Suspension/plate

**Year**: 1987

**GLP**: no data

**Test substance**: no data

**Method**: Limited details reported.

**Result**: No increase in revertants was recorded for dipropylene glycol with or without activation.

**Source**: Lyondell Chemical Co. Houston, Texas

**Reliability**: (2) valid with restrictions

summary data only available from NCI program.

**Flag**: Critical study for SIDS endpoint

29.05.2001

5.6 GENETIC TOXICITY 'IN VIVO'

**Type**: Micronucleus assay

**Species**: mouse

**Sex**: male

**Strain**: other: CD-1(ICR)BR

**Route of admin.**: gavage

**Exposure period**: 2 days

**Doses**: 500, 1000 and 2000 mg/kg

**Result**: negative

**Method**: other: US EPA OPPTS 870.5395 and OECD 474

**Year**: 1999

**GLP**: yes

**Test substance**: as prescribed by 1.1 - 1.4

Animals and treatments
Eight week old male CD-1 mice from Charles River Labs were used in these investigations. Six mice were used per treatment group (control, positive control, 3 levels of dipropylene glycol). The dose levels were based on a preliminary study. Dipropylene glycol was administered for two consecutive days by gavage at 500, 1000 or 2000 mg/kg/day in water at a dose volume of 10 ml/kg. 120 mg/kg/day cyclophosphamide was administered on two consecutive days by gavage at a dose volume of 10 ml/kg.

Preparation and examination of bone marrow smears
Mice were killed by carbon dioxide inhalation 24 hr after second gavage dose of dipropylene glycol. Femoral marrow cells were isolated, smeared onto clean glass slides, fixed with methanol and stained with Wright-Giemsa. The preparations were coded and analyzed without identification of animal number or treatment. Two thousand polychromatic erythrocytes (PCEs) per mouse were examined using light microscopy (x100), and the number of micronucleated polychromatic erythrocytes (MNPCEs) was recorded. The proportion of PCEs among the total erythrocytes was also evaluated by observation of 200 erythrocytes on the same slide.

Positive control substance
120 mg/kg/day cyclophosphamide administered two consecutive days by gavage.

Statistical analysis
The frequency of MNPCEs in each treatment was analyzed by one-way
analysis of variance. A result was considered positive if the increase in MNPCEs differed from the spontaneous data at $P<0.01$. Any dose-response relationship was tested using the Cochran-Armitage trend test, with $P<0.01$ indicating a positive result.

**Result**

No mice died during the course of the study. There was no statistically significant increase or trend in MNPCE numbers following gavage administration of dipropylene glycol at doses up to and including 2000 mg/kg. The percentage of PCEs in the dipropylene glycol-treated groups were not significantly different from controls. A significant, dose-related increase in MNPCEs was obtained with the positive control, cyclophosphamide, indicating that the test system was capable of detecting a positive response.

<table>
<thead>
<tr>
<th>Dose</th>
<th>MN-PCE/2000 PCE</th>
<th>%PCE</th>
</tr>
</thead>
<tbody>
<tr>
<td>0</td>
<td>2.8±2.5</td>
<td>60.7±4.7</td>
</tr>
<tr>
<td>500</td>
<td>2.8±1.9</td>
<td>57.6±8.7</td>
</tr>
<tr>
<td>1000</td>
<td>1.3±0.5</td>
<td>58.7±4.5</td>
</tr>
<tr>
<td>2000</td>
<td>1.5±1.4</td>
<td>52.0±8.2</td>
</tr>
<tr>
<td>cyclophosphamide</td>
<td>40.0±16.9</td>
<td>50.2±6.7</td>
</tr>
</tbody>
</table>

**Source**

Lyondell Chemical Co. Houston, Texas

**Conclusion**

Dipropylene glycol produced no detectable increase in micronucleated polychromatic erythrocytes when administered by gavage to mice at doses up to 2000 mg/kg.

**Reliability**

(1) valid without restriction
meets all criteria: follows guidelines, GLP.

**Flag**

Critical study for SIDS endpoint

---

### 5.7 CARCINOGENITY

### 5.8 TOXICITY TO REPRODUCTION

### 5.9 DEVELOPMENTAL TOXICITY/TERATOGENICITY

**Species**

rat

**Sex**

female

**Strain**

other: Crl: CD BR VAF/Plus

**Route of admin.**

gavage

**Exposure period**

GD 6 - 15

**Frequency of treatment**

daily

**Duration of test**

20 days

**Doses**

800, 2000, and 5000 mg/kg / day in water at dose volume of 5 ml/kg

**Control group**

other: Yes; treated with water

**NOAEL Maternal**

= 800 mg/kg bw

**NOAEL Teratogen**

= 5000 mg/kg bw

**Method**

other: NTP

**Year**

1992

**GLP**

yes

**Test substance**

as prescribed by 1.1 - 1.4

**Method**

Animals and treatment

Mated and presumed pregnant female rats were allocated to a sham control group ($n=27$) or one of three treatment groups (800, 2000, and 5000 mg / kg/day, $n=27, 26$, and 26 respectively) at the start of the study. Treatment commenced on GD6 and continued to GD15.
Maternal observations
All animals were observed daily for appearance and behavior. Body
weights were recorded on GD0, 3, 6-15, 18 and 20. Food and water
consumption data were collected during the periods of gestation days 0-3,
3-6, 6-9, 9-12, 12-15, 15-18, and 18-20.

Fetal examination
On GD20 all dams were subjected to Caesarian section under anesthesia,
and the numbers of implantation sites, resorption sites and live and dead
fetuses recorded. The body weights of the live pups were also recorded. All
fetuses were examined for visceral (Staples technique) and skeletal
abnormalities (Alcian Blue/Alizarin Red S stain).

Statistical analysis
Standard ANOVA for parametric data and Fisher's exact for nominal scale
data.

Result:
Maternal parameters
One of 25 pregnant dams treated at 2000 mg/kg and 2 of 24 at 5000
mg/kg died on GD14. All other dams survived. Signs of toxicity at 2000
and 5000 mg/kg/day included ataxia, unstable gait, and piloerection.
Significantly reduced body weight gain, reduced food consumption and
increased water consumption were seen at 5000 mg/kg/day.

At necropsy on GD 20, dams treated at 2000 and 5000 mg/kg/day had
increased relative liver weights compared to controls.

Pregnancy parameters
Treatment with dipropylene glycol was without effect on the number of live
litters, the total or average number of implant sites, total and partial
resorptions, the total and average number of live fetuses and their sex
ratio, the number of dead fetuses or fetal weight.

Fetal parameters
No increase in visceral or skeletal abnormalities or variations occurred with
exposure to dipropylene glycol.

Source
Lyondell Chemical Co. Houston, Texas
Test substance
Identity and purity verified by chemical analysis: >96% pure
Conclusion
Under the conditions of the study, there was no adverse effect on
pregnancy parameters or malformations at exposures of dipropylene glycol
up to 5000 mg/kg/day. Dipropylene glycol induced maternal toxicity
occurred at 2000 and 5000 mg/kg as evidenced by deaths, signs of CNS
depression, reduced feed consumption and body weight gain, and
increased relative liver weights.

Reliability
(1) valid without restriction
meets guideline and GLP.
Flag
Critical study for SIDS endpoint
(4)
Species: rabbit
Sex: female
Strain: New Zealand white
Route of admin.: gavage
Exposure period: GD 6 - 19
Frequency of
Treatment: daily
Duration of test: 30 days
Doses: 200, 400, 800, 1200 mg / kg / day in water at dosing volume of 2 ml/kg.
Control group: other: Yes, treated with water
NOAEL Maternalt.: = 1200 mg/kg bw
NOAEL Teratogen: = 1200 mg/kg bw
### Method
- **Method**: other: NTP
- **Year**: 1992
- **GLP**: yes
- **Test substance**: as prescribed by 1.1 - 1.4
- **Animals and treatment**
  
  Rabbits were artificially inseminated with sperm from proven male breeders. Inseminated female rabbits (n = 24/group) were allocated to a sham control group and four treatment groups (200, 400, 800 and 1200 mg/kg/day) at the start of the study. Treatment commenced on GD6 and continued to GD19.

### Maternal observations
All animals were observed daily for appearance and behavior. Body weights were recorded on GD0, 6-19, 25, and 30. Food weights were recorded on gestations days 0, 6, 9, 12, 15, 18, 22, 25, 28, and 30. Maternal livers and kidneys were weighed at necropsy.

### Fetal examination
On GD30 all dams were subjected to Caesarian section under anesthesia, and the numbers of implantation sites, resorption sites and live and dead fetuses recorded. The body weights of the live pups were also recorded. All fetuses were examined for external abnormalities. All fetuses were examined for visceral (Staples technique) and skeletal abnormalities (Alcian Blue/Alizarin Red S stain).

### Statistical analysis
Standard ANOVA for parametric data and Fisher's exact for nominal scale data.

### Result
- **Maternal parameters**
  
  There was no evidence of maternal toxicity in this study based on signs of toxicity, body weight, food consumption, organ weights or gross necropsy. The high dose (1200 mg/kg/day) was in the lethal range for dipropylene glycol in pregnant NZW rabbits; in a preliminary study 22 and 38% maternal mortality was found from gavage exposures GD6-19 at 800 and 1500 mg/kg/day, respectively.

- **Pregnancy parameters**
  
  Treatment with dipropylene glycol was without effect on the number of live litters, the total or average number of implant sites, total and partial resorptions, the total and average number of live fetuses and their sex ratio, the number of dead fetuses or fetal weight.

- **Fetal parameters**
  
  There was no increase in visceral or skeletal defects attributed to treatment with dipropylene glycol.

### Source
- **Test substance**: Lyondell Chemical Co. Houston, Texas
- **Analytical data verified identity**: 97 to 99% pure.

### Conclusion
Under the conditions of the study, there was no adverse effect on pregnancy parameters or maternal or fetal survival after exposure to up to 1200 mg/kg/day dipropylene glycol.

### Reliability
- **Reliability**: (1) valid without restriction
  
  meets guideline and GLP.

### Flag
- **Flag**: Critical study for SIDS endpoint
  
  (3)

### 5.10 OTHER RELEVANT INFORMATION

- **Type**: Metabolism
### 5. Toxicity

**Method**
- **Species:** rat
- **Sex:** male
- **Strain:** F344
- **Number of animals:** 10
- **Route of admin.:** gavage
- **Exposure period:** one dose
- **Frequency of treatment:** once
- **Duration of test:** 24 hrs
- **Doses:** 40 mg/kg/day TPG in water at dosing volume of 4 ml/kg.
- **Control group:** No
- **Year:** 1995
- **GLP:** yes

**Animals and treatment**

Male F344 rats weighing between 187 and 197 g were obtained from Charles River and allowed 2 days for acclimation in metabolism cages before dosing. After dosing, the animals were held in the metabolism cages for 24 hours.

**Sample collections**

Air drawn through the metabolism chambers was passed through a trap for collection of radioactive CO2. All voided urine was collected and frozen for analysis. Feces were collected in dry-ice chilled containers for analysis. After 24 hours the animals were sacrificed for collection of blood, liver, kidney, fat, brain, muscle, and the remaining carcass.

**Sample analysis**

Radioactivity in all collected samples was determined using a liquid scintillation counter. Urine samples were analyzed by gas chromatography for tripropylene glycol, dipropylene glycol and propylene glycol following derivatization to pentafluorobenzoyl esters.

**Source**

Lyondell Chemical Co. Houston, Texas

**Test substance**

Analytical data verified identity: 99.8% tripropylene glycol, 0.08% dipropylene glycol pure. 14C-labelled test material TM = 14C-labelled TPG.

**Conclusion**

Result: 91.4±2.1% of the administered dose was recovered from CO2, urine, feces, tissues and carcass. 20.7±0.6% was recovered as CO2; 52.3±3.5% was recovered in urine. Approximately 10% of the administered radioactivity remained in the carcass after 24 hours.

Urine analysis

13% of the administered dose of tripropylene glycol was recovered in the urine as free or conjugated tripropylene glycol, 8.4% as free or conjugated dipropylene glycol, and 3.9% as free or conjugated propylene glycol.

Administration of propylene glycol resulted in 12% recovered as propylene glycol in the urine.

Conclusion: Tripropylene glycol is rapidly and extensively metabolized to dipropylene glycol which is further metabolized to propylene glycol and subsequently to lactic or pyruvic acids followed by oxidation through the citric acid cycle (21% to CO2) or production of glycogen. Much of the radioactivity remaining in the carcass probably represented metabolically incorporated carbons atoms derived from propylene glycol. (author conclusion)

**Reliability**

(1) valid without restriction

meets guidelines and GLP.

**Flag**

Critical study for SIDS endpoint (18)
Type: other; Carcinogenicity

Species: rat

Sex: male/female

Strain: Charles River, CD strain

Route of admin.: oral feed

Exposure period: 104 wk

Frequency of treatment: daily

Post. obs. period: none

Doses: 6250, 12500, 25000 or 50000 ppm

Result: negative

Control group: yes, concurrent vehicle

Year: 1972

GLP: No

Method: Animals

30 male (bw 120 - 150g) and 30 female (bw 120-140g) rats

General

Individual body weights were recorded at 2-wk intervals, with food intake measured over the preceding 24 hr.

Hematology

Blood was collected (tail vein) from 8 male and 8 female rats fed diets containing 0, 25000 or 50000 ppm propylene glycol at wk 13, 21, 52 and 80. Additional samples were collected from 6250 and 12500 ppm groups at wk 54. Samples were analysed for Hb content, PCV and counts of erythrocytes, and total and differential leucocytes. Reticulocyte counts were determined at wk 52, 54 and 80. Terminal observations (wk 104) were limited to Hb concentration and microscopic examination of a stained smear.

Renal function

A urinary concentration test was conducted on 6 - 10 rats from the control, 25000 and 50000 ppm groups. Measurements included specific gravity, urine volume under different water loading conditions and a urinary cell count.

Terminal observations

Surviving animals were killed at wk 104 (exsanguination under barbiturate anesthesia) and subject to a full necropsy, including macroscopic observations and key organ weights. Samples of the following tissues were preserved for subsequent histopathological assessment: brain, heart, liver, spleen, kidneys, adrenals, gonads, stomach, small intestine, cecum, salivary gland, trachea, aorta, thymus, lymph nodes, pituitary, urinary bladder, colon, rectum, pancreas, uterus, muscle and any additional tissue that appeared abnormal.

Statistical methods

Applied, but details not given.

Remarks: Because this study was conducted before any standardized guidelines were established, the question of guideline methodology and GLP conduct is not applicable.

Current guidelines indicate that the concentration of test substance should not exceed 5% of the diet to avoid any concerns about nutritional imbalances.

Source: Lyondell Chemical Co. Houston, Texas

Test substance: structurally related propylene glycol

Conclusion: Result: There was a high incidence of mammary fibroadenomas and pituitary adenomas, affecting mostly female rats, but this did not differ
statistically between the treated and control animals and no dose-response relationship was present. There was no evidence of any treatment-related increase in tumors.

Conclusion: No carcinogenic potential was detected under the conditions of this study following dietary administration up to 50000 ppm (approx 1700 - 2100 mg/kg bw/day).

Reliability: (1) valid without restriction
Flag: Critical study for SIDS endpoint

Type: other: Continuous Breeding
Method: Species: mouse
Sex: male/female
Strain: CD-1
Route of admin.: drinking water
Exposure period: continuous (see methods)
Frequency of treatment: daily
Premating exposure period
Male: 7 days prior to first mating, then continuous exposure
Female: 7 days prior to first mating, then continuous exposure
Doses: 1%, 2.5%, 5% in drinking water
Control group: yes, concurrent vehicle
NOAEL Parental: = 5%
NOAEL F1 Offspr.: = 5%
NOAEL F2 Offspr.: = 5%
Method: NTP Reproductive Assessment by Continuous Breeding
Year: 1989
GLP: no data
Method: Animals
There were 40 controls per sex, along with 20 males and 20 females per treatment group in the F0 generation. The F1 mating groups comprised 20 animals per sex from the control and high dose groups only. Animals were housed in single sex groups during a one week pre-mating period, then in breeding pairs or individually. Deionised filtered water and ground rodent chow were available ad libitum.

Treatment
The mice were exposed during a 7-day premating period, after which they were randomly assigned to mating pairs and cohabited and treated continuously for 98 days. At the end of the cohabitation period, the pairs were separated but treatment continued. Any litters born during this time (F1) were delivered, and kept until weaning on PND21: treatment of the mothers continued throughout this period. Treatment of high dose animals from the F1 generation continued until mating at around 74 days of age.

Parental observations
Body weight and water consumption data were collected at unspecified times during the study.

Pup observations
Data (body weight, proportion of males, number of litters per pair, number of live and dead pups) were collected on all new born animals within 12 hr of birth. Litters were then discarded, with the exception of the final F1 generation which was used for breeding purposes. The F2 litters were examined for litter size, sex and pup weight.

Necropsy observations
No necropsy data were collected on the F0 generation, however the F1 adults were subjected to a detailed examination after delivery of the F2
pups.

Statistical analysis
Methods used included Cochran-Armitage test, Fisher's exact test, Kruskall-Wallis test, Wilcoxon-Mann-Whitney test and two-sided t-test

Remark: The treatment levels used in this study were extremely high: dams from the high dose group received the equivalent of 10000 mg/kg bw/day.

Source: Lyondell Chemical Co. Houston, Texas
Test substance: related substance
Conclusion: Results are reported briefly, either as a text summary or in tabulated form in the reference, and no quantitative data were available for evaluation.

Based on data collected during a preliminary dose range finding study, treated animals received the equivalent of 1800, 4800 and 10100 mg/kg bw/day.

Water consumption was consistently higher (6 - 15%) for all groups in the F0 generation, but this was not statistically significant. Body weight in the F0 generation was unaffected by treatment.

There was no treatment-related effect on pup weight adjusted for litter size in either the F1 or F2 generations. The viability and growth of the F1 litter was unaffected by propylene glycol treatment. There were no treatment-related effects on mating, fertility or on the number, weight or viability of the F2 pups.

Necropsy of the F1 adults revealed no effect on body weight or organ weight in males and females, no change in sperm endpoints and no alteration in estrous cycle parameters.

Conclusion: Under the conditions of the study, propylene glycol had no effect on fertility or reproduction in F0 or F1 mice, up to a maximum dose of 10000 mg/kg bw/day. Based on metabolism of dipropylene glycol to propylene glycol, these data on propylene glycol are relevant to dipropylene glycol toxicity characterization. (Reviewer)

Reliability: (1) valid without restriction
Flag: Critical study for SIDS endpoint

Type: other: OECD 422 Combined repeat dose and reproductive/developmental toxicity screening test
Method: Species: rat
Sex: male/female
Strain: Crl: CD(SD)
Route of admin.: gavage
Exposure period: Premating through day 3 lactation: see methods
Frequency of treatment: daily
Premating exposure period
Male: 28 day
Female: 14 days
Duration of test: Through day 3 of lactation
Doses: 8, 40, 200, and 1000 mg/kg/day
Control group: yes, concurrent vehicle
Method: OECD 422 Combined repeat dose and reproductive/developmental toxicity screening test
GLP: Yes
Method: Animals
There were 12 males and 12 females per treatment group.
Treatment
The rats were exposed during a 14-day premating period, after which they were randomly assigned to mating pairs and cohabited until evidence of mating.

Parental observations
Clinical signs, body weight, food consumption, hematology, and clinical chemistry were assessed during the study. At termination, the rats were necropsied, organs were weighed and examined histopathologically.

Pup observations
Litter size, sex and viability on day 0 and 4 of lactation. Visual external examination on day 4.

Statistical analysis
not described

Source: Lyondell Chemical Co. Houston, Texas
Test substance: Tripropylene glycol CAS #24800-44-0
Conclusion: Result: The report is in Japanese; summary results and tables are available in English.

Parental animals
There was no effect on parental body weight, food consumption, hematology, clinical chemistry, necropsy or histopathology. Increased salivation was noted in males treated at 1000 mg/kg/day. Males and females at 1000 mg/kg/day had increased liver weights and males also had increased kidney weights. There was no effect on estrus cycle, mating, pregnancy index, length of gestation, or gestation index.

Pups
There was no difference in mean litter size, mean live pups born/litter, survival to day 4, sex ratio, or frequency of external abnormalities.

NOAEL
Parental: 200 mg/kg/day
Reproductive: 1000 mg/kg/day
Developmental: 1000 mg/kg/day

Conclusion: Under the conditions of the study, tripropylene glycol had no effect on fertility or reproduction. Based on metabolism of tripropylene to dipropylene glycol to propylene glycol, these data on tripropylene glycol are relevant to dipropylene glycol toxicity characterization.

Reliability: (2) valid with restrictions
Not guideline, pre-GLP, methods and results only briefly described but generally acceptable.
Flag: Critical study for SIDS endpoint
29.05.2001

5.11 EXPERIENCE WITH HUMAN EXPOSURE
6. References


6. References


(19) EPIWIN (Estimation Program Interface for Windows).


(33) Spanjers, MT; Til, HP. (1980). Determination of the acute oral toxicity of dipropylene glycol in rats. Report from CIVO -TNO to Dow Chemical Europe, Horgen, Switzerland.

82 UNEP PUBLICATIONS


7. Risk Assessment

7.1 END POINT SUMMARY

7.2.1 HAZARD SUMMARY

7.3 RISK ASSESSMENT